

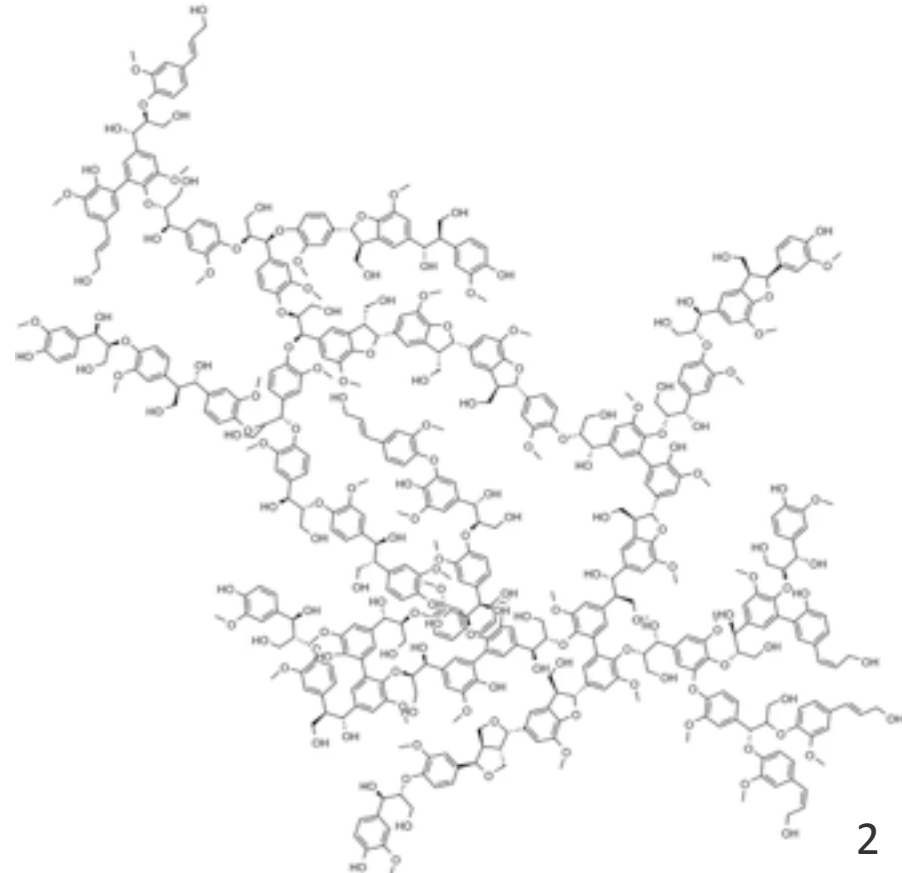
Binding Affinity Dependence of Lignin- Cellulose Complexes on Cellulose Faces and Lignin Composition

Josh Vermaas, Gregg Beckham, and Michael
Crowley

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Boston, Massachusetts

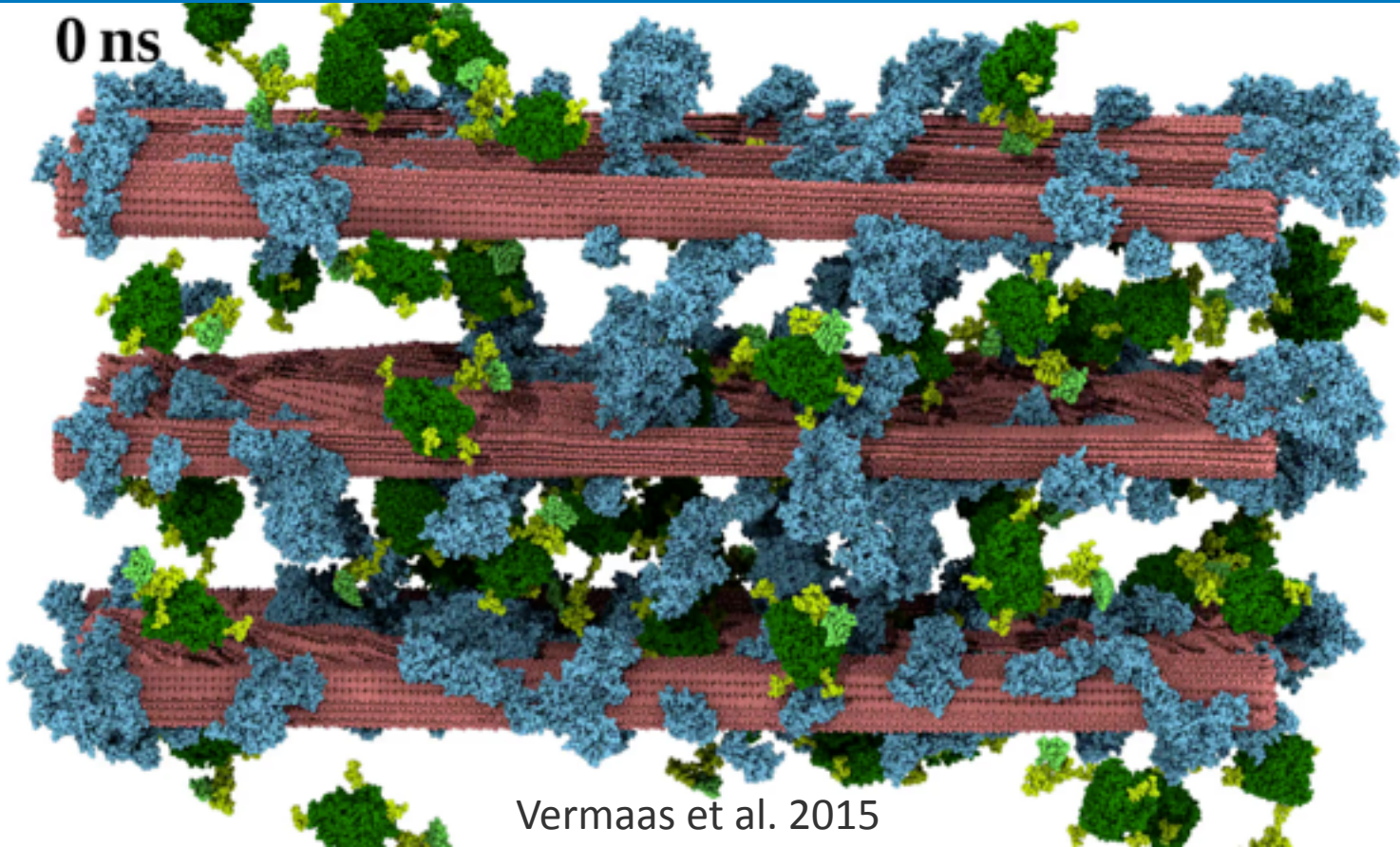
Lignin Introduction

- Lignin is an aromatic heteropolymer that makes up between 15-40% of the dry weight of terrestrial plants
- Largest source of renewable aromatics
- Many potential industrial uses
- Implicated as a significant driver of cell wall recalcitrance



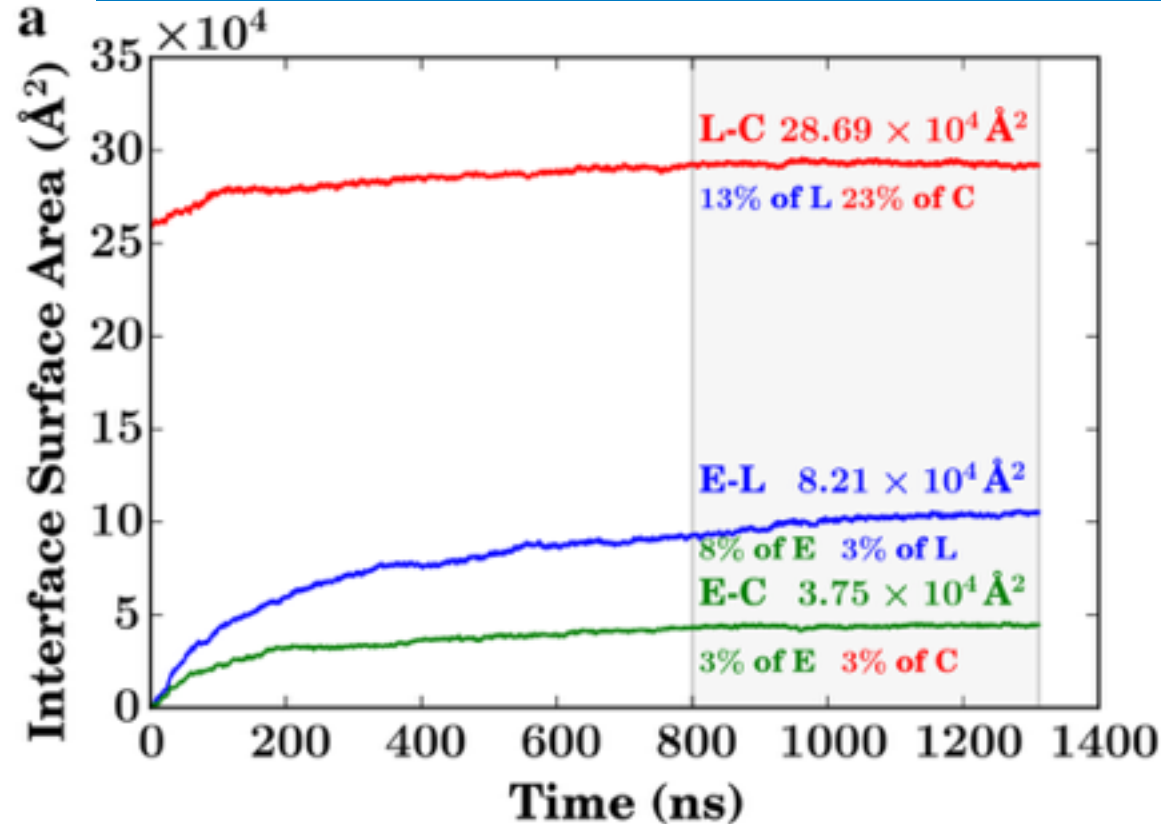
Lignocellulose Simulation

0 ns



Vermaas et al. 2015

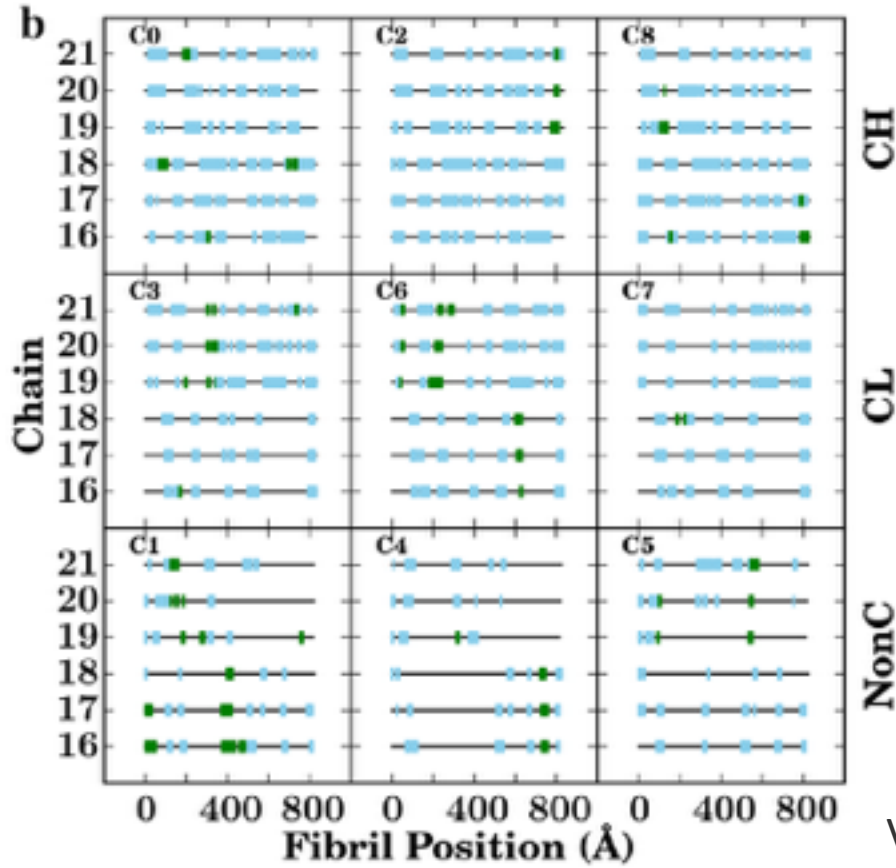
Lignin Binds to Cellulose...



Vermaas et al. 2015

- Lignin covers roughly a quarter of the cellulose surface
- Does it coat the surface evenly?
- Does lignin chemistry matter?

... Particularly Hydrophobic Cellulose Faces

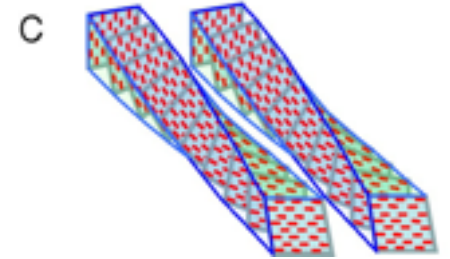
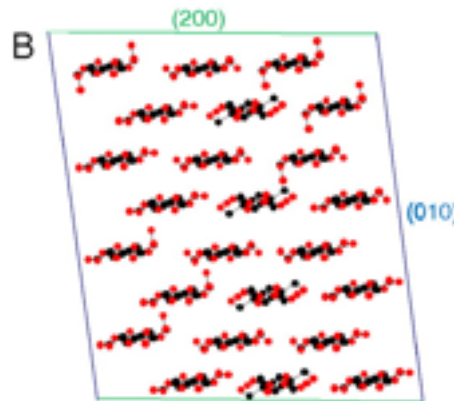
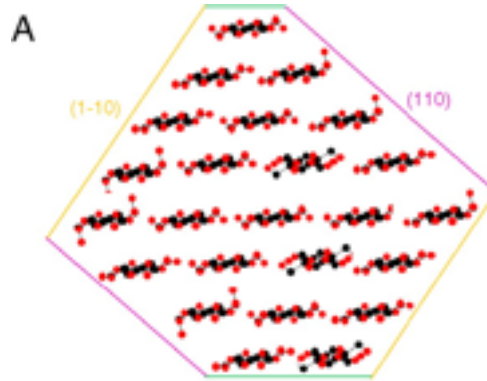


	A_T (10^4 \AA^2)	A_E (10^4 \AA^2)	A_L (10^4 \AA^2)	A_E/A_T	A_L/A_T
CH	6.51 6.74	0.18 0.15	3.02 1.33	0.03 0.02	0.46 0.20
CL	6.51 6.74	0.23 0.24	2.11 0.97	0.04 0.04	0.32 0.14
NonC	6.57 7.76	0.30 0.36	1.16 1.17	0.05 0.05	0.18 0.15

- Prefers hydrophobic faces
- Blocks cellulase binding
- Can we quantify the thermodynamic preference between faces?

The Many Faces of Cellulose

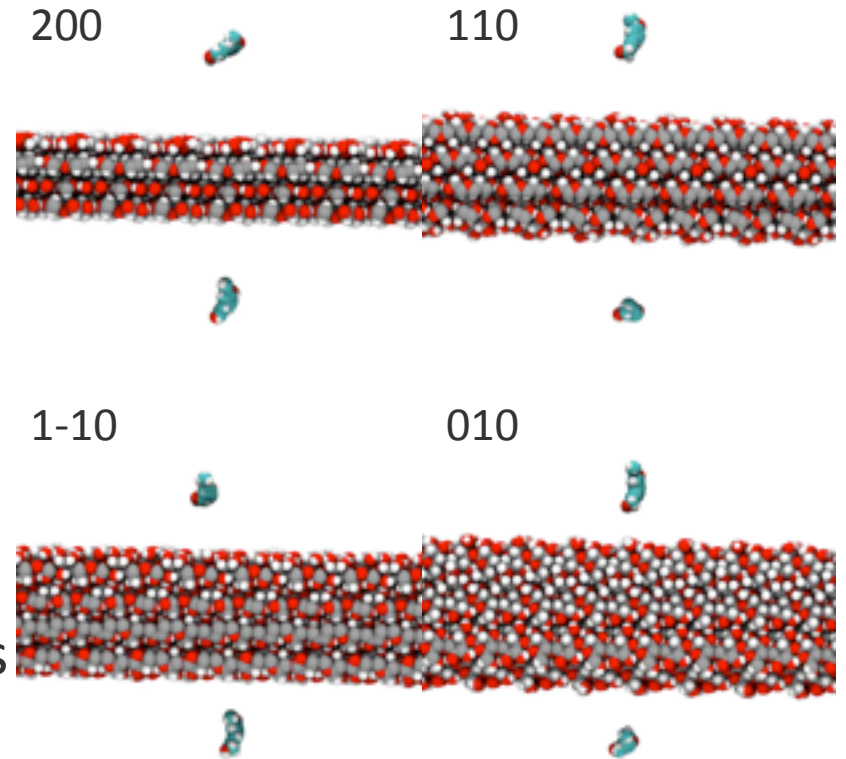
- Only the 200 face is hydrophobic
- Other primary faces are hydrophilic
- Normal cellulose twists
- Infinite cellulose does not



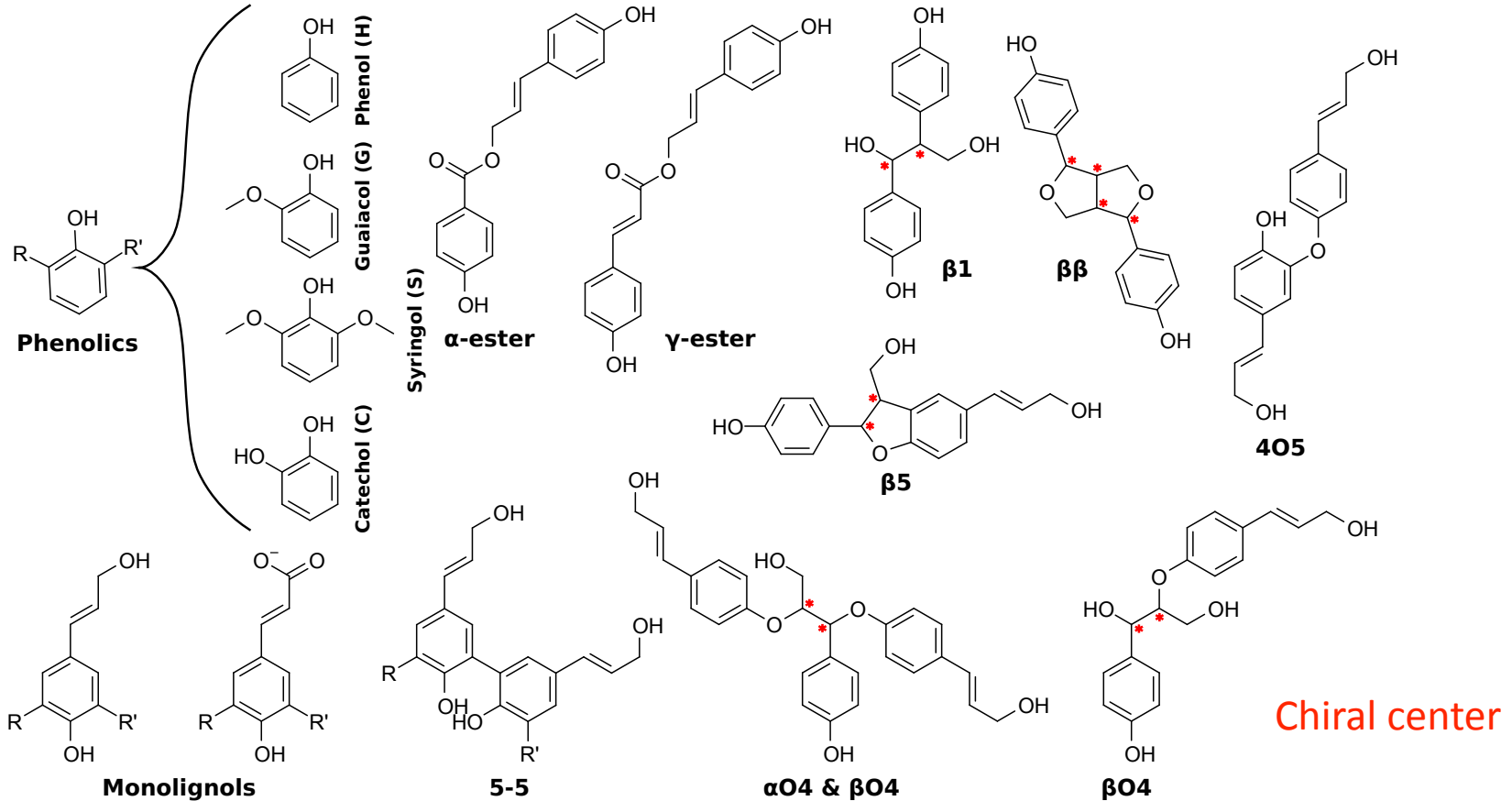
Fernandes et al. 2015

The Next Experiment

- Construct “infinite” sheets of cellulose with different faces exposed to solution
- Place lignin derived compounds in solution
- Simulate for 200 ns
- Determine binding free energies of compounds to individual faces



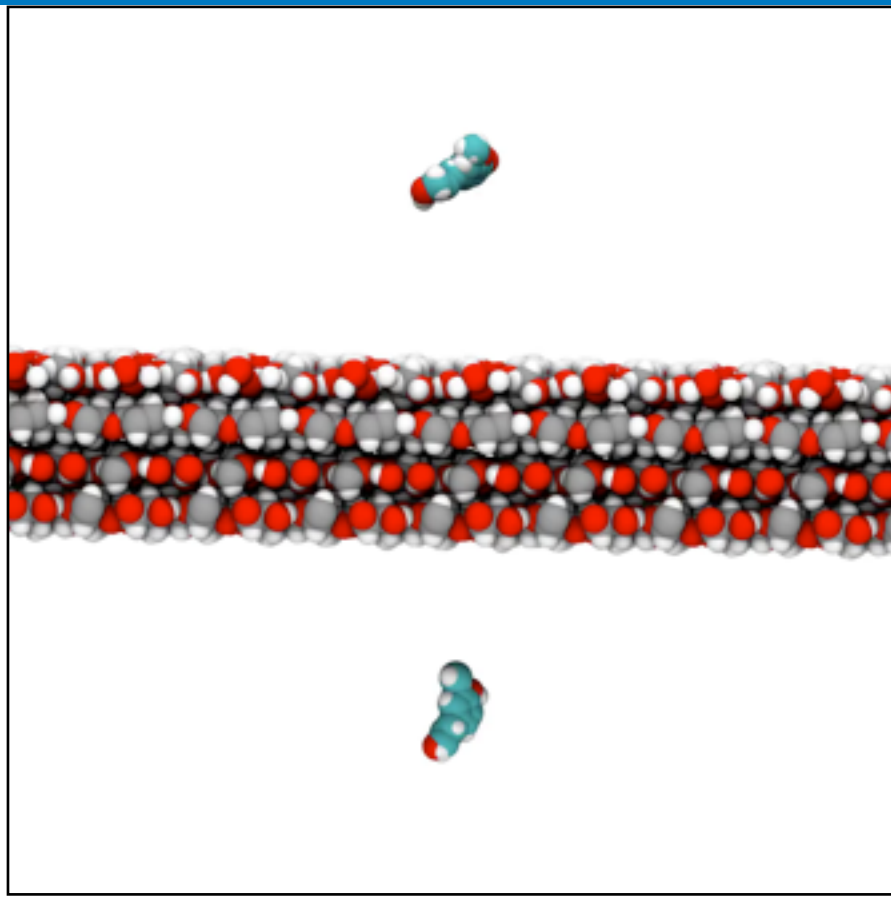
What Compounds?



Potential Binding Metrics for Free Energy Determination

$$\Delta G_{binding} = -RT \ln \frac{P_{bound}}{P_{unbound}}$$

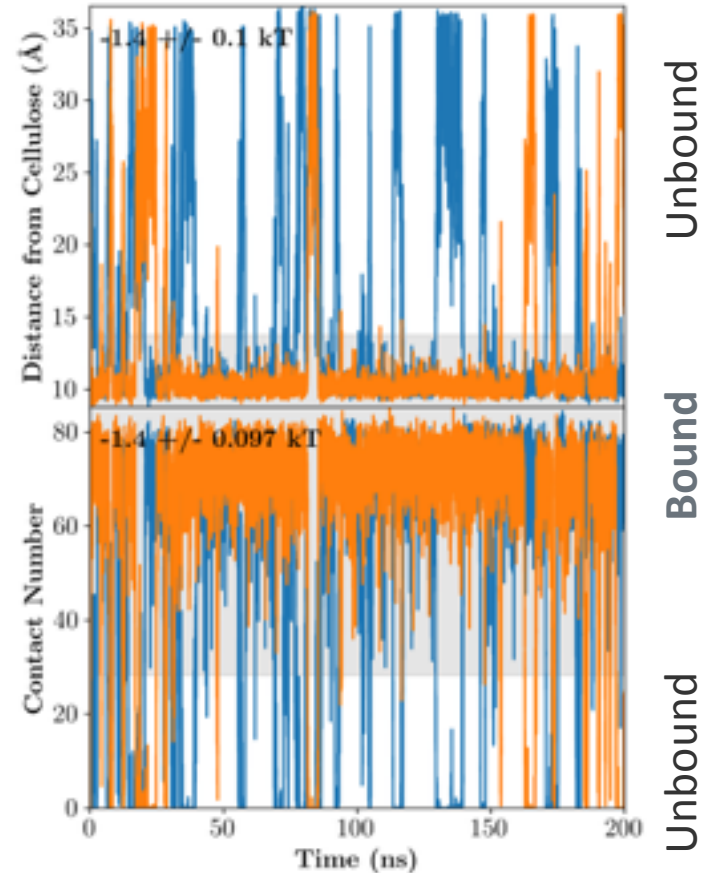
- Free energy is related to probability
- Trajectories can be used to determine relative probabilities
- How do we define a bound vs unbound state?



Distance and Contact Metrics, Compared

- Distance
 - Lignin Center of mass to cellulose center of mass
 - Very easy to explain
- Contact number
 - Pairwise contacts between cellulose and lignin
 - Is not biased by lignin shape

$$C = \sum_{i,j \in \text{pairs}} \frac{1 - (|x_i - x_j| / 12\text{\AA})^6}{1 - (|x_i - x_j| / 12\text{\AA})^{24}}$$



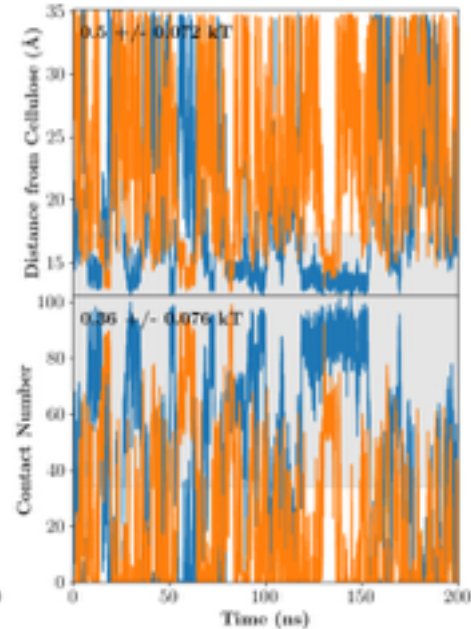
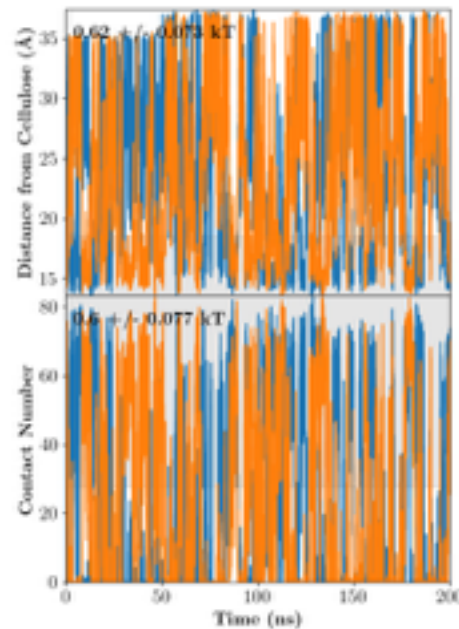
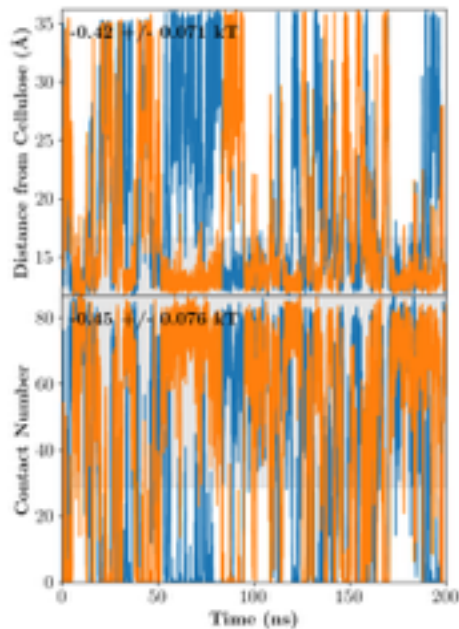
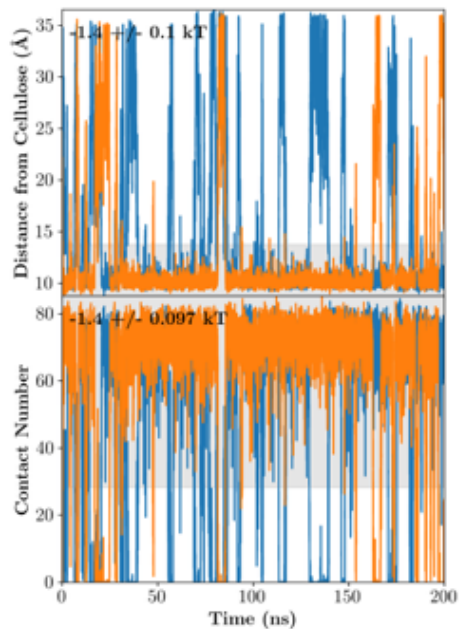
A Closer Look at G-Lignin

200

110

1-10

010



-1.4kT

-.4kT

0.6kT

0.4kT

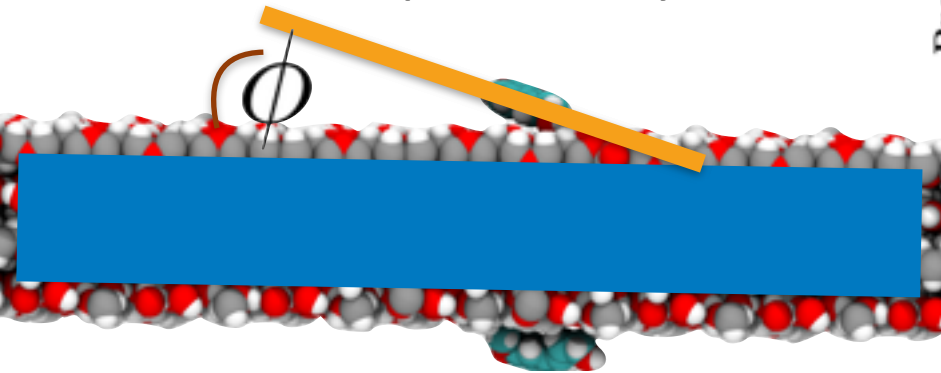
Cellulose Binding for Monomers

- Hydrophobic 200 face predominant for binding
- Ferulate always prefers solution
 - Fits well with its physiological role
- 110 face better for binding than the other hydrophilic faces

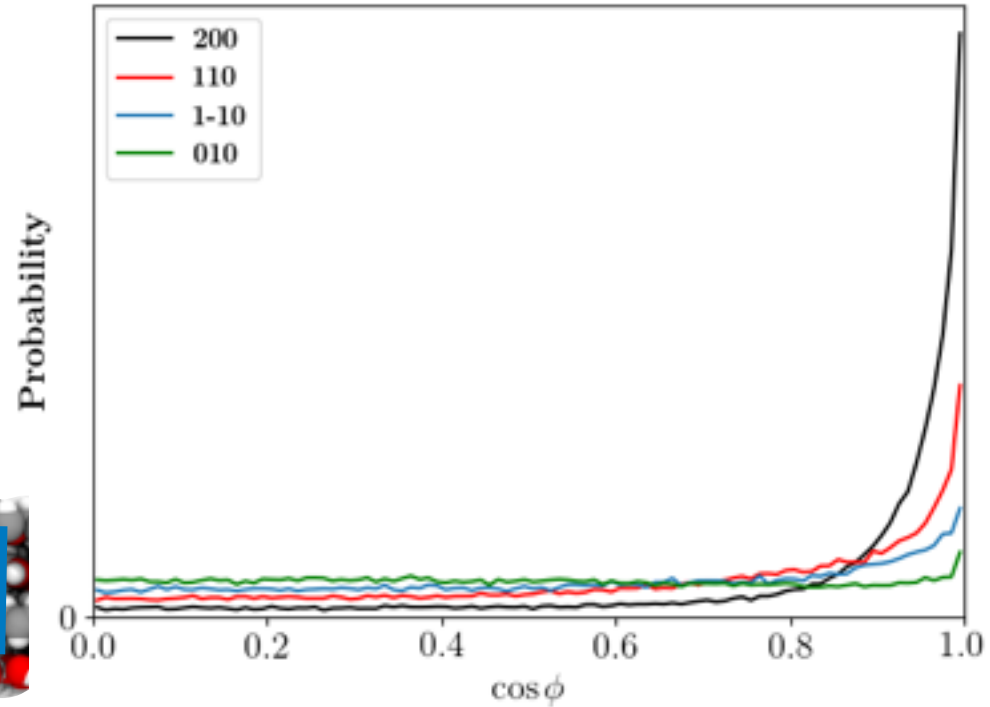
Compound	Cellulose Face Binding Energy (kT)			
	200	110	1-10	010
Phenol	-0.27 ± 0.08	0.15 ± 0.07	0.46 ± 0.08	-0.19 ± 0.06
Guaiacol	-0.29 ± 0.06	0.14 ± 0.07	0.52 ± 0.07	0.85 ± 0.08
Syringol	-0.50 ± 0.08	0.25 ± 0.07	0.35 ± 0.08	0.22 ± 0.07
Catechol	-0.24 ± 0.06	0.02 ± 0.07	0.41 ± 0.08	-0.30 ± 0.08
<i>p</i> -Coumaryl alcohol	-1.02 ± 0.09	-0.84 ± 0.09	0.05 ± 0.07	-0.23 ± 0.08
Coniferyl alcohol	-1.44 ± 0.10	-0.45 ± 0.08	0.60 ± 0.08	0.36 ± 0.08
Sinapyl alcohol	-2.44 ± 0.13	-0.57 ± 0.07	-0.04 ± 0.08	0.79 ± 0.08
Caffeyl alcohol	-1.48 ± 0.09	-0.72 ± 0.09	-0.29 ± 0.07	-0.67 ± 0.07
<i>p</i> -Coumarate	-0.18 ± 0.07	0.08 ± 0.08	0.55 ± 0.09	0.24 ± 0.08
Ferulate	0.06 ± 0.07	0.30 ± 0.08	0.58 ± 0.08	0.93 ± 0.07
Sinapate	-1.12 ± 0.09	-0.04 ± 0.07	0.54 ± 0.07	-0.19 ± 0.07
Caffeate	-0.68 ± 0.08	0.12 ± 0.07	0.59 ± 0.09	-0.21 ± 0.07

Aromatic Angle Distribution Highly Skewed

- Observed distribution favors aromatic rings coplanar with the cellulose surface
- Random distribution yield uniform probability



Coniferyl alcohol distribution

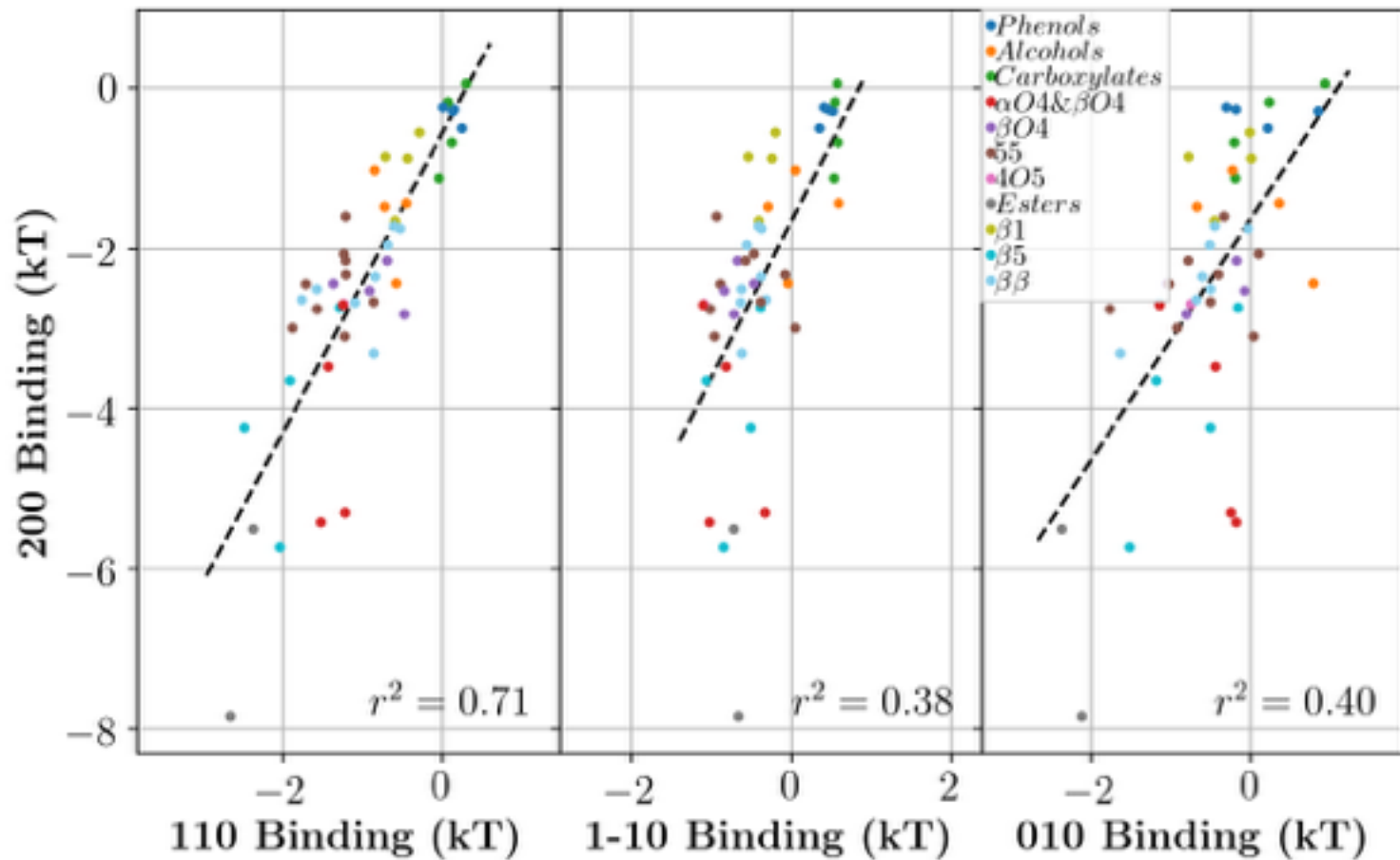


Multimeric Binding

- 200>>110>010>1-10
- Planar lignin molecules like esters demonstrate particular affinity to the 200 face
- Comparisons between stereoisomers suggest longer simulation may be beneficial

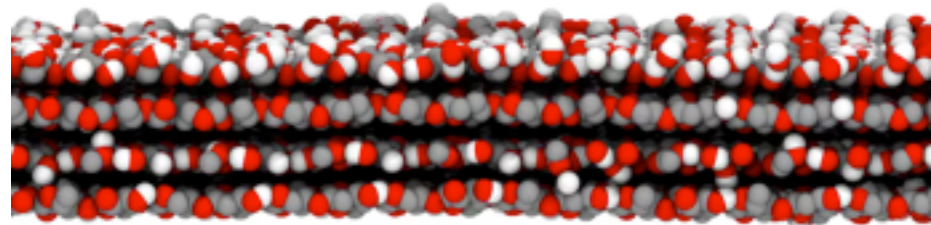
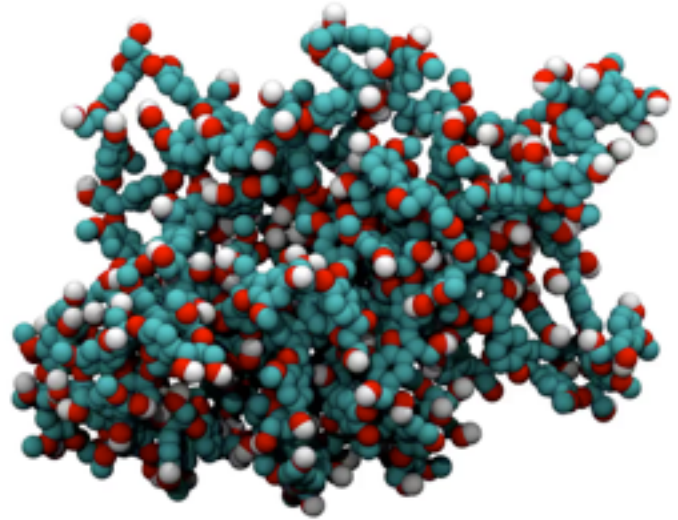
Molecule		Cellulose Face Binding Energy (kT)			
Linkage	Feature	200	110	1-10	010
β O4 α O4	RR	-5.42 \pm 1.00	-1.51 \pm 0.11	-1.02 \pm 0.09	-0.18 \pm 0.08
	RS	-3.48 \pm 0.24	-1.42 \pm 0.09	-0.81 \pm 0.07	-0.44 \pm 0.07
	SR	-5.30 \pm 1.00	-1.21 \pm 0.09	-0.33 \pm 0.06	-0.25 \pm 0.07
	SS	-2.71 \pm 0.16	-1.24 \pm 0.09	-1.10 \pm 0.08	-1.14 \pm 0.08
β O4	RR	-2.44 \pm 0.14	-1.36 \pm 0.09	-0.46 \pm 0.07	-1.05 \pm 0.08
	RS	-2.16 \pm 0.13	-0.69 \pm 0.08	-0.67 \pm 0.08	-0.18 \pm 0.08
	SR	-2.53 \pm 0.17	-0.91 \pm 0.09	-0.83 \pm 0.09	-0.07 \pm 0.08
	SS	-2.83 \pm 0.15	-0.47 \pm 0.08	-0.71 \pm 0.08	-0.80 \pm 0.08
55	CC	-2.76 \pm 0.14	-1.56 \pm 0.10	-1.01 \pm 0.07	-1.76 \pm 0.11
	CG	-2.07 \pm 0.11	-1.23 \pm 0.10	-0.47 \pm 0.07	0.10 \pm 0.08
	CP	-2.45 \pm 0.13	-1.70 \pm 0.09	-0.88 \pm 0.08	-1.02 \pm 0.09
	GC	-3.10 \pm 0.18	-1.22 \pm 0.10	-0.95 \pm 0.08	0.04 \pm 0.07
	GG	-2.33 \pm 0.13	-1.20 \pm 0.09	-0.07 \pm 0.07	-0.40 \pm 0.07
	GP	-2.67 \pm 0.15	-0.86 \pm 0.07	-0.38 \pm 0.08	-0.50 \pm 0.09
	PC	-1.60 \pm 0.09	-1.21 \pm 0.08	-0.93 \pm 0.08	-0.33 \pm 0.09
	PG	-3.00 \pm 0.17	-1.87 \pm 0.10	0.05 \pm 0.08	-0.92 \pm 0.08
	PP	-2.16 \pm 0.12	-1.21 \pm 0.08	-0.57 \pm 0.06	-0.78 \pm 0.08
4O5		-2.70 \pm 0.15	-1.23 \pm 0.09	-1.09 \pm 0.10	-0.75 \pm 0.08
Ester	α	-7.85 \pm 1.00	-2.65 \pm 0.14	-0.66 \pm 0.09	-2.11 \pm 0.11
	γ	-5.51 \pm 1.00	-2.36 \pm 0.13	-0.72 \pm 0.08	-2.36 \pm 0.15
β 1	RR	-0.55 \pm 0.08	-0.29 \pm 0.06	-0.20 \pm 0.07	-0.01 \pm 0.07
	RS	-0.86 \pm 0.07	-0.71 \pm 0.07	-0.54 \pm 0.07	-0.78 \pm 0.08
	SR	-1.66 \pm 0.10	-0.59 \pm 0.08	-0.40 \pm 0.08	-0.45 \pm 0.07
	SS	-0.88 \pm 0.08	-0.43 \pm 0.07	-0.24 \pm 0.07	0.01 \pm 0.08
β 5	RR	-5.73 \pm 1.00	-2.03 \pm 0.11	-0.84 \pm 0.09	-1.51 \pm 0.09
	RS	-4.24 \pm 0.35	-2.47 \pm 0.15	-0.50 \pm 0.08	-0.50 \pm 0.07
	SR	-2.74 \pm 0.18	-1.28 \pm 0.09	-0.38 \pm 0.08	-0.16 \pm 0.08
	SS	-3.65 \pm 0.24	-1.91 \pm 0.11	-1.06 \pm 0.08	-1.18 \pm 0.10
$\beta\beta$	RRRR	-1.96 \pm 0.11	-0.68 \pm 0.08	-0.55 \pm 0.08	-0.51 \pm 0.07
	RRRS	-1.72 \pm 0.10	-0.60 \pm 0.07	-0.41 \pm 0.07	-0.45 \pm 0.08
	RRSR	-2.35 \pm 0.14	-0.84 \pm 0.09	-0.37 \pm 0.08	-0.60 \pm 0.08
	RRSS	-2.68 \pm 0.14	-1.09 \pm 0.08	-0.63 \pm 0.08	-1.14 \pm 0.09
	RSRR	-1.76 \pm 0.10	-0.52 \pm 0.07	-0.37 \pm 0.07	-0.03 \pm 0.07
	RSRS	-2.64 \pm 0.16	-1.75 \pm 0.09	-0.31 \pm 0.07	-0.68 \pm 0.07
	RSSR	-3.32 \pm 0.22	-0.85 \pm 0.09	-0.62 \pm 0.07	-1.63 \pm 0.10
	RSSS	-2.51 \pm 0.17	-1.56 \pm 0.10	-0.62 \pm 0.07	-0.50 \pm 0.07

Cellulose Face-Dependent Binding Relationships



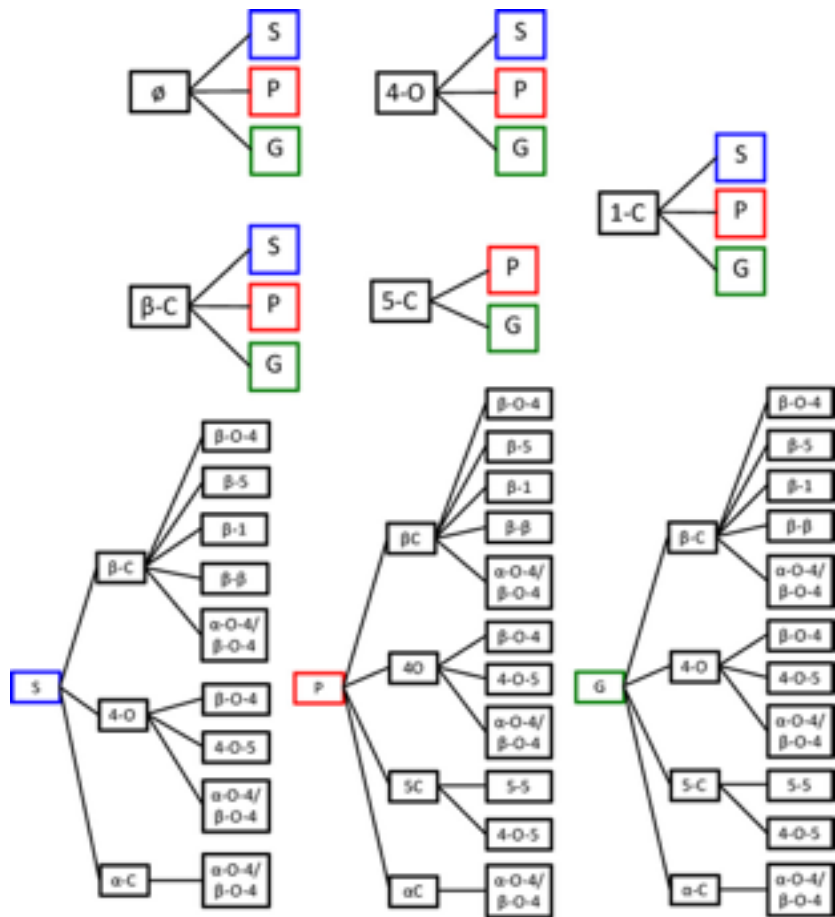
How Does Binding Scale with Polymer Size?

- Monomers and dimers are not perfect models for real lignin
- Draw from established lignin polymer libraries
- Determine lignin binding affinity



Building lignin

- The Broadbelt group has created a program to generate lignin topology libraries based on experimental observations
- From this library we will create 3D lignin models for spruce-like lignin with varying sizes



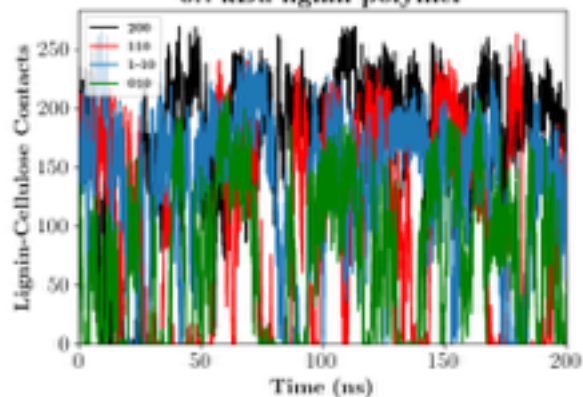
Brick by Brick Assembly

1. Generate monomers
2. Link them together in the specified way
3. Generate coordinates by superimposing successive dimeric structures
4. Fix any atomic overlaps/ring piercings
5. Solvate and simulate

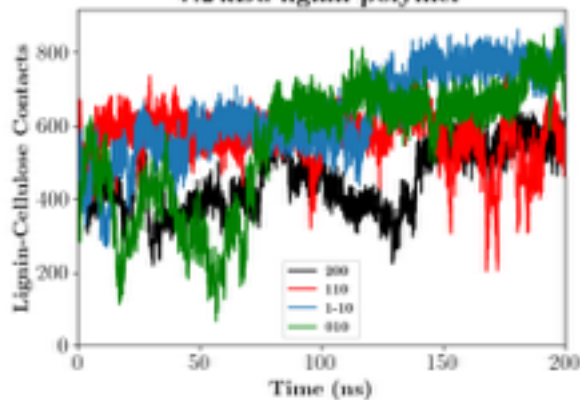


Unbiased Lignin Contacts to Cellulose

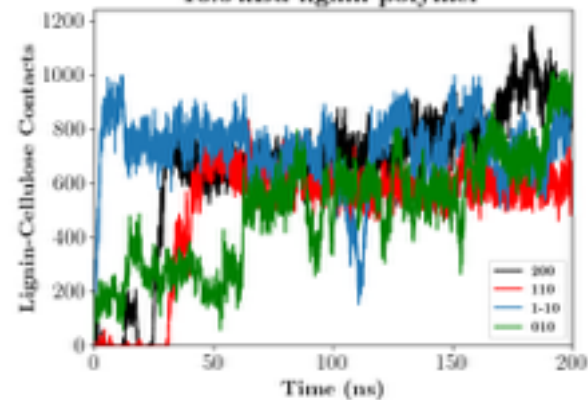
0.7 kDa lignin polymer



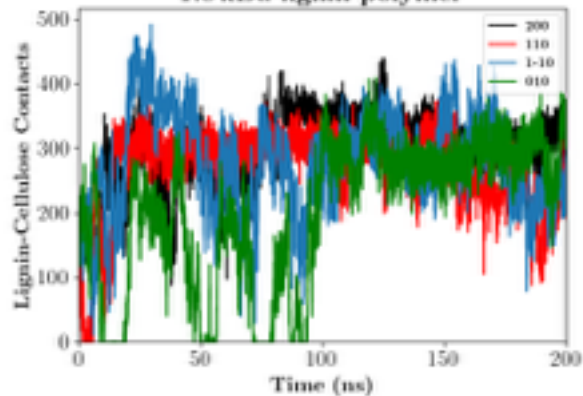
7.2 kDa lignin polymer



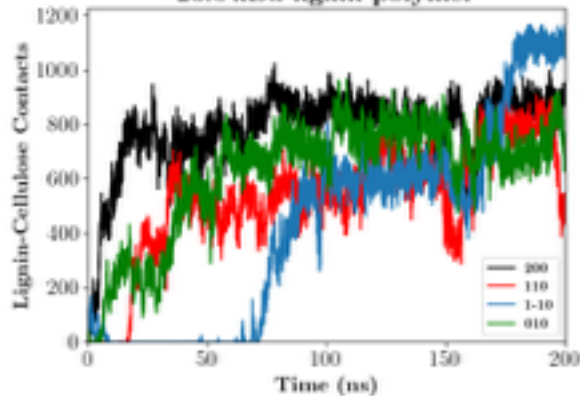
16.0 kDa lignin polymer



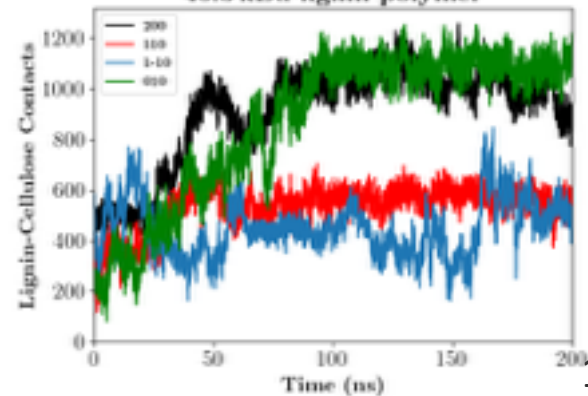
1.8 kDa lignin polymer



23.5 kDa lignin polymer



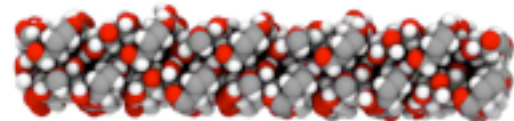
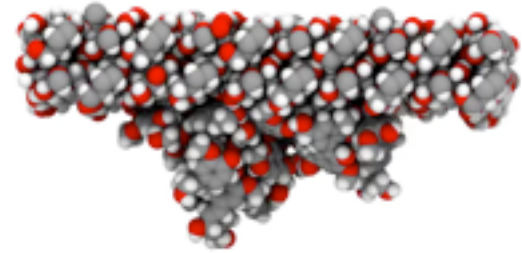
45.3 kDa lignin polymer



Technical Challenges

- Binding is spontaneous and irreversible on typical MD timescales for larger lignin
- Use replica exchange umbrella sampling on a contact-number reaction coordinate to estimate binding free energy

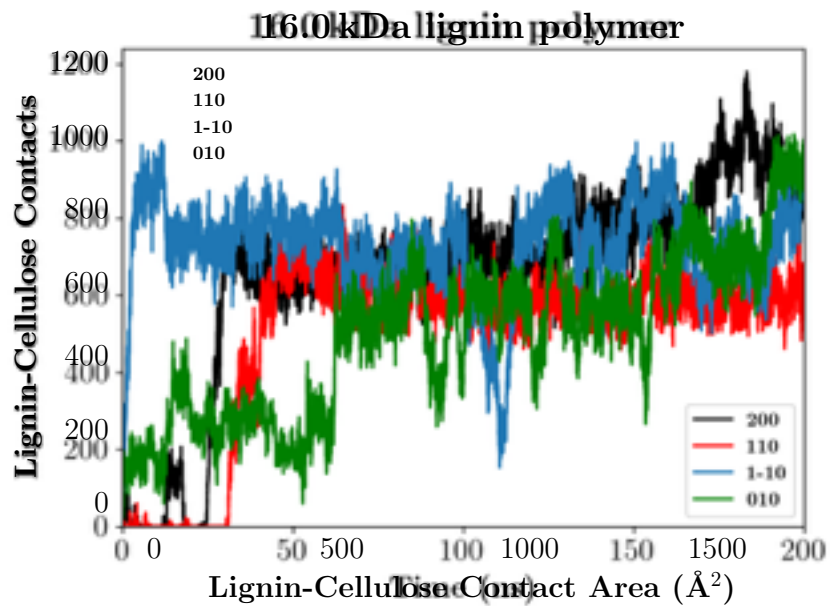
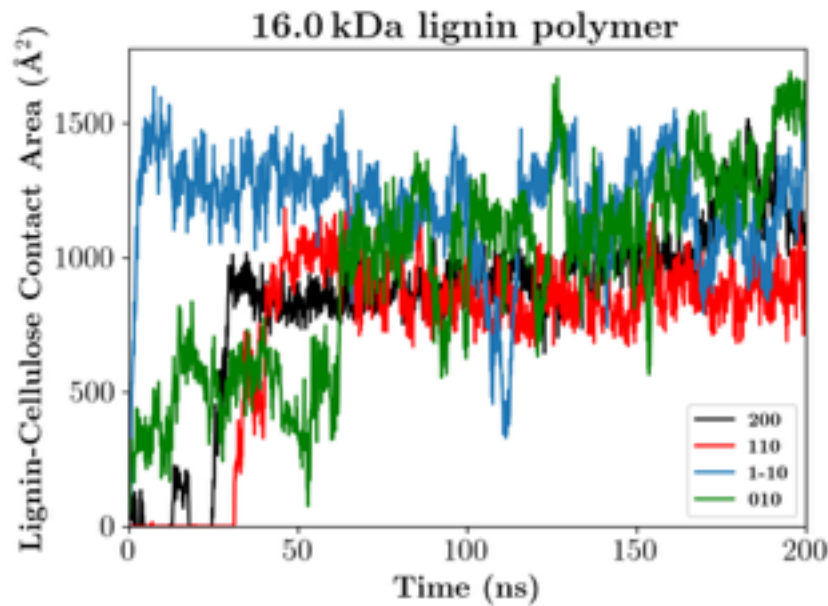
$$C = \sum_{i,j \in \text{pairs}} \frac{1 - (|x_i - x_j| / 12\text{\AA})^6}{1 - (|x_i - x_j| / 12\text{\AA})^{24}}$$



Surface Area Comparison

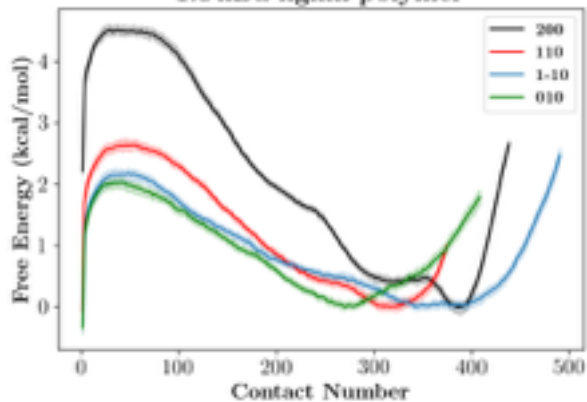
- Contact number as defined is a strong proxy for contact surface area between lignin and cellulose

$$C = \sum_{i,j \in \text{pairs}} \frac{1 - (|x_i - x_j| / 12\text{\AA})^6}{1 - (|x_i - x_j| / 12\text{\AA})^{24}}$$

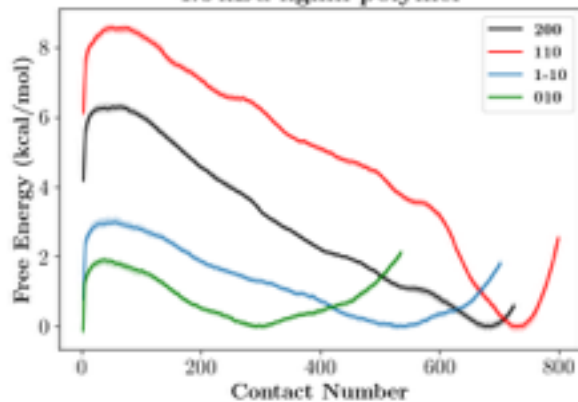


Binding Free Energy Profiles

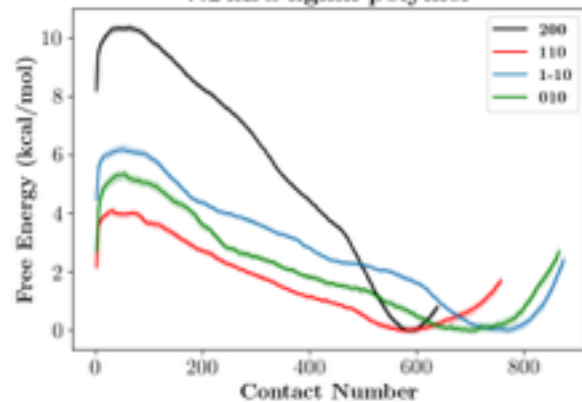
1.8 kDa lignin polymer



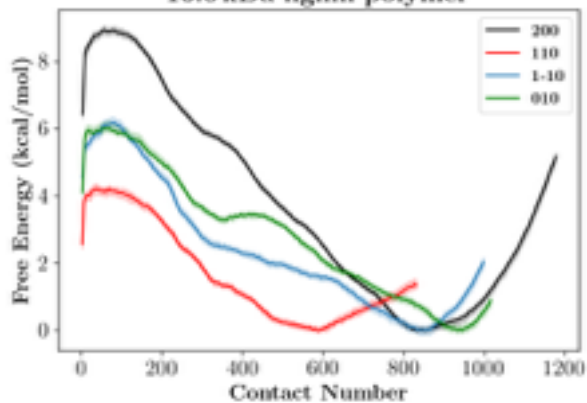
4.0 kDa lignin polymer



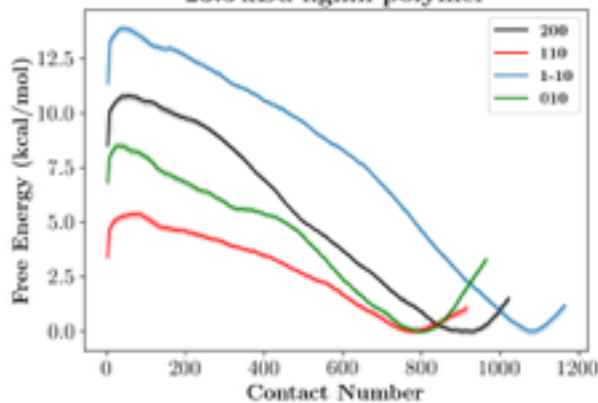
7.2 kDa lignin polymer



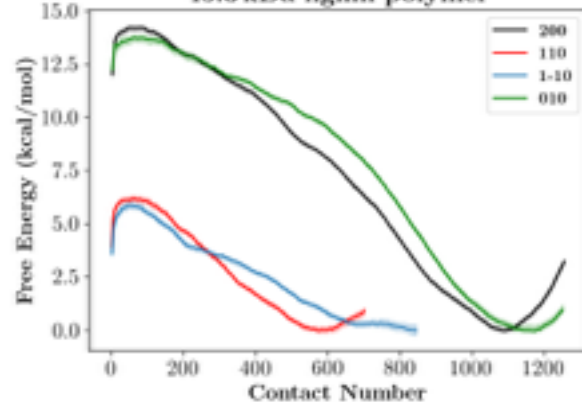
16.0 kDa lignin polymer



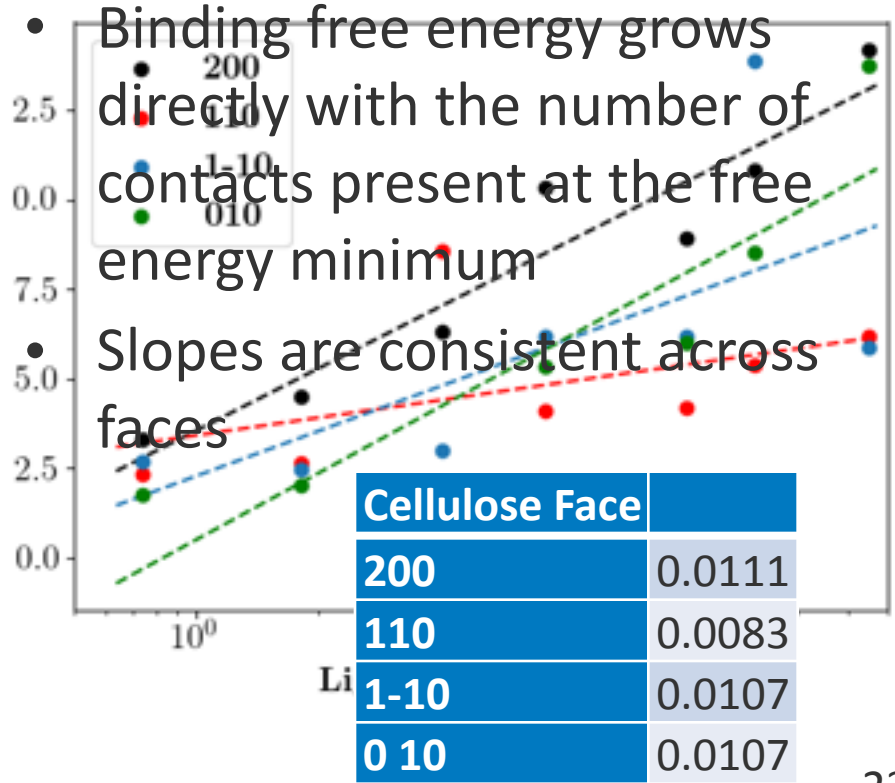
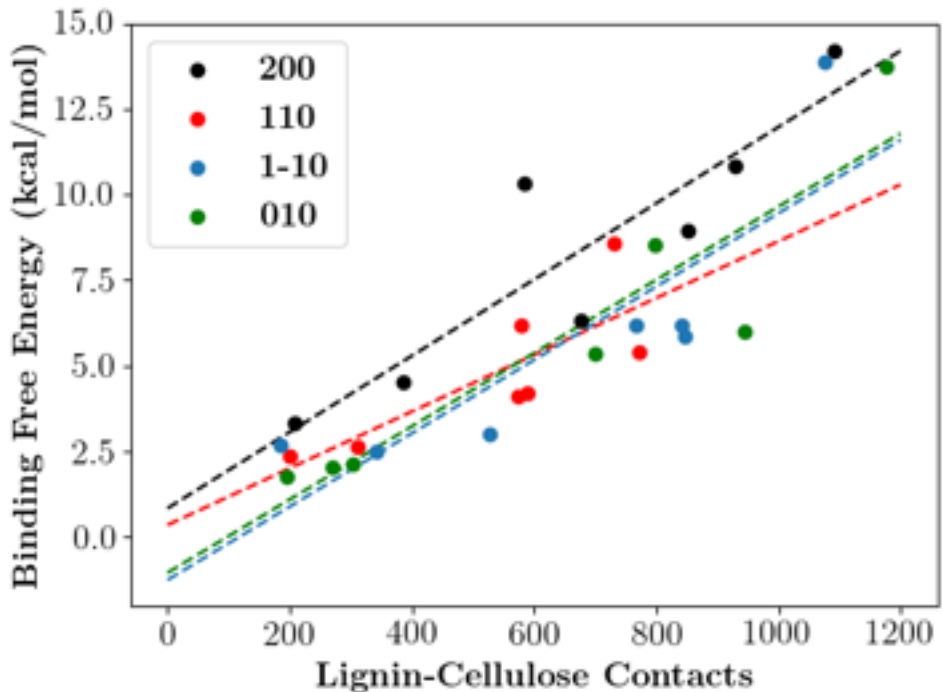
23.5 kDa lignin polymer



45.3 kDa lignin polymer



Size Dependence of Lignin-Cellulose Interactions



Summary

- Binding strength quantification can inform lignin modification
- Lignin related compounds prefer to bind to hydrophobic cellulose faces by about a factor of 2
 - May be related to the “flatness” of the 200 face, which fits well with planar aromatic rings, creating more nonspecific interaction sites
- Lignin contact number/contact area is more important than polymer size in determining binding strength
 - Lignin agglomeration is not considered, but would likely additively increase binding strength



Gregg Beckham

Michael Crowley

Questions?



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