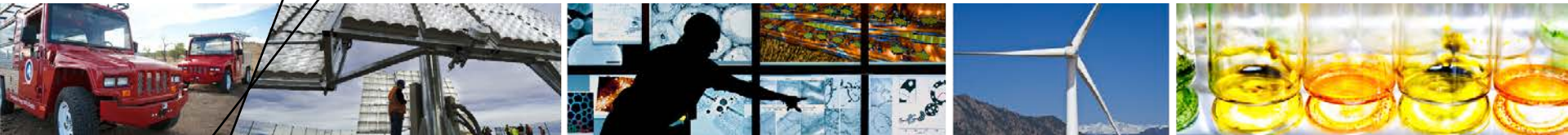


# Indirect Liquefaction of Biomass to Transportation Fuels Via Mixed Oxygenated Intermediates



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*National Renewable Energy Laboratory (NREL) Golden, CO*

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*Session: Biofuels Production: Design, Simulation, and Economics Analysis I*

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# Acknowledgements

## Modeling and Analysis



### Comparative techno-economic analysis and process design for indirect liquefaction pathways to distillate-range fuels via biomass-derived oxygenated intermediates upgrading

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DOE's Bioenergy Technologies Office (BETO)

<http://www.eere.energy.gov/biomass>



Energy Efficiency & Renewable Energy

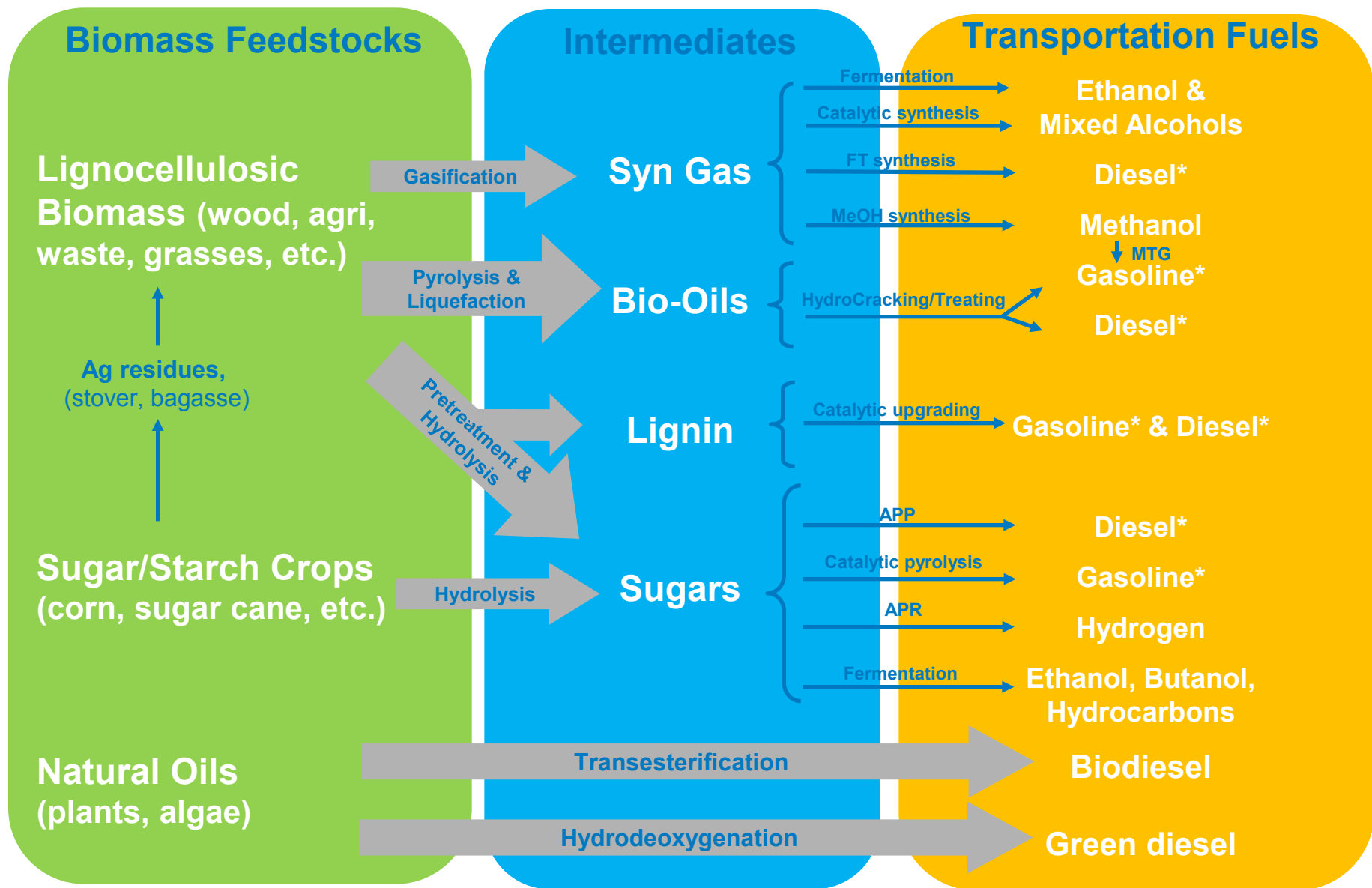
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Speaker information:

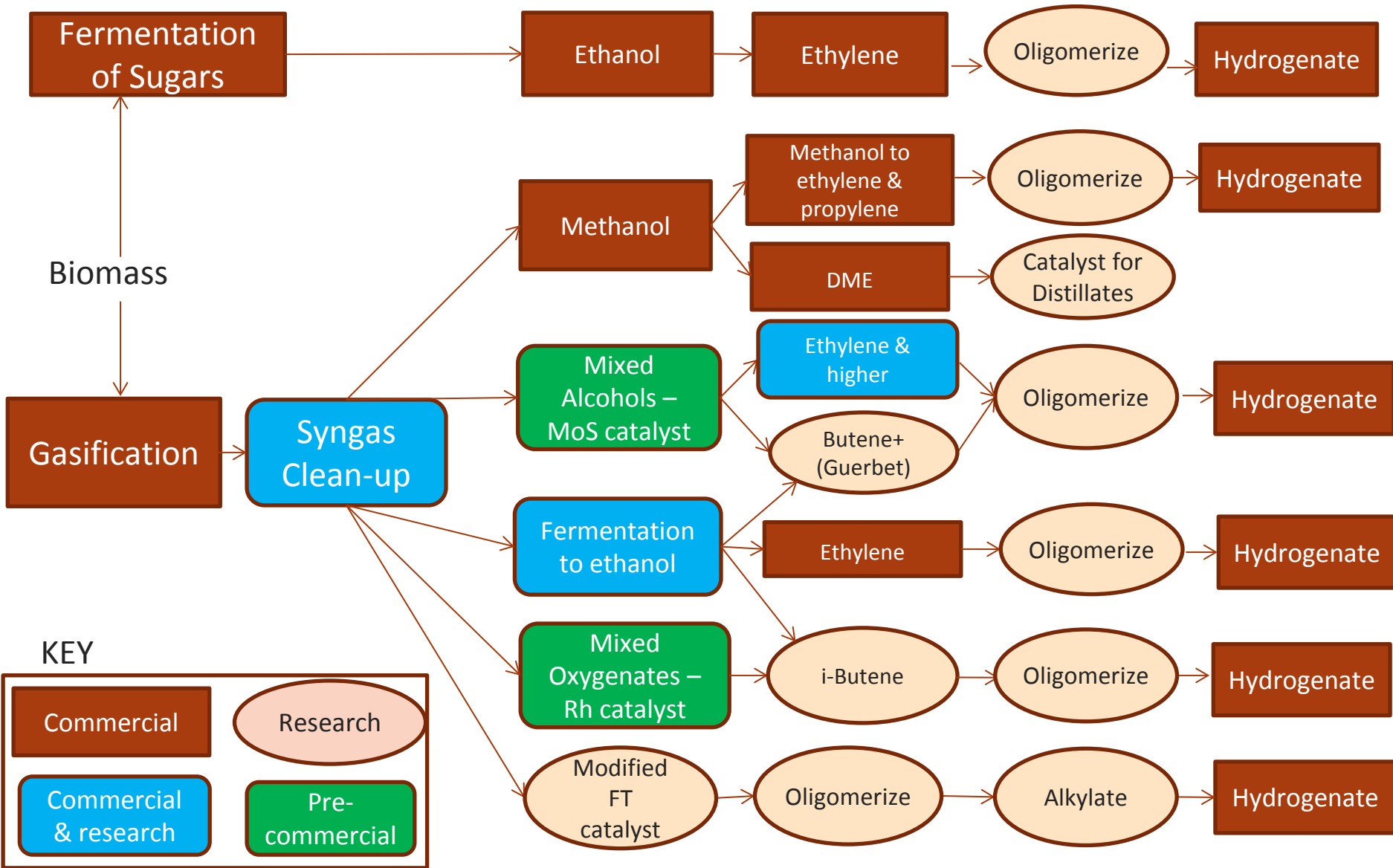
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<http://www.nrel.gov/biomass>

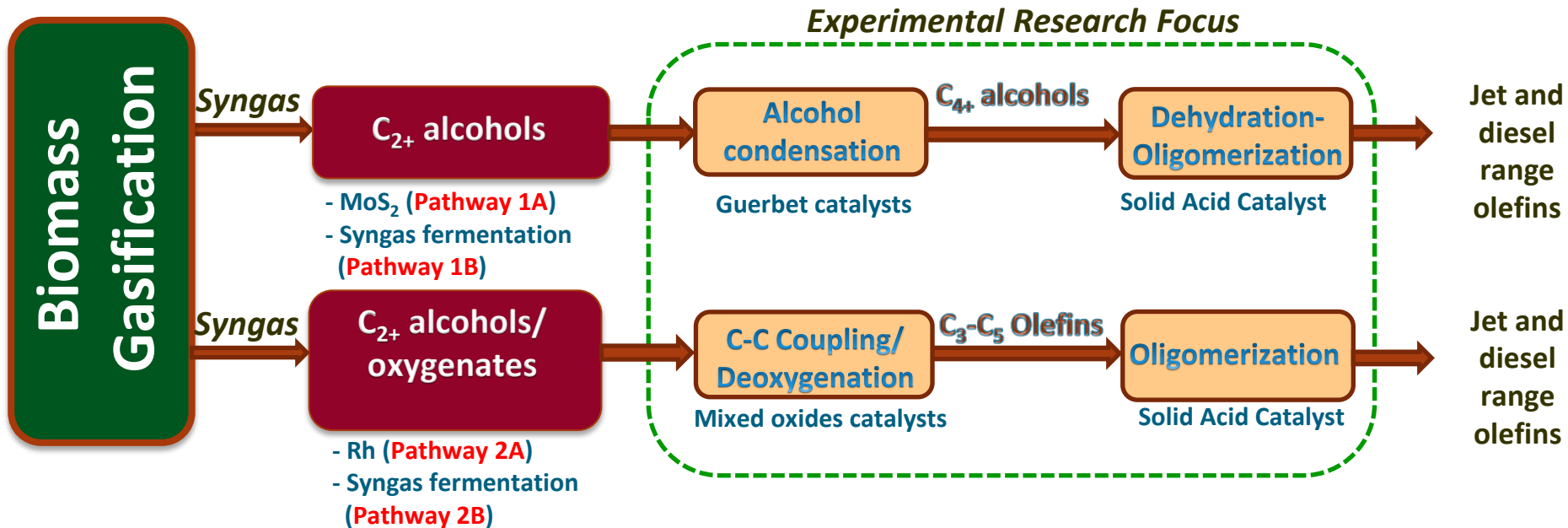
# Background – Biofuels Transportation Options



# IDL Opportunities for Distillates

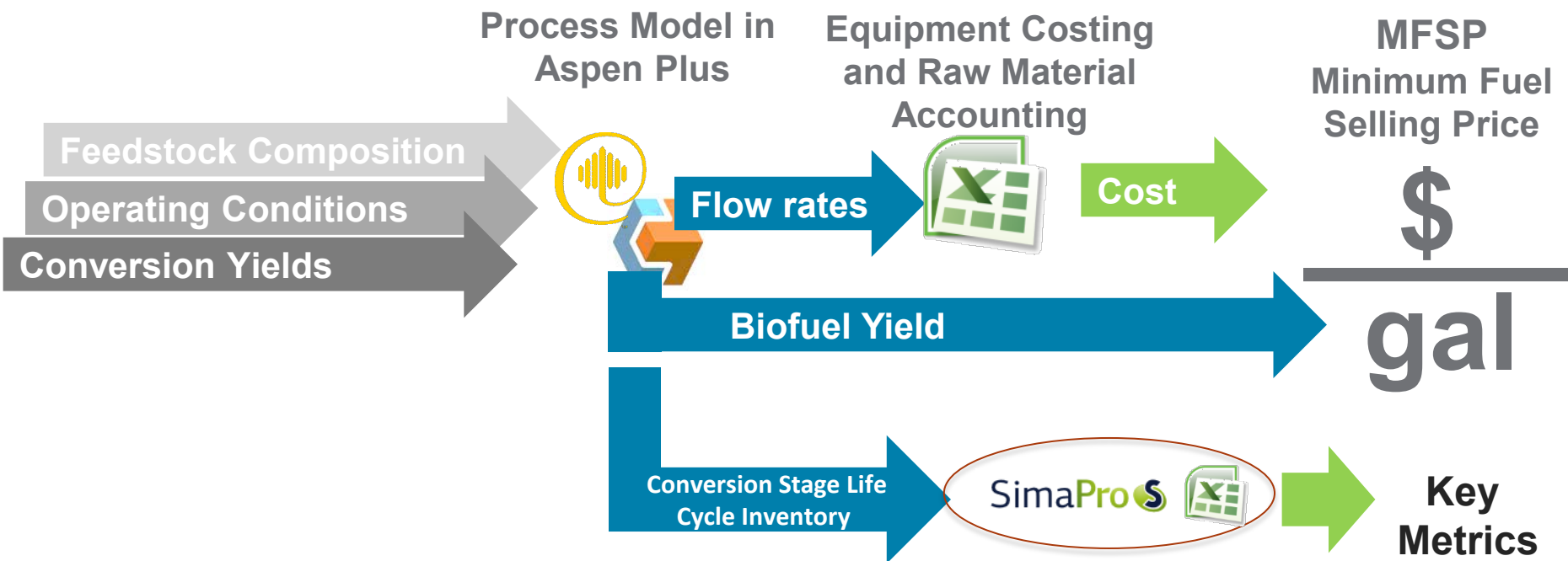


# Syngas to Distillates Routes Evaluated



- This study leverages past research performed for BETO to produce oxygenated intermediates, mixed C<sub>2+</sub> alcohols and mixed C<sub>2+</sub> oxygenates and incorporates the latest developments from current research on the upgrading of oxygenates to hydrocarbon fuels.
- This study also leverages advances in syngas production and clean-up technologies developed for the thermochemical pathway from biomass to mixed alcohols, demonstrated in 2012.

# Approach / Methodology



## Sustainability Metrics Approach:

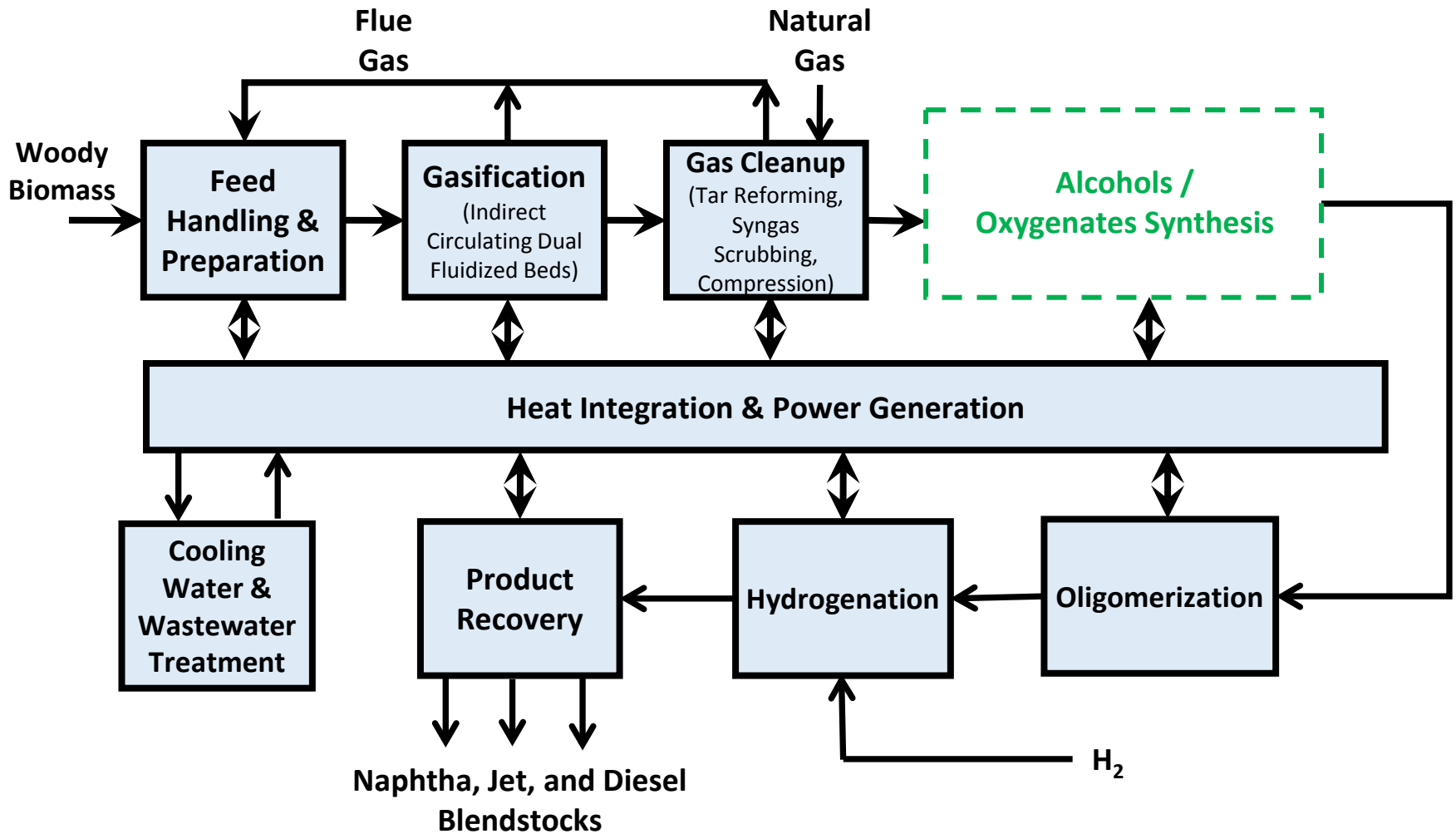
- Partial LCA -- the boundary for the metrics is the biorefinery. The rationale for performing a partial LCA is that the overall focus of this study is the conversion stage. Therefore, to isolate this stage, all others are excluded for quantification of metrics.
- Systematically quantify and assess key sustainability metrics which allow for conversion pathway evaluation and comparison.
- For certain pathways, full LCA is performed for sensitivity cases to understand effects on feedstock stages and limitations around RFS thresholds.

- Fossil GHGs
- Fossil Energy Use
- Fuel Yield
- Carbon-to-Fuel Efficiency
- Water Consumption

# n<sup>th</sup>-Plant TEA Assumptions

Description of Economic Parameter	Analysis Value / Basis
Delivered Feedstock Cost	\$80.00 / US Dry Ton
Internal Rate of Return (IRR)	10.0 %
Plant Financing by Equity / Debt	40% / 60% of Total Capital Investment
Plant Life	30 Years
Income Tax Rate	35.0 %
Interest Rate for Debt Financing	8.0 % Annually
Term for Debt Financing	30 Years
Working Capital Cost	5.0% of Fixed Capital Investment
Depreciation Schedule	Total Plant: 7-Year MACRS
Construction Period (Spending Schedule)	3 Years (8% Y1, 60% Y2, 32% Y3)
Start-Up Time	6 Months
Revenue and Costs During Start-Up	Revenue = 50% of Normal Operation Variable Costs = 75% of Normal Operation Fixed Costs = 100% of Normal Operation
On-Stream Percentage After Start-Up	90% (7,884 Operating Hours per Year)

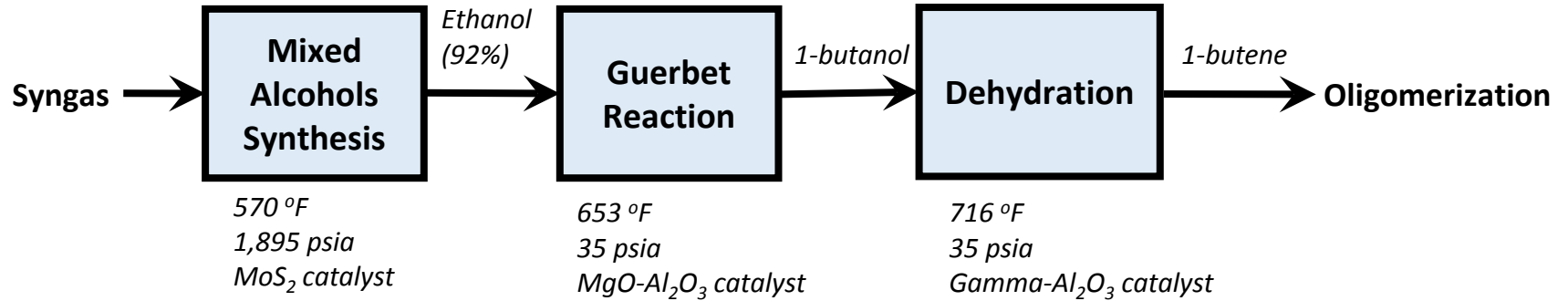
# Main Process Steps



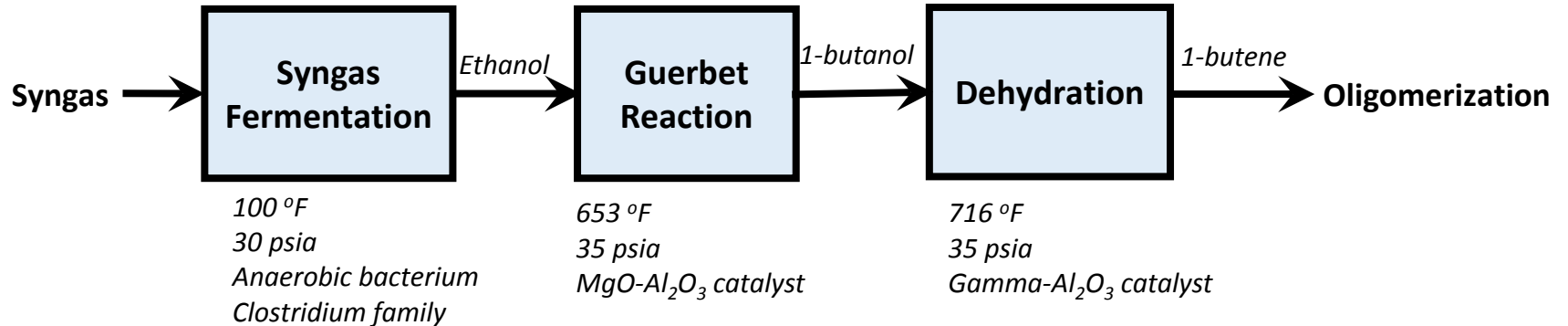


# Alcohols / Oxygenates Synthesis

## Pathway 1A



## Pathway 1B



## Pathway 1 – Design Parameters

### Guerbet reactor

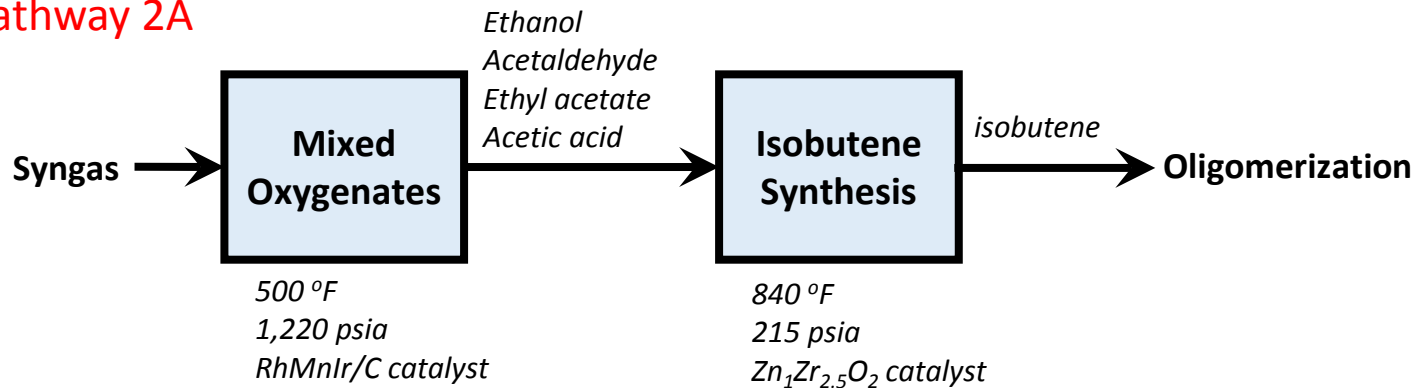
Design Parameters	
Reactor Temperature, °F (°C)	653 (345)
Reactor Pressure, psia (bar)	35 (2.41)
Catalyst	MgO-Al <sub>2</sub> O <sub>3</sub>
WHSV (hr <sup>-1</sup> )	1.0
Single-Pass Conversion	60% (ethanol)
Catalyst Life, yr	4
Oxygenates Selectivities (C), wt%	
Butanol	69.20
Pentanol	8.00
Hexanol	9.30
Heptanol	2.80
2-methyl-1-Butanol	4.20
2-ethyl-1-Butanol	4.20
Octanol	1.00
Others	1.30

### Oligomerization reactor

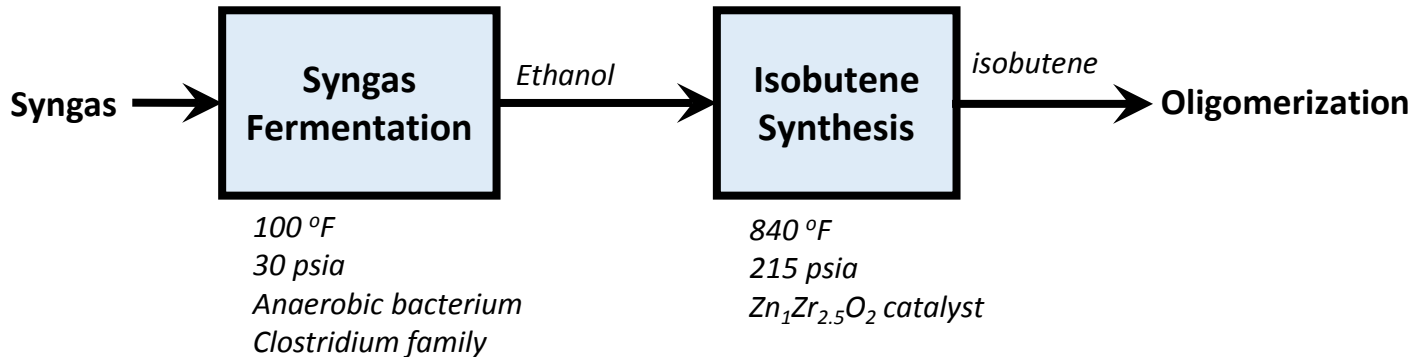
Design Parameters			
Reactor Temperature, °F (°C)	482 (250)		
Reactor Pressure, psia (bar)	435 (30)		
Catalyst	HZSM-23		
WHSV (hr <sup>-1</sup> )	0.21		
Single-Pass Conversion	95% (n-butene)		
Catalyst Life, yr	3		
Product Distribution	C Selectivity (%)	Isomer Selectivity (%)	
C8	26.20	2,3-Dimethyl-1-hexene	20.10
		2-Methyl-1-heptene	72.30
		1-Octene	7.60
C12	43.00	2,4-Dimethyl-1-decene	23.20
		2-Methyl-1-undecene	71.10
		1-Dodecene	5.70
C16	21.90	2,4-Dimethyl tetradecene	26.20
		2-Methyl-1-pentadecene	68.70
		1-Hexadecene	5.10
C20+	8.90	1-Eicosene	100.00

# Alcohols / Oxygenates Synthesis

## Pathway 2A



## Pathway 2B



## Pathway 2 – Design Parameters

### Mixed oxygenates reactor (Rh catalyst)

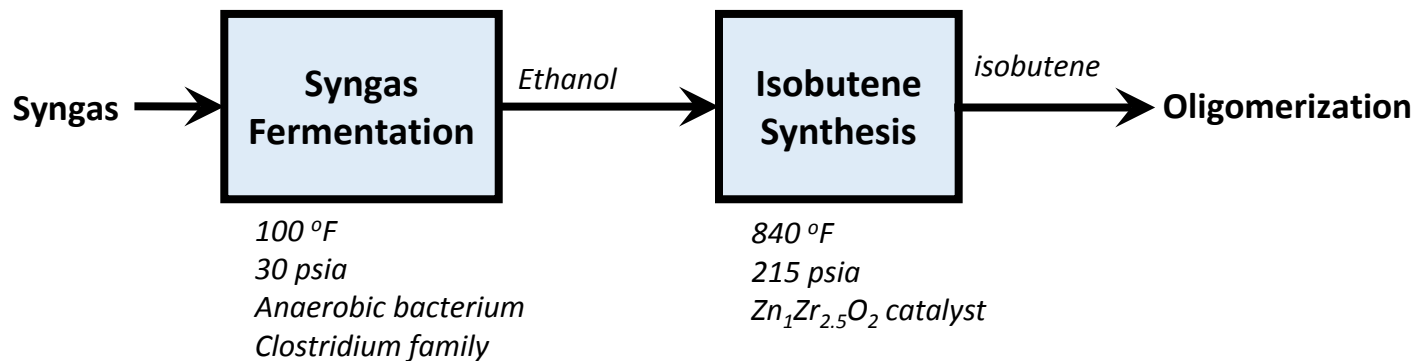
Design Parameters	
Operating Pressure, psia (bar)	1,220 (84.1)
H <sub>2</sub> +CO Partial Pressure, psi (bar)	1,000 (68.9)
Operating Temperature, °F (°C)	500 (260)
Catalyst	RhMnIr/C
GHSV (hr <sup>-1</sup> )	3,247
H <sub>2</sub> :CO Ratio	1.3
Single-Pass Conversion	35%
Catalyst Life, yr	4
Catalyst Rh Loading, wt%	5.6
C Selectivity to C <sub>2+</sub> Oxygenates	85%
Individual Oxygenates Selectivity	
	%
Methanol	0.90
Ethanol	32.80
N-Propanol	1.50
Isobutanol	0.10
1-Butanol	1.40
1-Pentanol	0.20
Acetaldehyde	18.40
Ethyl Acetate	15.70
Acetic Acid	12.10
N-Butyraldehyde	1.20
Propionaldehyde	0.60
Methyl Acetate	1.00
Methane	9.50
Ethane	1.00
Propane	0.30
I-Butane	0.10
N-Pentane	0.00
Ethylene	0.50
Propylene	1.50
1-Butene	0.80
Trans-2-Butene	0.10
Cis-2-Butene	0.10

### Isobutene reactor

Design Parameter	
Operating pressure, psia (bar)	215 (14.8)
Operating temperature, °F (°C)	840 (449)
Catalyst	Zn <sub>1</sub> Zr <sub>2.5</sub> O <sub>2</sub>
GHSV (hr <sup>-1</sup> )	2,000
Single-Pass Conversion	100%
Catalyst Life, yr	4
Carbon Selectivity	
	%
C <sub>3</sub> =	5.76
i-C <sub>4</sub> =	47.4
1-C <sub>4</sub> =	1.70
C <sub>5</sub> =	7.38
CH <sub>4</sub>	2.75
CO <sub>2</sub>	31.6
CO	1.51
C <sub>2</sub> =	0.95
C <sub>2</sub> -C <sub>5</sub> alkanes	0.24
Acetone	0.27
Other oxygenates	0.48

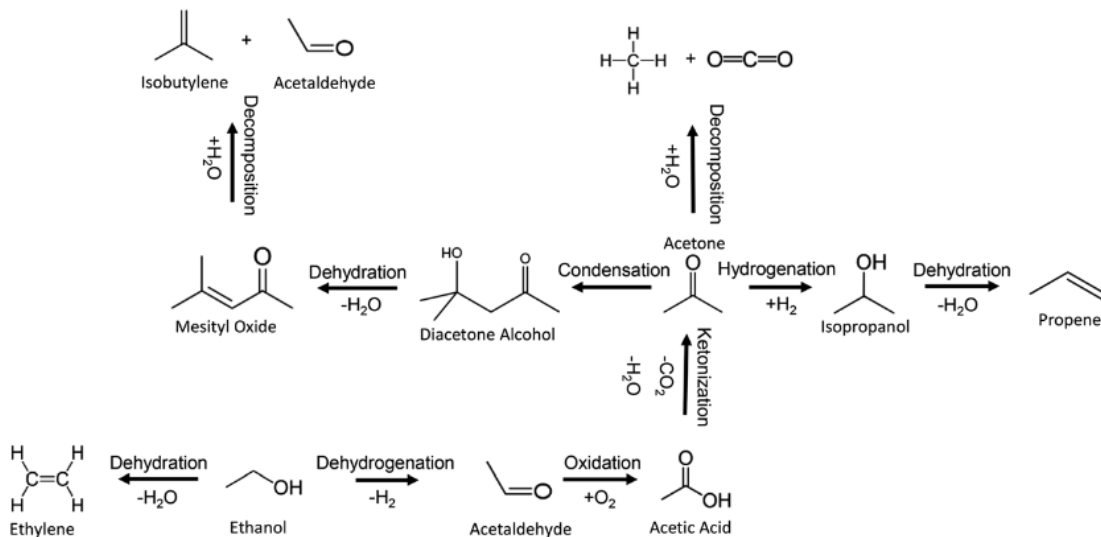
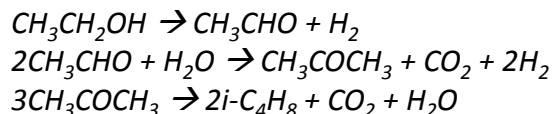
# Alcohols / Oxygenates Synthesis

## Pathway 2B

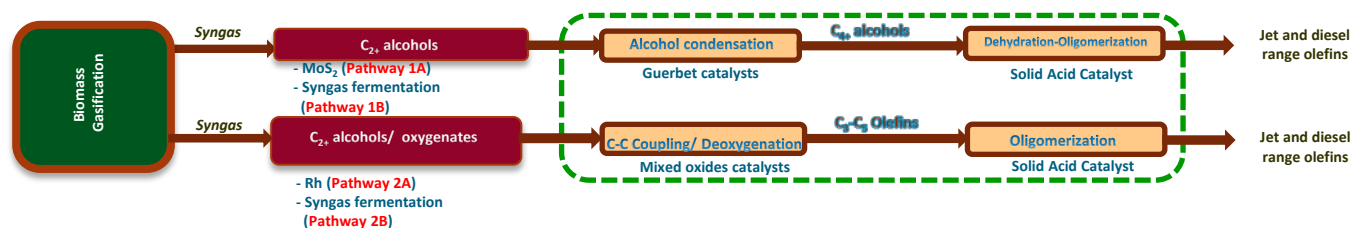


## Ethanol to isobutene reaction network

Dagle et al, Green Chem., 2016,18, 1880-1891



# Process Performance Summary

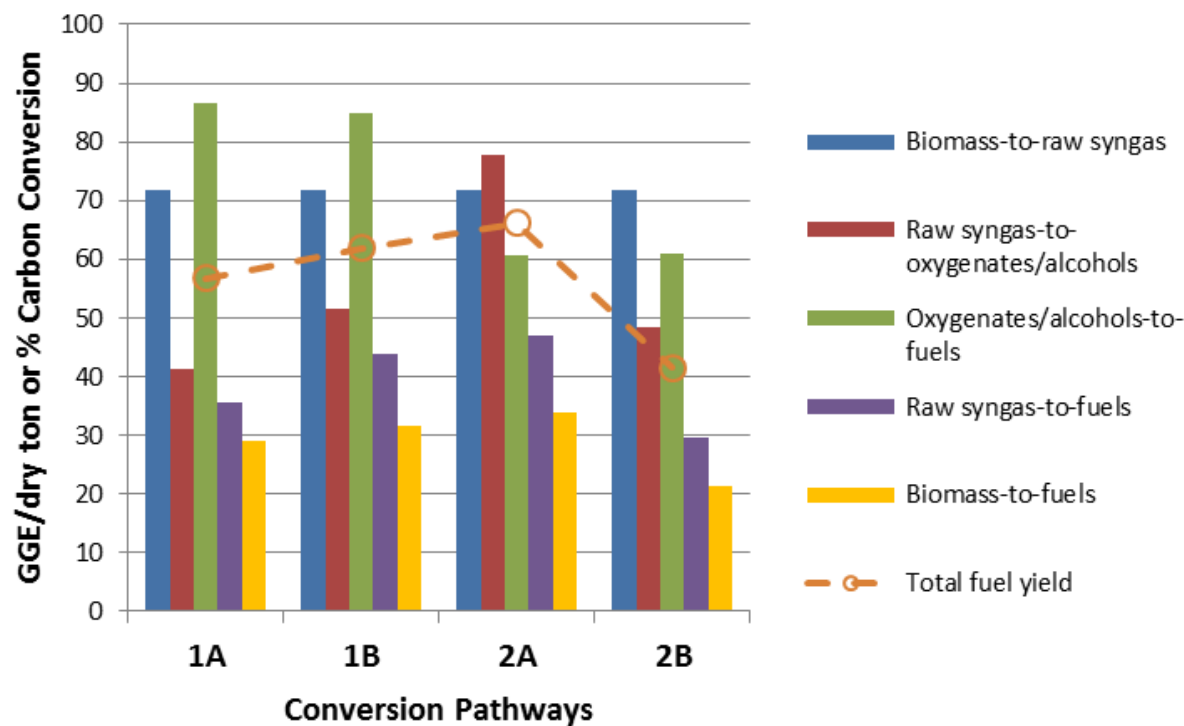
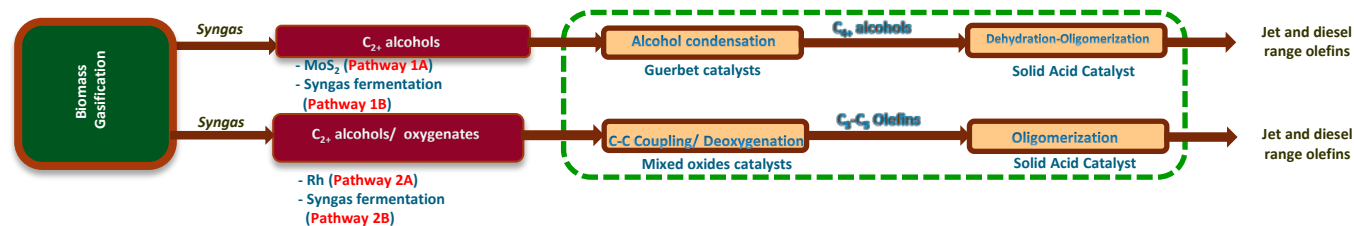


Key Process Targets	Conversion Pathways			
	1A	1B	2A	2B
Oxygenate Intermediate Product from Syngas (lb/hr)	Mixed alcohols <sup>1</sup>	Ethanol	Mixed oxygenates <sup>2</sup>	Ethanol
	59,833	66,283	101,250	65,121
<b>Fuel Production (MMGGE/yr):</b>				
Naphtha range	-	-	11.5	7.23
Jet range	19.5	21.7	31.6	19.84
Diesel range	21.6	23.1	4.80	3.03
<b>Total Fuel Product</b>	<b>41.1</b>	<b>44.8</b>	<b>48.0</b>	<b>30.1</b>
<b>Fuel yield (GGE/dry ton biomass):</b>				
Naphtha range	-	-	15.9	10.0
Jet range	26.9	30.0	43.7	27.4
Diesel range	29.8	31.8	6.70	4.18
<b>Total Fuel Yield</b>	<b>56.7</b>	<b>61.9</b>	<b>66.2</b>	<b>41.6</b>
<b>Carbon Conversion Efficiency:</b>				
Biomass to raw syngas	71.8%	71.8%	71.8%	71.8%
Raw syngas to alcohols/oxygenates	41.2%	51.6%	77.7%	48.5%
Oxygenates to fuels	86.6%	85.0%	60.6%	61.0%
<b>Biomass to fuels</b>	<b>28.9%</b>	<b>31.5%</b>	<b>33.8%</b>	<b>22.1%</b>

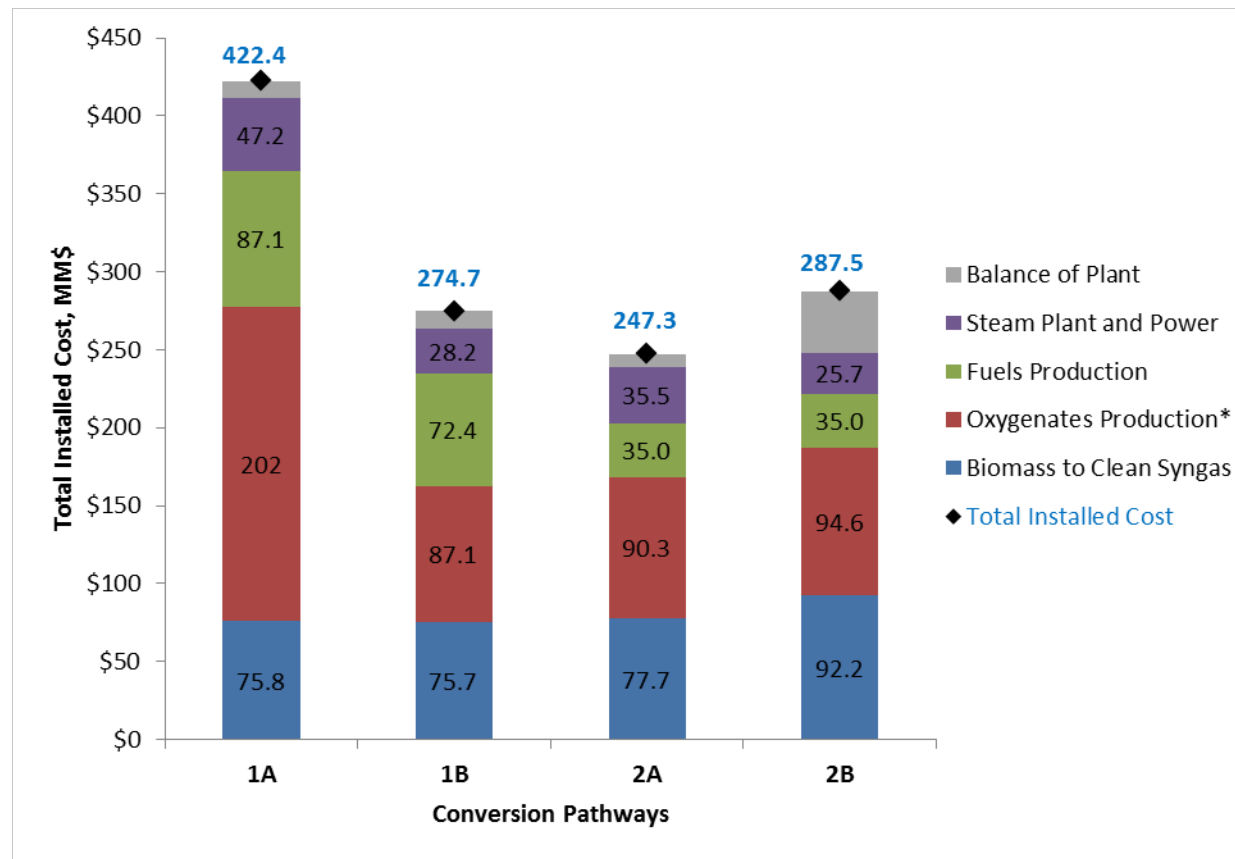
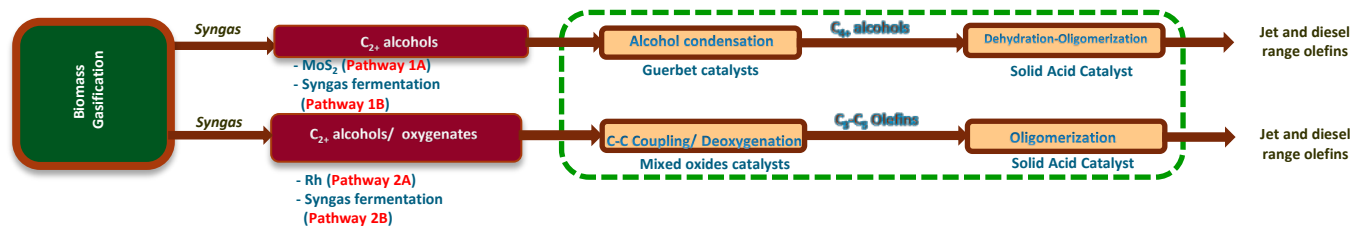
<sup>1</sup> Methanol, ethanol, and propanol

<sup>2</sup> Ethanol, ethyl acetate, acetic acid, and some higher oxygenates, some LPG type material, some methanol

# Correlation Between Carbon Conversion Efficiency and Fuel Yield

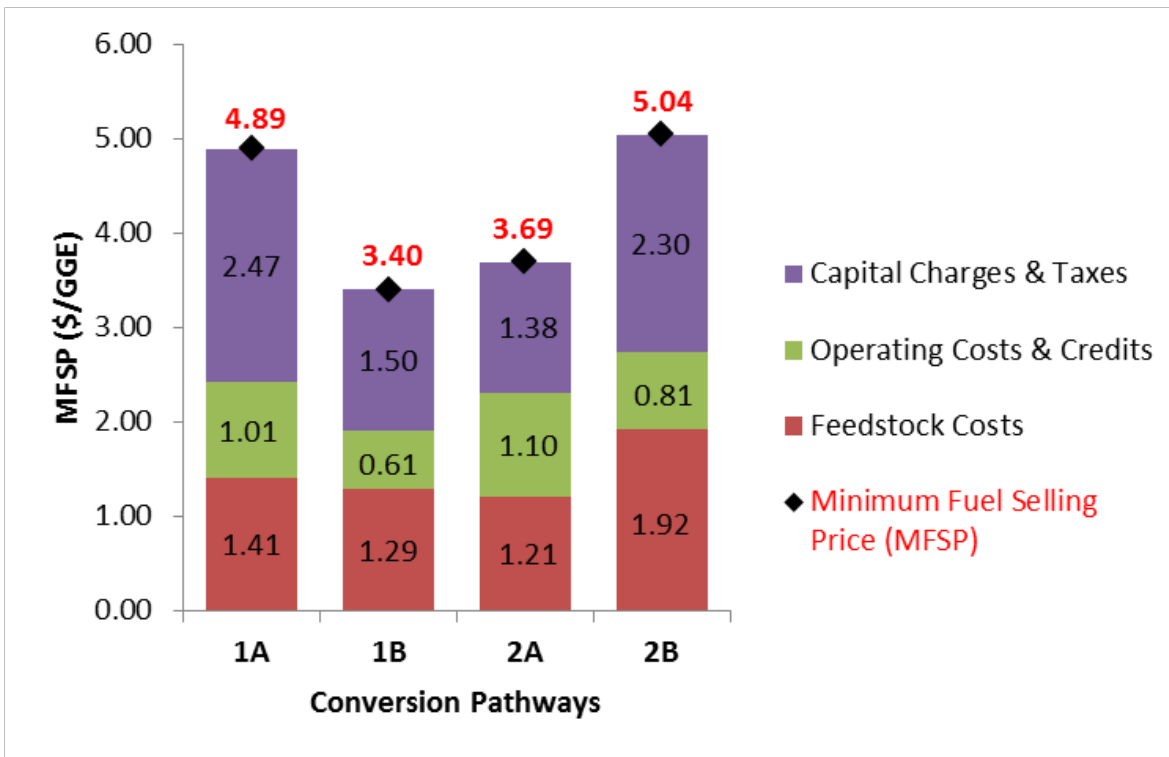
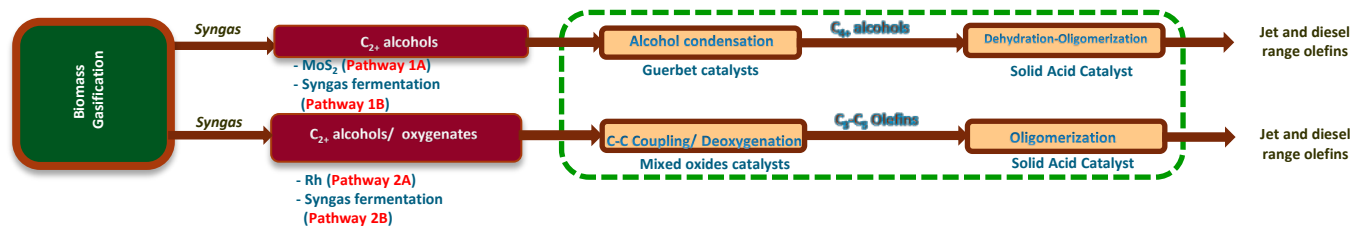


# Total Installed Cost (2011\$)



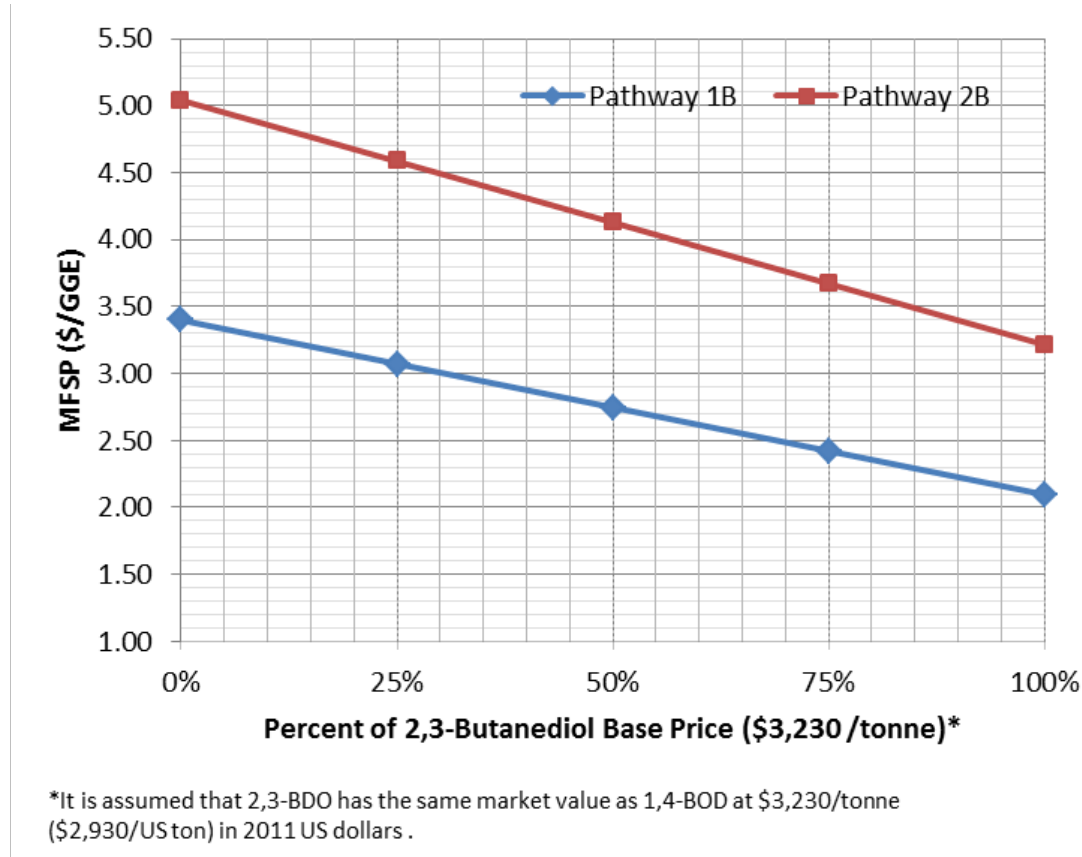


# Minimum Fuel Selling Price (2011\$)



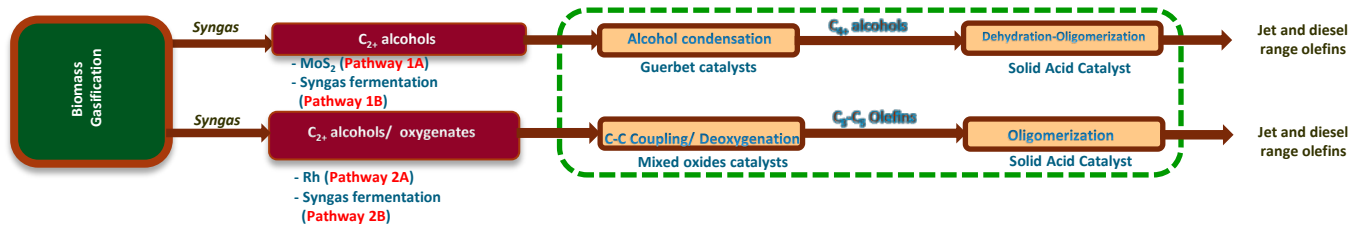
As a reference, \$3.58/GGE for commercial Fischer-Tropsch benchmark.

# Minimum Fuel Selling Price As A Function of Co-product Credits



MFSPs at 0% represents the base cases without the co-production of 2,3-BDO during the syngas fermentation.

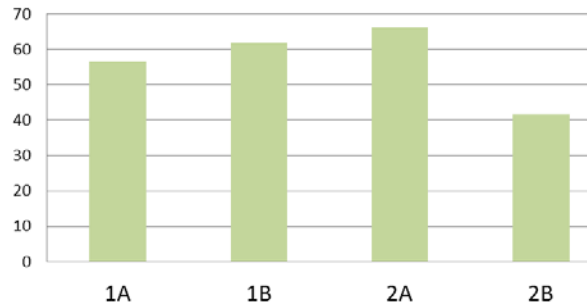
# Environmental Sustainability Metrics



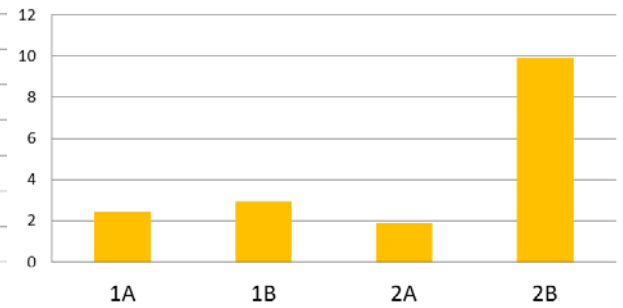
GHG (kg CO<sub>2e</sub>/GGE)



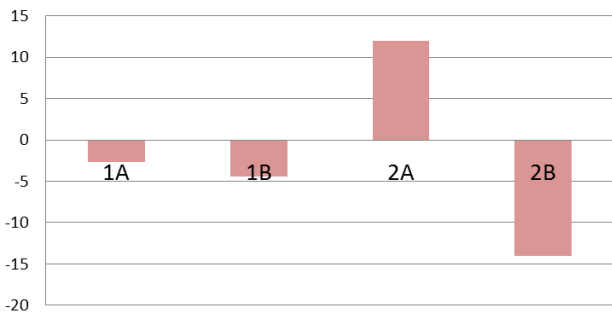
Yield (GGE/dry US ton)



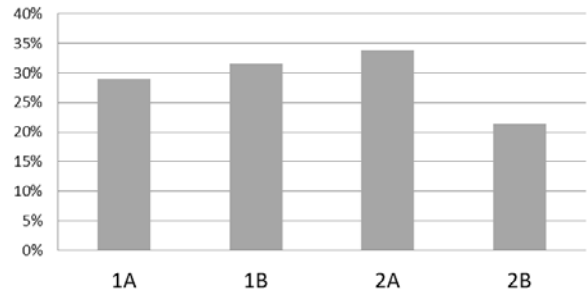
Water Consumption (gal/GGE)



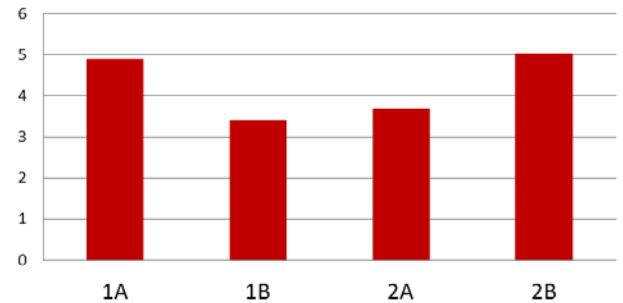
FEC (MJ/GGE)



C Conv Eff (%)



MFSP (\$/GGE)



# Summary

- This paper presents a comparative techno-economic analysis of four emerging conversion pathways from biomass to gasoline-, jet-, and diesel-range hydrocarbons via indirect liquefaction with specific focus on pathways utilizing oxygenated intermediates.
- The processing steps include: biomass-to-syngas via indirect gasification, gas cleanup, conversion of syngas to alcohols/oxygenates followed by conversion of alcohols/oxygenates to hydrocarbon blendstocks via dehydration, oligomerization, and hydrogenation.
- Conversion of biomass-derived syngas to oxygenated intermediates occurs via three different pathways, producing: 1) mixed alcohols over a  $\text{MoS}_2$  catalyst, 2) mixed oxygenates (a mixture of C2+ oxygenated compounds, predominantly ethanol, acetic acid, acetaldehyde, ethyl acetate) using an Rh-based catalyst, and 3) ethanol from syngas fermentation.
- This is followed by the conversion of oxygenates/alcohols to fuel-range olefins in two approaches: 1) mixed alcohols/ethanol to 1-butanol rich mixture via Guerbet reaction, followed by alcohol dehydration, oligomerization, and hydrogenation, and 2) mixed oxygenates/ethanol to isobutene rich mixture and followed by oligomerization and hydrogenation.
- MFSPs for the four developing pathways range from \$3.40/GGE to \$5.04/GGE, in 2011 US dollars.
- Sensitivity studies show that MFSPs can be improved with co-product credits and are comparable to the commercial Fischer-Tropsch benchmark (\$3.58/GGE).
- Overall, this comparative TEA study documents potential economics for the developmental biofuel pathways via mixed oxygenates.

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# Backup Slides

## Breakdown of operating cost contribution (2011 \$)

	Pathway 1A		Pathway 1B		Pathway 2A		Pathway 2B		Benchmark FT	
	MM\$/year	\$/GGE	MM\$/year	\$/GGE	MM\$/year	\$/GGE	MM\$/year	\$/GGE	MM\$/year	\$/GGE
Feedstock	58.0	1.41	58.0	1.29	58.0	1.21	58.0	1.92	58.0	1.22
Natural gas	--	--	--	--	1.44	0.03	--	--	--	--
Catalysts	8.63	0.21	5.11	0.11	24.9	0.52	3.61	0.12	4.63	0.10
Olivine and magnesium oxide	0.41	0.01	0.54	0.01	0.58	0.01	0.54	0.02	0.54	0.01
Hydrogen	1.64	0.04	1.94	0.04	0.00	0.00	0.00	0.00	0.00	0.00
Other raw materials	0.41	0.01	0.72	0.02	2.40	0.05	1.29	0.04	1.11	0.02
Waste disposal	0.41	0.01	0.88	0.02	0.48	0.01	0.85	0.03	0.60	0.01
Co-product credits	-1.64	-0.04	-4.32	-0.10	--	--	-4.88	-0.16	-4.57	-0.10
<b>Total variable costs</b>	<b>67.8</b>	<b>1.65</b>	<b>62.8</b>	<b>1.40</b>	<b>87.7</b>	<b>1.83</b>	<b>59.4</b>	<b>1.97</b>	<b>60.3</b>	<b>1.27</b>
Fixed operating costs	31.6	0.77	22.3	0.50	23.0	0.48	23.1	0.77	25.9	0.55
<b>Total operating costs</b>	<b>99.5</b>	<b>2.42</b>	<b>85.2</b>	<b>1.90</b>	<b>110.8</b>	<b>2.31</b>	<b>82.4</b>	<b>2.74</b>	<b>86.2</b>	<b>1.81</b>
Capital depreciation	23.4	0.57	15.2	0.34	13.4	0.28	15.8	0.53	18.3	0.38
Average income tax	11.1	0.27	7.30	0.16	10.1	0.21	16.4	0.54	19.1	0.40
Average return on investment	67.0	1.63	44.63	1.00	42.72	0.89	37.17	1.24	46.5	0.98
<b>Total</b>	<b>201.0</b>	<b>4.89</b>	<b>152.2</b>	<b>3.40</b>	<b>177.0</b>	<b>3.69</b>	<b>151.8</b>	<b>5.04</b>	<b>170.0</b>	<b>3.58</b>

## Variable operating costs (2011 \$)

Variable	Information and operating cost
Feedstock	Blended biomass contains 45% pulpwood, 32% wood residues, 3% switchgrass, and 20% construction and demolition waste. Price: \$80.00/dry U.S. ton
Gasifier bed material	Synthetic olivine and MgO. Initial fill then a replacement rate of 0.01 wt% of circulation or 7.2 wt% per day of total inventory. Delivered to site by truck with self-contained pneumatic unloading equipment. Disposal by landfill. Olivine price: \$275/tonne MgO price: \$580/tonne
Tar reformer catalyst (Ni-Mg-K/Al <sub>2</sub> O <sub>3</sub> )	To determine the amount of catalyst inventory, the tar reformer was sized for a gas hourly space velocity (GHSV) of 2,476/h based on the operation of the tar reformer at NREL's pilot plant demonstration unit. GHSV is measured at standard temperature and pressure. Initial fill then a replacement rate of 0.15 wt% of catalyst inventory per day. Price: \$47.70/kg based on NREL calculations using metals pricing and costs for manufacturing processes.
Mixed alcohols synthesis catalyst (MoS <sub>2</sub> )	Initial fill then replaced every 2 years based on expected catalyst lifetime. Catalyst inventory based on GHSV of 5,000/h. Price: \$31.23/lb (initial load); \$27.12/lb (after initial load)
Guerbet catalyst (MgO-Al <sub>2</sub> O <sub>3</sub> )	Initial fill then replaced every 4 years based on expected catalyst lifetime. Catalyst inventory based on WHSV of 1.0/h. Price: \$25.00/lb
Dehydration catalyst (Gamma alumina)	Initial fill then replaced every 3 years based on expected catalyst lifetime. Catalyst inventory based on WHSV of 1.0/h. Price: \$10.30/lb
Oligomerization catalyst (HZSM-23)	Initial fill then replaced every 3 years based on expected catalyst lifetime. Catalyst inventory based on WHSV of 0.21/h. Price: \$30.80/lb

## Variable operating costs (2011 \$)

Variable	Information and operating cost
Dimerization catalyst (Nafion)	Initial fill then replaced every 4 years based on expected catalyst lifetime. Catalyst inventory based on WHSV of 1.0/h. Price: \$9.89/lb (Ion Power Inc., New Castle, DE)
Hydrogenation catalyst (Pd/Al <sub>2</sub> O <sub>3</sub> )	Initial fill then replaced every 3 years based on expected catalyst lifetime. Catalyst inventory based on WHSV of 1.0/h. Price: \$55.20/lb (PEP 2014 Yearbook, 0.4% Pd on Al <sub>2</sub> O <sub>3</sub> )
Rhodium-based catalyst (RhMnIr/C)	Initial fill then replaced every 4 years based on expected catalyst lifetime. Catalyst inventory based on GHSV of 3,247/h. Price: \$552/lb (PNNL estimate)
Isobutene production catalyst (Zn <sub>1</sub> Zr <sub>2.5</sub> O <sub>2</sub> )	Initial fill then replaced every 4 years based on expected catalyst lifetime. Catalyst inventory based on GHSV of 2,000/h. Price: \$30.00/lb (PNNL estimate)
Oligomerization catalyst (Amberlyst 36)	Initial fill then replaced every 1 year based on expected catalyst lifetime. Catalyst inventory based on WHSV of 0.756/h. Price: \$15.62/lb (Dow Chemicals)
Solids disposal	Price: \$18.20/ton (tar reformer catalyst disposal) Price: \$54.00/ton (sand and ash purge)
Diesel fuel	Usage: 38 L/h plant-wide use. Price: \$22.39/GJ (2012 price projection)
Natural gas	Price: \$5.10 per 1,000 standard cubic feet (EIA, 2011 industrial average)
Purchased hydrogen	Price: \$0.684/lb
Co-products	Price: \$608/tonne acetic acid (2013 IHS Chemical Economics Handbook) Price: \$3,230/tonne 1,4-butanediol (2013 IHS Chemical Economics Handbook)
Electricity	Price: \$6.89/kWh (EIA, 2011 industrial average)
Water makeup	Price: \$0.35/tonne
Chemicals	Boiler feed water chemicals-Price: \$6.13/kg Cooling tower chemicals-Price: \$3.67/kg LO-CAT chemicals-Price: \$498.98/tonne sulfur produced from NREL/Harris Group Inc. estimates based on other projects. DEPG makeup-Price: \$81.59/million lb acid gas removed Selective amine makeup-Price: \$39.81/million kg acid gas removed.
Wastewater	Most wastewater is cleaned using an RO system and recycled. The balance of the wastewater is sent to off-site treatment facility. Price: \$0.83/tonne