

# BETO 2021 Peer Review: Enhancing Acetogen Formate Utilization to Value-Added Products 2.3.2.112

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Conversion  
March 11, 2021

# Market Trends

## Product



Anticipated decrease in gasoline/ethanol demand; diesel demand steady



Increasing demand for aviation and marine fuel



Increasing demand for renewable/recyclable materials

## Feedstock



Decreasing cost of renewable electricity



Sustainable waste management



Expanding availability of green H<sub>2</sub>



Closing the carbon cycle

## Capital



Challenges and costs of biorefinery start-up



## Social Responsibility



Carbon intensity reduction



## NREL's Bioenergy Program Is Enabling a Sustainable Energy Future by Responding to Key Market Needs

### Value Proposition

- How do we utilize CO<sub>2</sub> with cheaper renewable energy?
- What products can we make?

### Differentiator

- Liquid C1 compounds as medium for microbial upgrading
- Diversity of potential inputs
- Long term temporal storage, easy transport

# Quad Chart Overview

## Timeline

- 10/01/2018 through 9/30/2021

|             | FY20                     | Active Project        |
|-------------|--------------------------|-----------------------|
| DOE Funding | (10/01/2018 – 9/30/2021) | \$850,000 for 3 years |

## Barriers addressed

### Ct-H – C1 Fermentation Development

Liquid C1s are a novel and promising avenue for microbial conversion to bioproducts, but little is known about C1 liquid fermentation processes

### Ct-D – Advanced Bioprocess Development

Liquid C1s can be derived from a variety of renewable feedstocks and utilized in a variety of ways, but no industrial process exists

## Project Goal

- Develop acetogens as a platform for renewable liquid C1 conversion to value-added products
- Perform TEA/LCA analysis to understand CO<sub>2</sub> and economic considerations

## End of Project Milestone

- Demonstrate production of 2 g/L of C4 compound C1 feedstocks
- TEA/LCA analysis of potential process with generated fermentation metrics

## Funding Mechanism

Funded through BETO Conversion 2018 Lab call

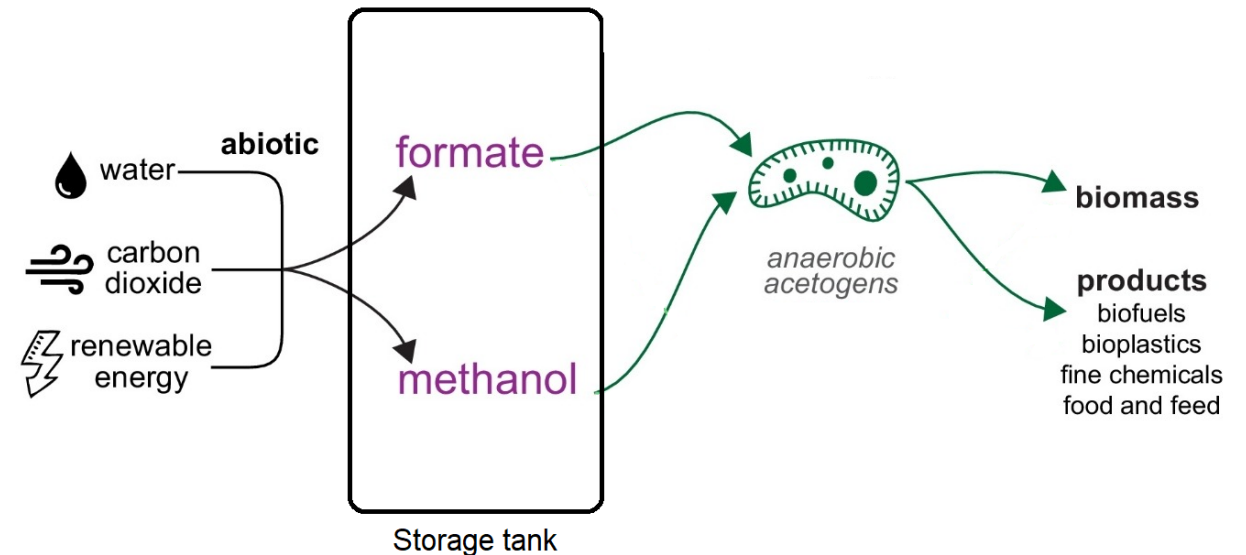


# Project Overview

- **Liquid C1 compounds represent an understudied avenue for renewable energy capture and CO<sub>2</sub> bioconversion**
  - Electro/thermochemical approaches have focused on syngas (H<sub>2</sub>, CO)
  - CO<sub>2</sub> capture to liquid feedstock (methanol/formate)
  - High energy density – Formate = 53 g/L H<sub>2</sub>, methanol = 100g/L H<sub>2</sub>
  - Easily stored/transported
  - Miscible - Avoiding mass transfer limits

- **Goal:** Develop a biological approach to convert liquid C1 into products

- Combine renewable **chemical CO<sub>2</sub> reduction** with **biological upgrading**



Renewable energy conversion of CO<sub>2</sub> combined with biological upgrading  
Adapted from Cotton et al. 2020 [j.copbio.2019.10.002](https://doi.org/10.1039/C9CB00022A)

# Project Overview

- **Relevance:** Chemicals with CO<sub>2</sub> and low-cost energy as feedstocks
  - Low-cost electricity to chemically reduce CO<sub>2</sub> to formate/methanol
  - Scalable strategy as a stand-alone process or value add to existing industry
- **Outcomes:**
  - **Proof of concept:**
    - General process outline
    - Feedstocks and Organism Identification/Characterization
    - Soluble C1 Conversion to C4 compounds at **2 g/L titer**
  - **Life cycle (LCA) and techno-economic analysis (TEA):**
    - Identify cost drivers and synergies with existing technologies

| Species         | Rate of Formation <sup>a</sup> | Selectivity <sup>b</sup> | Energy Efficiency <sup>c</sup> | Current TRL <sup>d</sup> |
|-----------------|--------------------------------|--------------------------|--------------------------------|--------------------------|
| Carbon Monoxide | High                           | High                     | High                           | High                     |
| Ethylene        | High                           | Medium                   | Low                            | Low                      |
| Formate         | Medium                         | High                     | Medium                         | Low                      |
| Methane         | High                           | High                     | Medium                         | High                     |
| Acetate         | Low                            | High                     | Medium                         | Low                      |
| Methanol        | High                           | High                     | High                           | High                     |

<sup>a</sup> High: >200 mA/cm<sup>2</sup> (or commercial TC), Medium: 200 >/>100 mA/cm<sup>2</sup>, Low: <100 mA/m<sup>2</sup>

<sup>b</sup> High: >80%, Medium 80% > FE > 60%, Low: < 60%

<sup>c</sup> High: >60%, Medium 60% > EE > 40%, Low: < 40%


<sup>d</sup> High: Operated at TRL > 6, Medium: Operated TRL 4-6, Low: Operated TRL 1-3

*Qualitative evaluation of product ease of formation. From [“Transforming the carbon economy: challenges and opportunities in the convergence of low-cost electricity and reductive CO<sub>2</sub> utilization”](#)*

# Project Overview

Perspective | Published: 11 January 2021

## An industrial perspective on catalysts for low-temperature CO<sub>2</sub> electrolysis

Richard I. Masel , Zengcai Liu, Hongzhou Yang, Jerry J. Kaczur, Daniel Carrillo, Shaoxuan Ren, Danielle Salvatore & Curtis P. Berlinguette

*Nature Nanotechnology* (2021) | [Cite this article](#)

“Units to convert CO<sub>2</sub> to formic acid are projected to reach pilot scale in the next year.”

| Methanol category          | Commercial  | Feasibility and R&D  |
|----------------------------|---|--|
| <b>Bio-methanol</b>        | <ul style="list-style-type: none"><li>■ BASF (GER)</li><li>■ BioMCN (NL)</li><li>■ Enerkem (CAN)</li><li>■ New Fuel (DEN)</li></ul> | <ul style="list-style-type: none"><li>■ Biogo (GER)</li><li>■ Enerkem (NL)</li><li>■ LowLands Methanol Heveskes Energy (NL)</li><li>■ NREL (USA)</li><li>■ Origin Materials (USA)</li><li>■ Södra (SE)</li></ul>   |
| <b>Renewable methanol</b>  | <ul style="list-style-type: none"><li>■ CRI (IC)</li><li>■ Innogy (GER)</li></ul>   | <ul style="list-style-type: none"><li>■ Advanced Chemical Technologies (CAN)</li><li>■ Asahi Kasei (JPN)</li><li>■ Blue Fuel Energy (CAN)</li><li>■ bse Engineering (GER)</li><li>■ Catalytic Innovations (USA)</li><li>■ CRI (CN/GER)</li><li>■ Gensoric (GER)</li><li>■ Infracore (GER)</li><li>■ Liquid Wind (SE)</li><li>■ MefCO2 (GER)</li><li>■ Neo-H2 (USA)</li><li>■ Port of Antwerp (BE)</li><li>■ Quantiam Technologies (CAN)</li><li>■ STEAG (GER)</li><li>■ Swiss Liquid Future (CH)</li><li>■ thyssenkrupp (GER)</li><li>■ USC (USA)</li><li>■ ZASt (GER)</li></ul> |
| <b>Low carbon methanol</b> | <ul style="list-style-type: none"><li>■ GPIC (BAH)</li><li>■ Methanex (CAN)</li><li>■ QAFAC (QAT)</li><li>■ SABIC (KSA)</li></ul>   | <ul style="list-style-type: none"><li>■ Carbon2Chem (GER)</li><li>■ FRESME (SE)</li><li>■ GasTechno (USA)</li><li>■ Haldor Topsoe (DEN)</li><li>■ Maverick Synfuels (USA)</li><li>■ NCF (CN)</li><li>■ OPTIMEoH (GER)</li></ul>  |

From [Methanol.org](https://methanol.org)

# Management

## Feedstocks

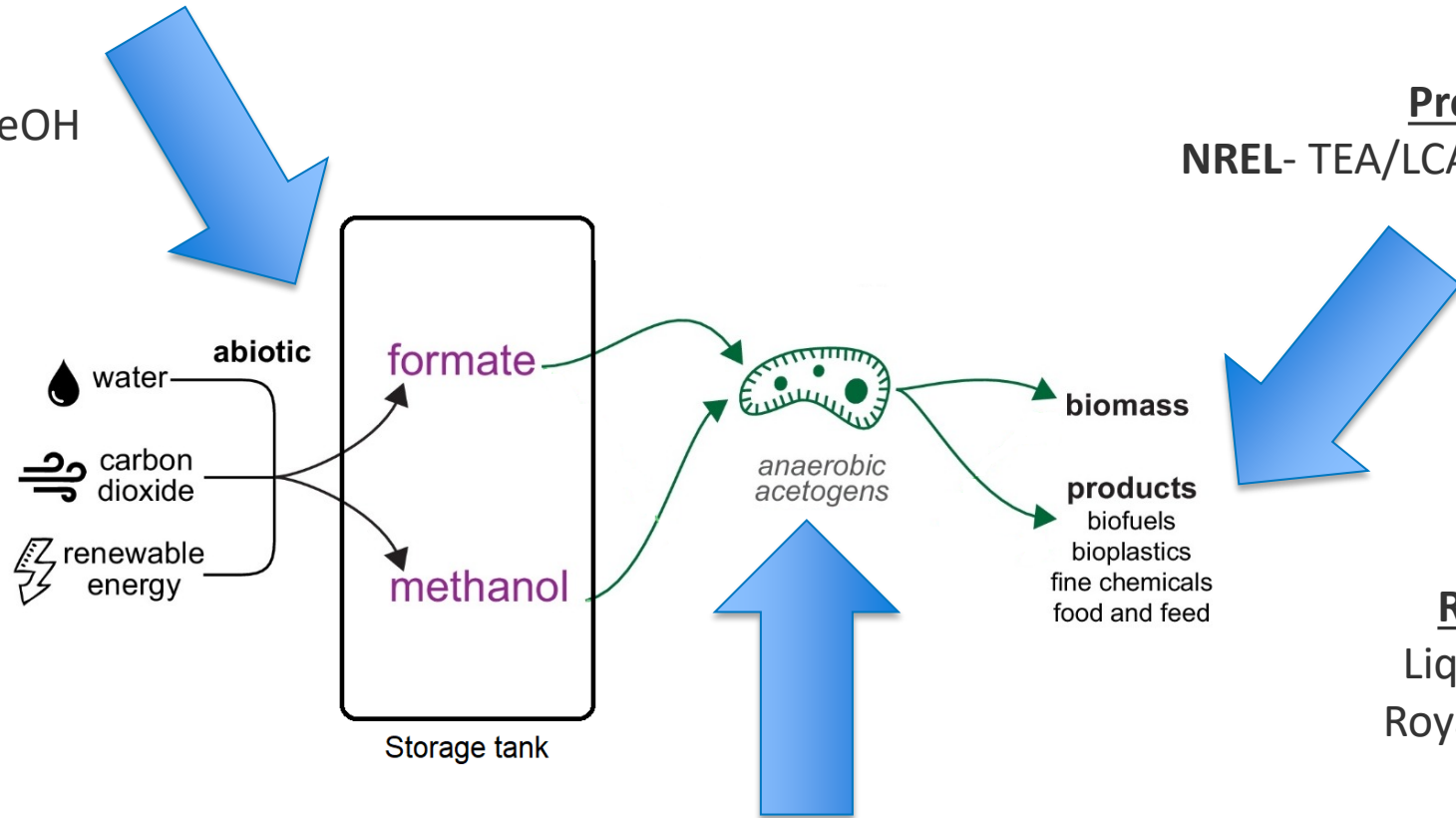
NREL- C1 Electrochemist

Kenneth Neyerlin

**Carbon Recycling**

**International - MeOH**

producer



## Products

NREL- TEA/LCA analysis Ling Tao

## Related Industry Contacts

Liquid C1 Conversion Startups

Royal Dutch Shell C1 conversion

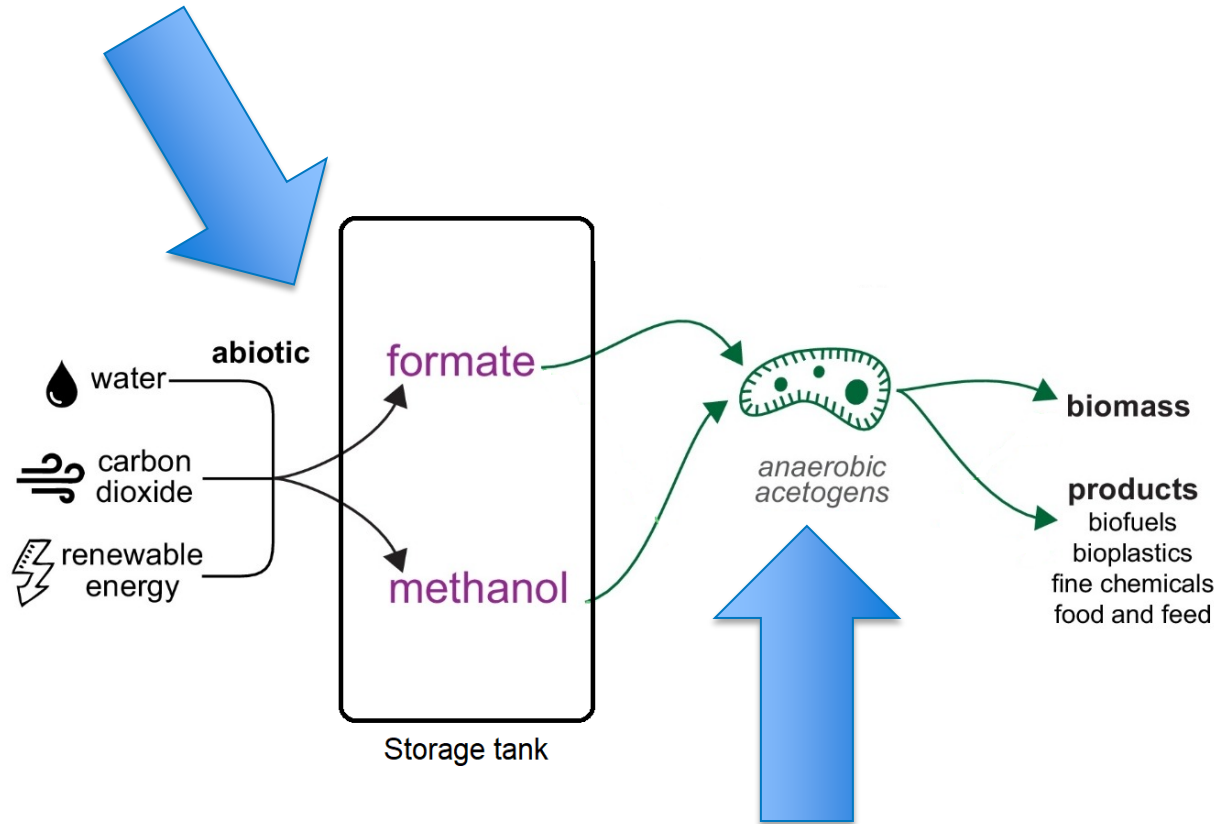
## C1 Organisms

NREL- Microbiologist Jonathan Lo

# Management

## Methanol and Formate Feedstocks

How are they generated?  
H<sub>2</sub>/CO<sub>2</sub> synergy?  
At what prices?



## Products

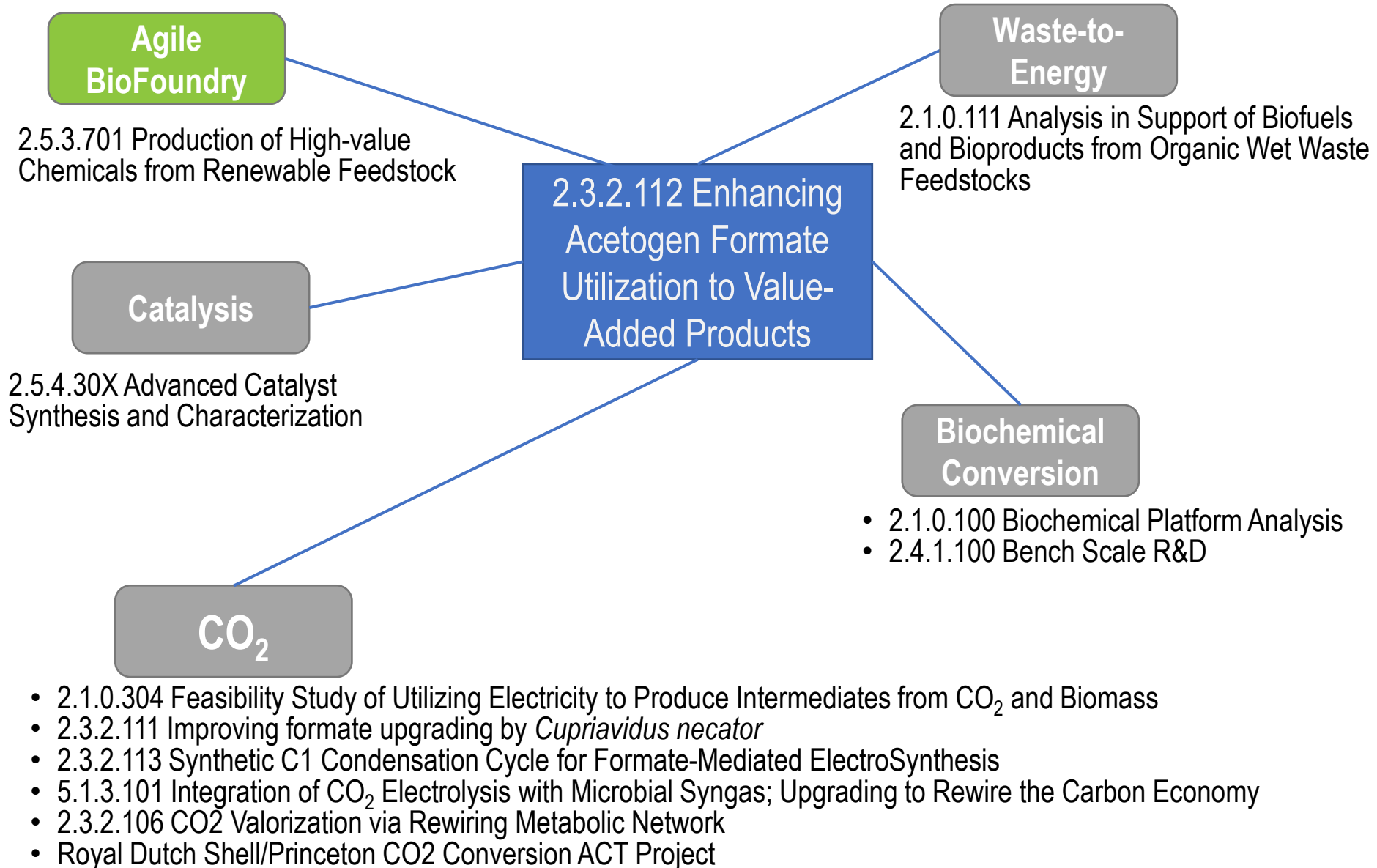
What products can be made?  
Cost? Market price?  
Carbon and electron efficiency?  
Upgrading paths and market size?

## C1 Microbes

Who can use them? How well?  
What do they make?  
Can they be genetically engineered?



# Approach and Management

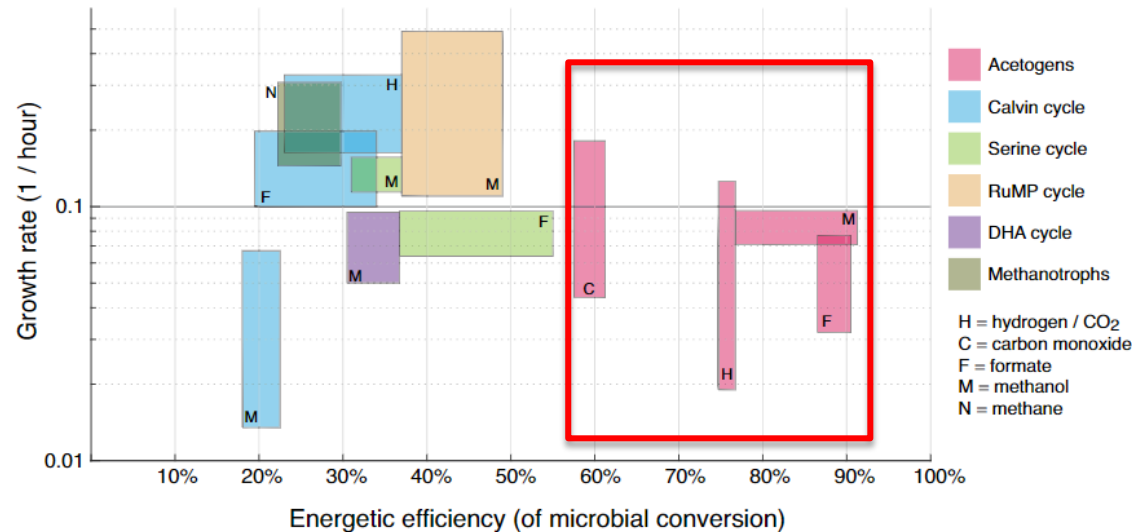


# Approach: Highest biological efficiency for CO<sub>2</sub> fixation

## Renewable methanol and formate as microbial feedstocks

Charles AR Cotton<sup>1</sup>, Nico J Claassens<sup>1</sup>, Sara Benito-Vaquerizo<sup>1</sup> and Arren Bar-Even

Current Opinion in Biotechnology 2020, 62:168–180

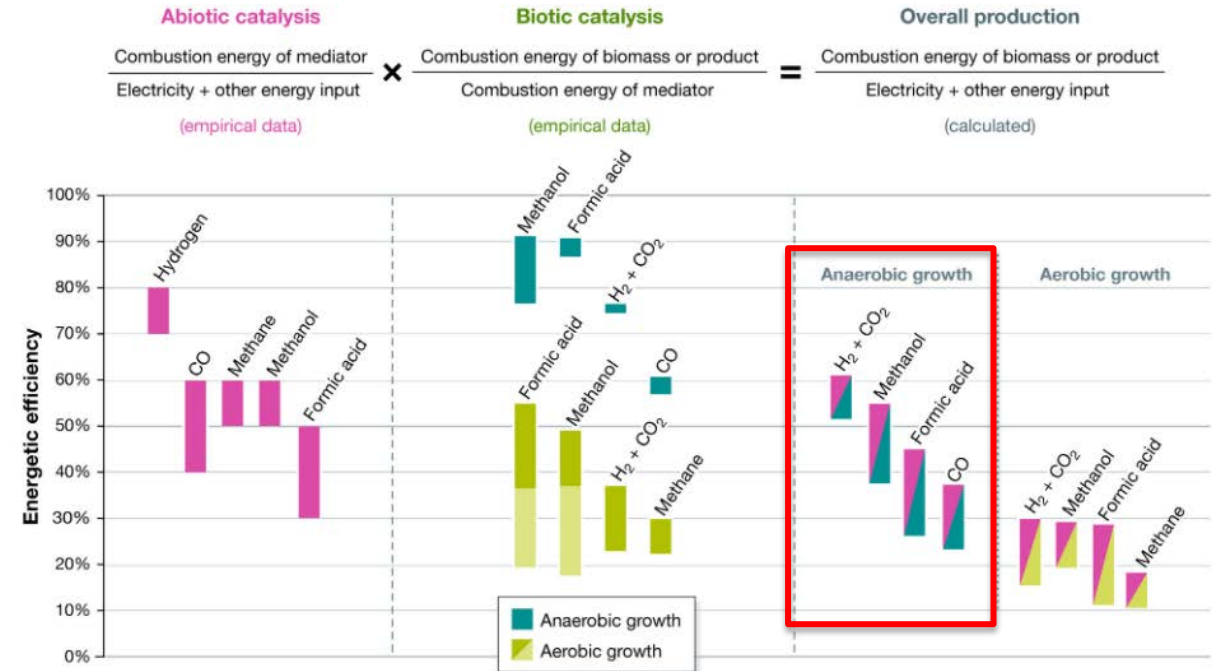


Current Opinion in Biotechnology

## A one-carbon path for fixing CO<sub>2</sub>

Ari Satanowski, Arren Bar-Even  

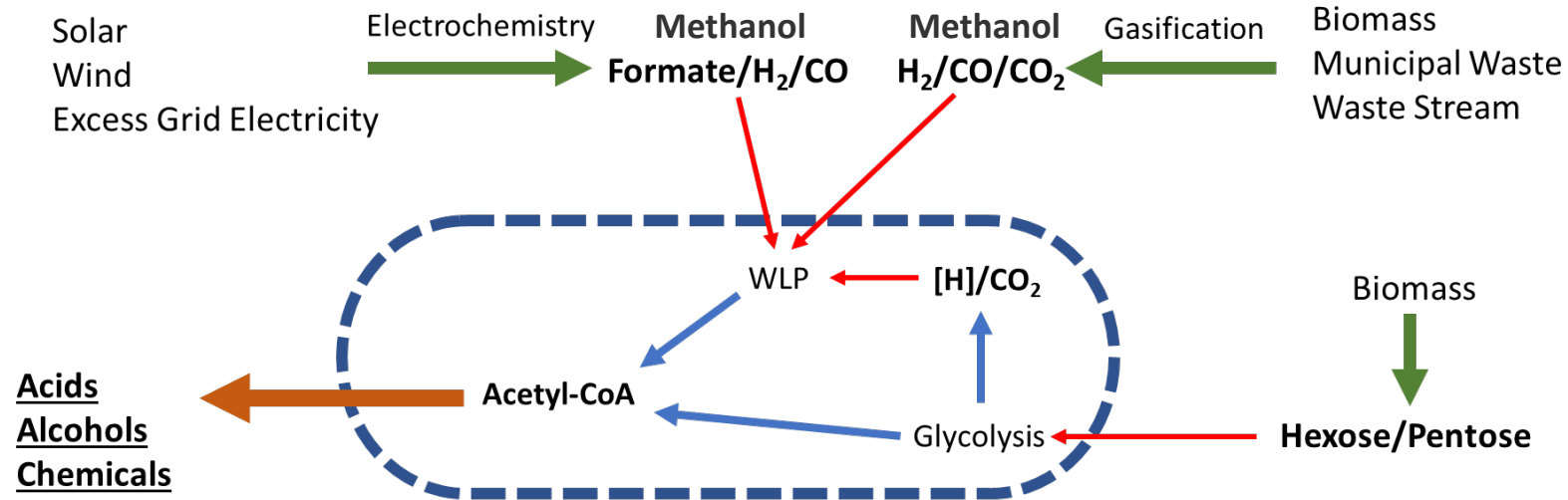
EMBO Rep (2020)21:e50273



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“Bioproduction with acetogens is thoroughly researched and commercially exploited using gaseous C<sub>1</sub> feedstocks...only a small number of acetogens have been tested for growth on methanol and formate...miscible carbon sources support higher energetic efficiencies of bioproduction”

# Approach: Acetogens for CO<sub>2</sub> fixation



Acetogens **non-photosynthetically, anaerobically** fix CO<sub>2</sub>

- Use Wood Ljungdahl Pathway (WLP), **most efficient for CO<sub>2</sub> fixation**
- Investigated for syngas conversion, but can use liquid C1 formate and methanol
- Avoids gas mass transfer issue, easier to store and transport
- Can simultaneously use gases, liquids, and biomass related sugars
- Produce interesting products at high carbon and electron efficiency
- Focus on C4 products (butanol/butyrate) due to their ease of upgrading to fuels

# Approach - Milestones

- Transform C4 overexpression pathway into acetogen to boost yield of C4 products (Q1)
- Growth  $\geq 1$ L reactor for TEA/LCA metrics analysis (Q2)
- TEA/LCA analysis of butanol/butyrate production with different feedstock mixes to determine cost drivers, carbon efficiencies, product separation, purification, and upgrading (Q3)

## End goal

- Proof of concept: Demonstrate production of 2 g/L of a C4 compound in an engineered acetogen using 1-carbon feedstocks (Q4)

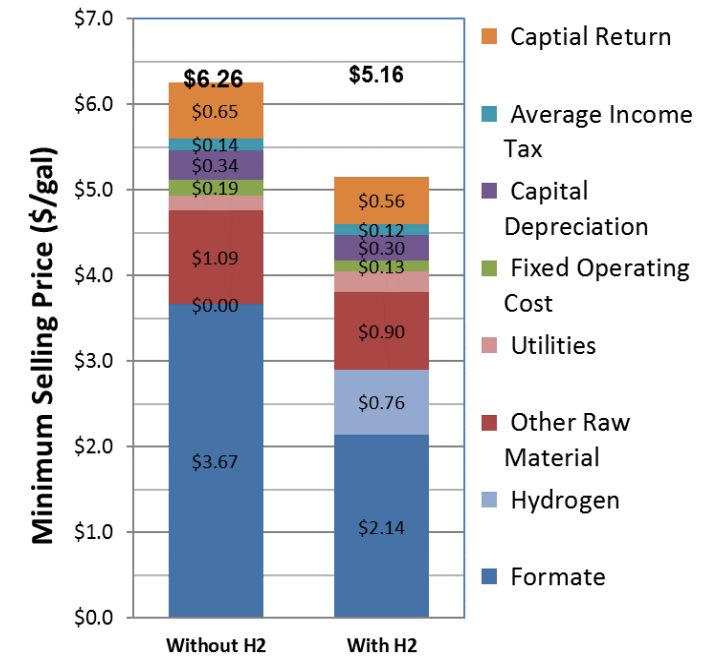
# Progress and Outcomes

## TechnoEconomic Analysis

### Assumptions for CO<sub>2</sub> Reduction to Formate

| Parameters                                     | Value   |
|--|---------|
| Cell Voltage (V)                               | 2.08    |
| Current Density ( mA/cm <sup>2</sup> )         | 300     |
| Faradaic Efficiency (%)                        | 95      |
| CO <sub>2</sub> Single-pass Conversion (%)     | 90      |
| CO Faradaic Efficiency (%)                     | 0       |
| H <sub>2</sub> Faradaic Efficiency (%)         | 5       |
| Electrolyzer Capital Cost (\$/m <sup>2</sup> ) | 10,000  |
| Electricity Price (\$/kWh)                     | 0.03    |
| CO Market Price (\$/kg)                        | 0.23    |
| H <sub>2</sub> Price (\$/kg)                   | 1.57    |
| Water Price (\$/kg)                            | 0.00022 |
| CO <sub>2</sub> Capture Price (\$/ton)         | 20      |

### Preliminary Minimum Butanol Selling Price (\$/gal)



- Formate is a poor electron source
- H<sub>2</sub> improves Carbon yield
- Butyrate versus butanol?
- Methanol is cheap, electron rich, soluble
- From methane or electrochemically
- Potential cosubstrate
- C1 miscible

| Equation   | [C] Yield | Mass Yield |
|--|-----------|------------|
| $C_6H_{12}O_6$ (sugar) $\rightarrow$ $C_4H_{10}O_2$ (butanol) + $2CO_2$ + $H_2$      | 0.67      | 0.41       |
| $12 CH_2O_2$ (formate) $\rightarrow$ $C_4H_{10}O$ (butanol) + $8CO_2$ + $7H_2O$      | 0.33      | 0.13       |
| $7 CH_2O_2$ (formate) + $5 H_2 \rightarrow C_4H_{10}O$ (butanol) + $3CO_2$ + $7H_2O$ | 0.57      | 0.23       |
| $4 CH_2O_2$ (formate) + $8 H_2 \rightarrow C_4H_{10}O$ (butanol) + $7H_2O$           | 1.00      | 0.40       |

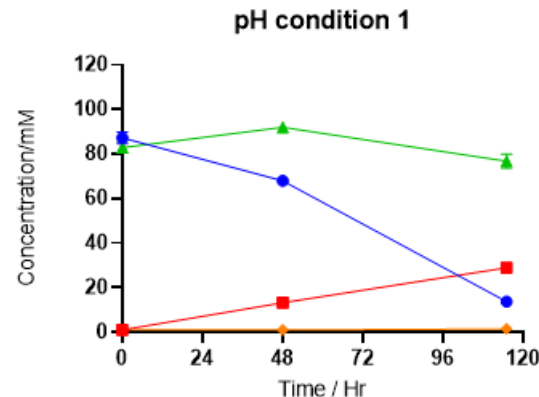
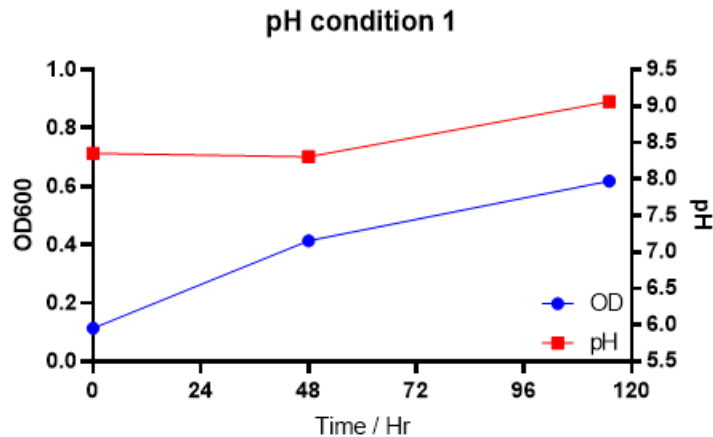
|          | USD/kg \$ | MW     | Mass yield g/g | Input cost \$ per kg | % Cost reduction |
|----------|-----------|--------|----------------|----------------------|------------------|
| Methanol | \$0.29    | 32.04  | 0.58           | \$0.49               |                  |
| Glucose  | \$0.29    | 180.16 | 0.41           | \$0.70               | 30%              |
| Butanol  | \$1.20    | 74.12  |                |                      |                  |



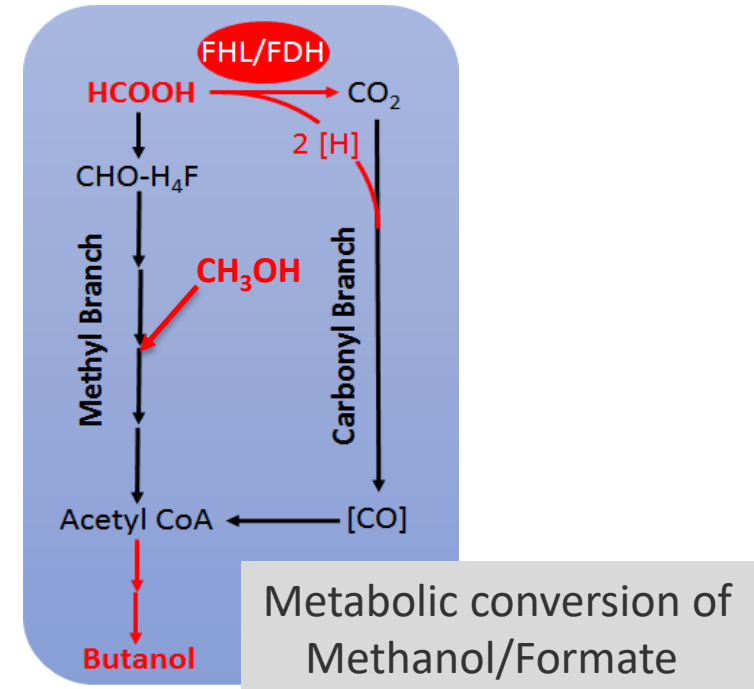
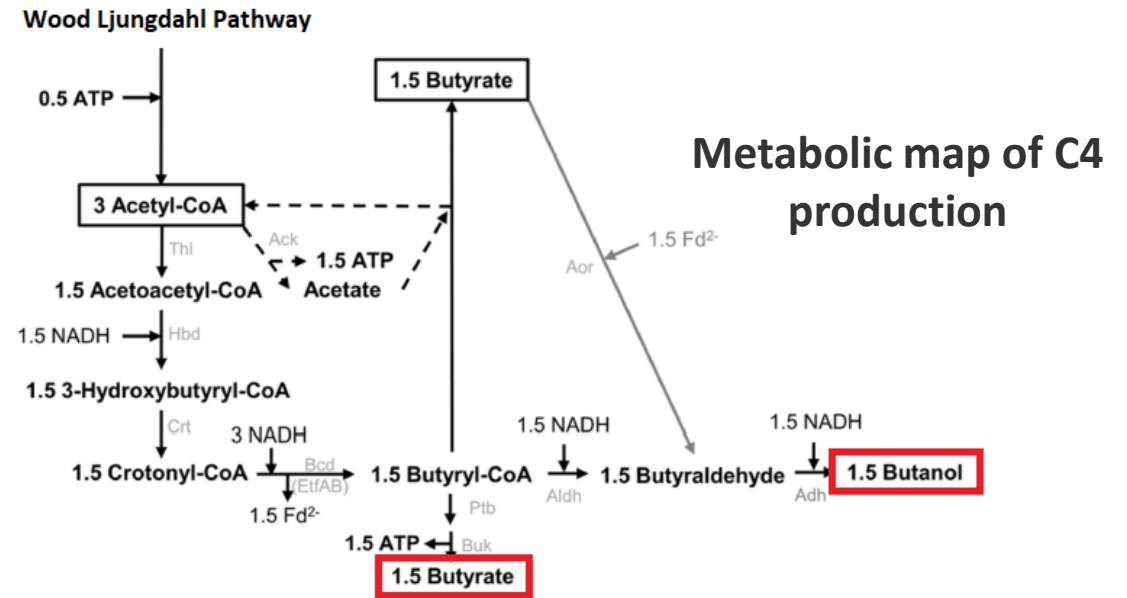
# Progress and Outcomes

## C1 conversion

- *Butyribacterium methylotrophicum* (Bm)
  - Can use formate/methanol alone
  - No genetic tools
  - No metabolic models
  - Acetate/ethanol/butyrate/butanol



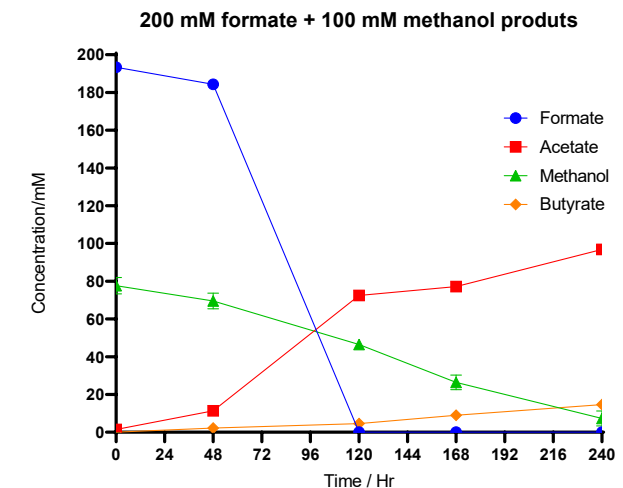
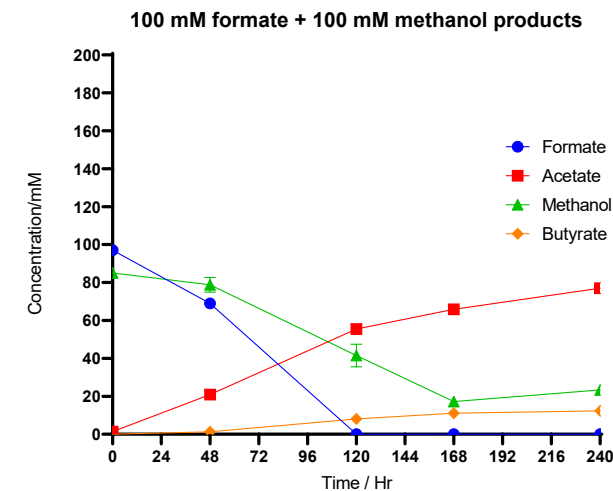
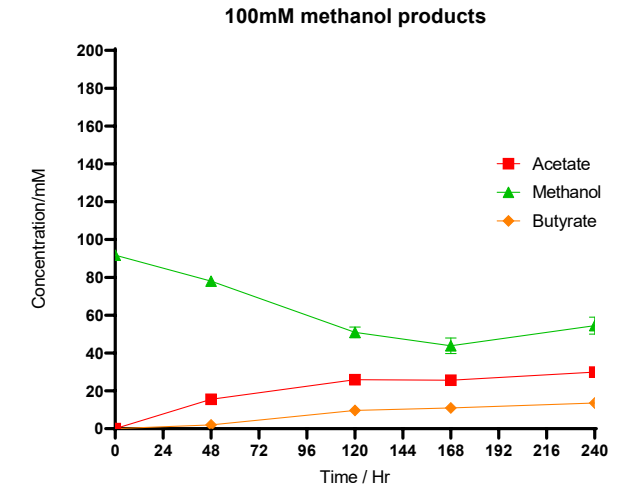
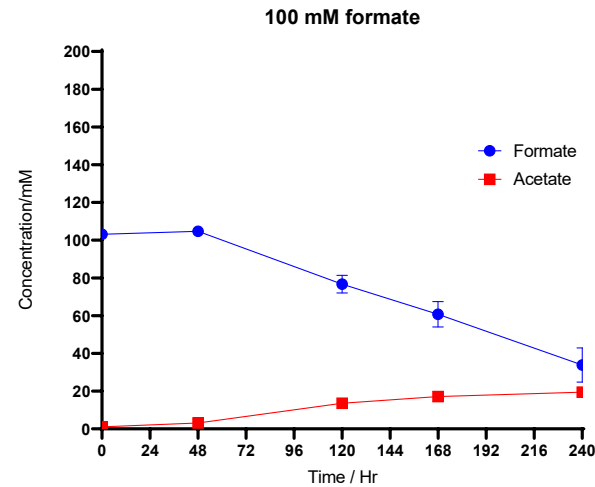
Growth and production on formate/methanol



# Progress and Outcomes

## C1 conversion specifics

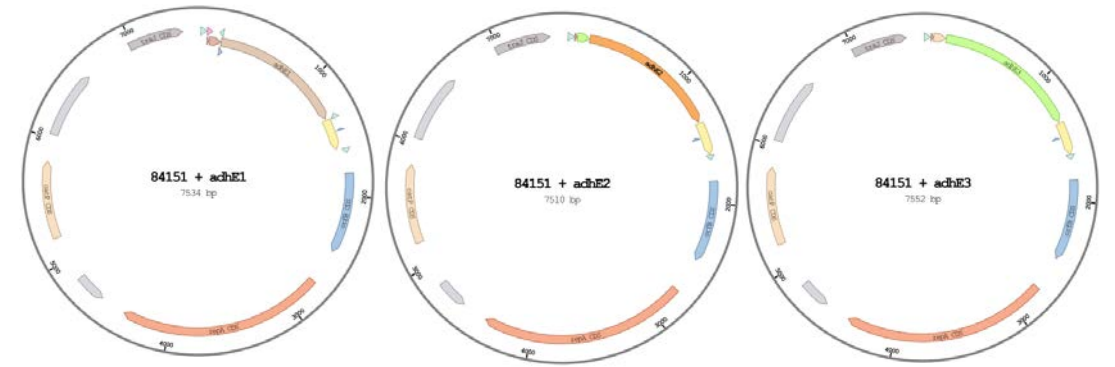
- **Bm** can naturally make C4 (butyrate/butanol)
  - 100 mM formate 200 mM MeOH
  - 41 mM (3.6 g/L) butyrate
- Methanol seems to drive formation for butyrate, but does not make butanol
- Different C1 mixtures and conditions for different product formation
- Single substrate C1 fermentation is slow
- C1 mix is synergistic
  - Tolerance (~500mM formate, >1M methanol)
  - Consumption kinetics (yields, rates, titers)



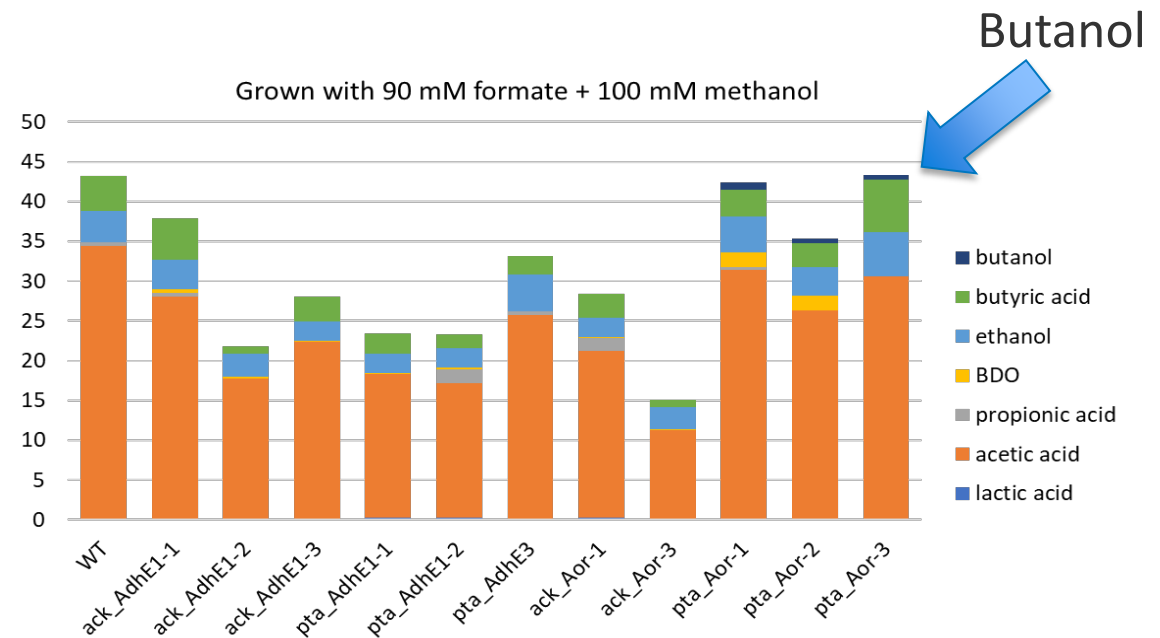
# Progress and Outcomes

## Organism development

- Genetics tools development
  - Protocol established
  - Building CRISPR/Cas9 plasmids
  - Butanol production strategies
  - Promoter testing
- Metabolic analysis
  - RNAseq analysis
  - Substrate specificity
  - Product profile
  - Linking genes to products



Plasmids expressing different alcohol dehydrogenases (AdhE). AdhE is known to convert aldehydes into alcohols and is needed to create ethanol and butanol.



Growth of *B. methyltrophicum* engineered strains on 90 mM formate and 100 mM methanol.

# Progress and Outcomes

## Organism development

- Test tube scale
  - 3.6 g/L butyrate
  - 50% Carbon efficiency
- Bioreactor scale up
  - Fermentation control
  - Experimenting with conditions
  - Feeding strategies
  - Improving yield/rate/titer
  - Capturing metrics for second TEA/LCA analysis



# Impact –Data and Dissemination

- Direct CO<sub>2</sub> conversion to C1 chemicals has a high technology readiness level, but C1 chemicals have a low market price.
- Larger compounds C2-C4 have a higher value
- Direct contact with Royal Dutch Shell, and renewable Methanol Carbon Recycling International (CRI)
- Several forthcoming publications around microbial C1 conversion.
  - Genetic engineering and describing metabolism
    - Describing techniques, new tools, analysis
  - Metrics regarding C1 fermentation
    - Yield/rate/titer, fermentation strategies to change products
  - Fermentation process using real liquid C1s from CO<sub>2</sub>

| Feedstocks |          | USD/kg | \$/mole |
|------------|----------|--------|---------|
| C1         | Methanol | \$0.29 | \$0.009 |
| C1         | CO2      | \$0.00 | \$0.000 |
| C1         | Formate  | \$1.00 | \$0.046 |
| Products   |          | USD/kg | \$/mole |
| C2         | Acetate  | \$0.80 | \$0.048 |
| C4         | Butyrate | \$1.50 | \$0.132 |

*Feedstock and product market costs.*



# Impact – Future Work

- Liquid C1 fermentation strategies
  - Implementing real CO<sub>2</sub> reduction streams
  - Fed batch, continuous, in situ extraction
  - Supplement CO/H<sub>2</sub> for better growth and carbon efficiency?
- Direct production of methanol conversion to C4?
  - Methanol is cheap (\$276/MT), readily available, derived from methane, or electrochemically renewable from CO<sub>2</sub> (Vulcanol).
- Engineering *B. methyltrophicum* to make C4 (butyrate /butanol) at higher yield
  - Target native pathways, adding C4 pathways, planning RNAseq experiment
  - CRISPR/Cas9 gene deletion, genomic integration
- **Proposals to further develop process, explore variations, outside partner collaboration**

# Summary

## Product



Anticipated decrease in gasoline/ethanol demand; diesel demand steady



Increasing demand for aviation and marine fuel



Increasing demand for renewable/recyclable materials

## Feedstock



Decreasing cost of renewable electricity



Sustainable waste management



Expanding availability of green H<sub>2</sub>



Closing the carbon cycle

## Capital



Challenges and costs of biorefinery start-up



## Social Responsibility



Carbon intensity reduction



## NREL's Bioenergy Program Is Enabling a Sustainable Energy Future by Responding to Key Market Needs

### Value Proposition

- CO<sub>2</sub> conversion to liquid C1 for microbial upgrading represents an interesting proposition for renewable energy integration with CO<sub>2</sub> as a feedstock

### Key Accomplishments

- Identified and developed microbial C1 conversion to C2 and C4 products
- Developed a TEA/LCA analysis for understanding the process
- Filling in knowledge gaps and disseminating knowledge among academic and industry institutions

## NREL

- Jonathan Humphreys
- Lauren Magnusson
- Holly Rohrer
- Yi Pei Chen
- Wei Xiong
- Ling Tao
- KC Neyerlin
- Pin Ching Maness

# Thank You

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**[www.nrel.gov](http://www.nrel.gov)**

NREL/PR-2700-79300

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Program 2.3.2.112

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# Responses to Previous Reviewers' Comments

This project aims to convert formate to butanol using *Clostridia ljungdahlii*. Formate is one of the target intermediates that can be produced via chemical synthesis. It is not clear whether butanol is the best target product considering their value and subsequent separation issues etc. Thus, a better system engineering should be incorporated to fully evaluate the proposed technology.

**Response:** The TEA analysis is utilized to evaluate which products could best be made from formate. In our system, formate is first converted into acetyl-CoA, which is a precursor to many other products that could be made instead of formate, including ethanol, butyrate, and mevalonate, which could be made instead of butanol.

As a benchmark, assuming 100% formate conversion and 3 V for the electrochemical cell producing formate, the electricity demand will be ~26 kWh per kg butanol. The energy demand for state-of-the-art formate producing cells is substantially more than the thermodynamic minimum. Even if electricity is cheap, the extra cost of an inefficient energy conversion may make the process uncompetitive.

**Response:** Having high conversion to product is an important consideration for reaching economically feasible. It may be that we need higher efficiency of formate conversion to products, and that may be through co feeding other substrates like H<sub>2</sub>/CO to better efficiencies so that the formate carbon and electrons are better matched towards products.

# Publications, Patents, Presentations, Awards, and Commercialization

We anticipate at least 2 publications from this project in progress:

*Butyribacterium methylotrophicum* C1 liquid conversion characterization via RNAseq analysis and bioreactor data

Development and Genetic Engineering of *Butyribacterium methylotrophicum* as a chassis organism for conversion of C1 compounds to Value Added products