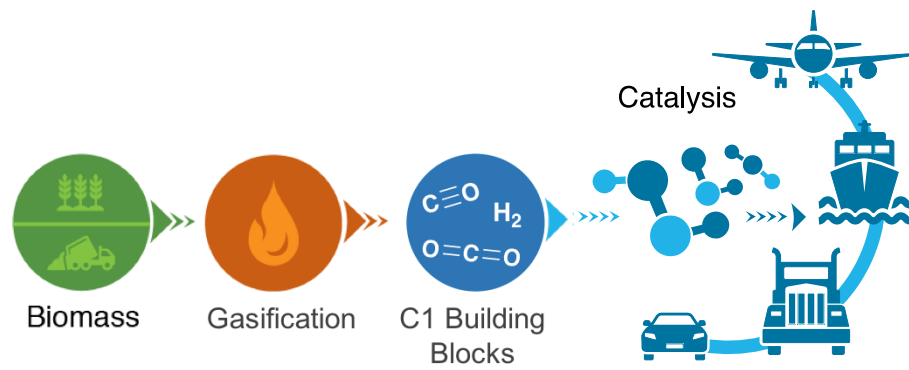


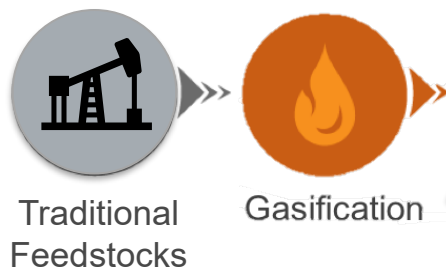
Process Intensification for Direct Conversion of Biomass-Based Syngas to High Octane Gasoline

Claire Nimlos, Connor Nash, Anh To, Dan Dupuis, Daniel Ruddy*

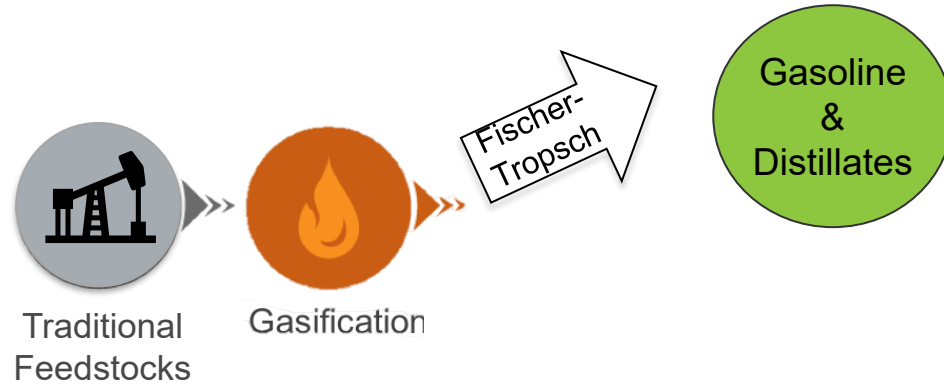
Catalytic Carbon Transformation & Scale-up Center, National Renewable Energy Laboratory



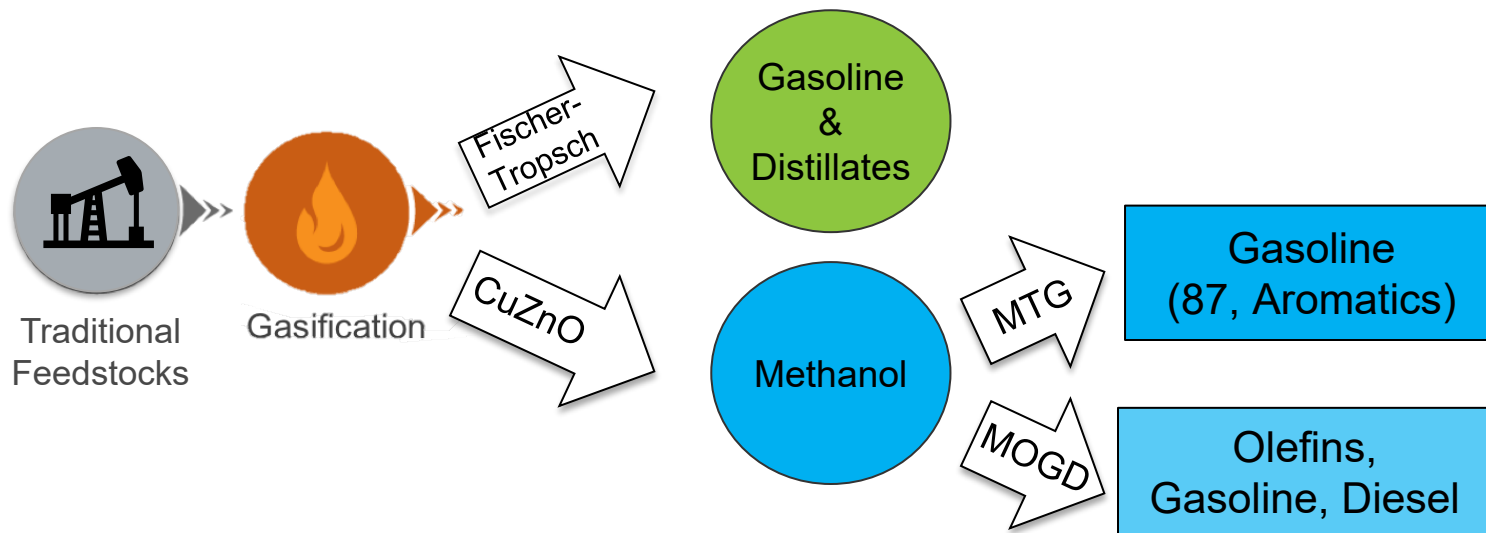
Traditional processes for syngas-to-fuels



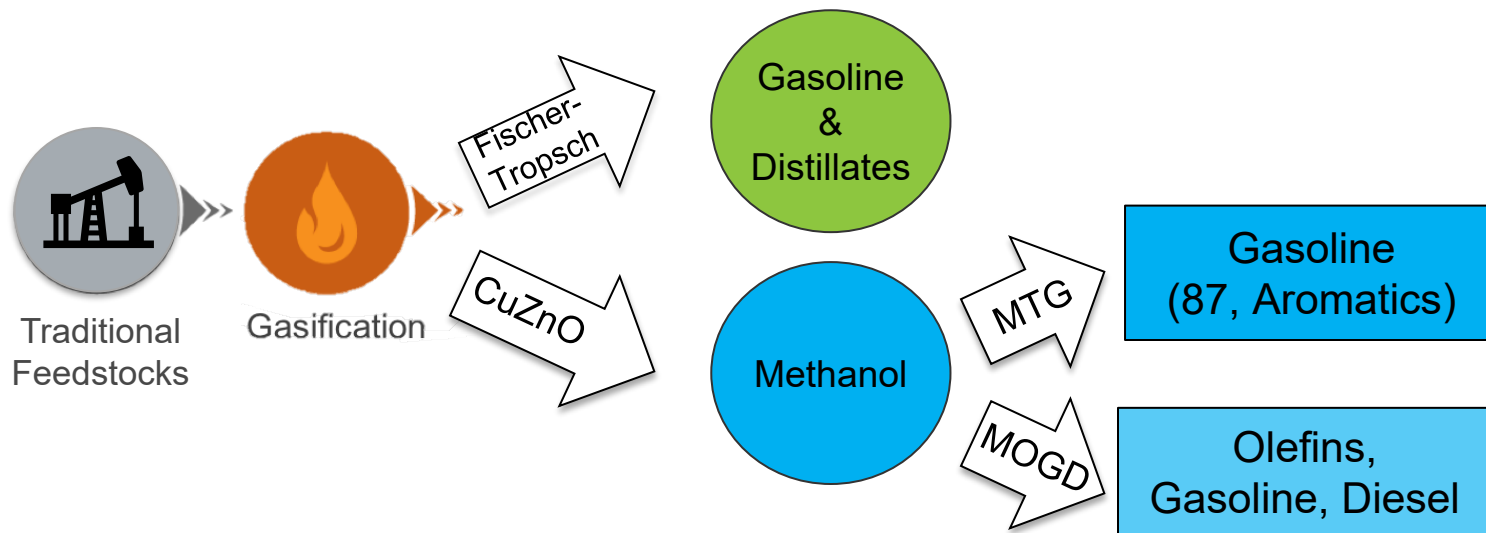
Traditional processes for syngas-to-fuels



Traditional processes for syngas-to-fuels

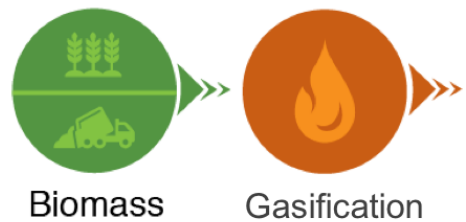


Traditional processes for syngas-to-fuels



*Traditional syngas to fuels have known drawbacks, with reduction in process complexity and separations while producing **higher value** products necessary*

Conversion of biomass to sustainable hydrocarbon products



Fuel Product

Fuel gas

HOG
C₅₋₈

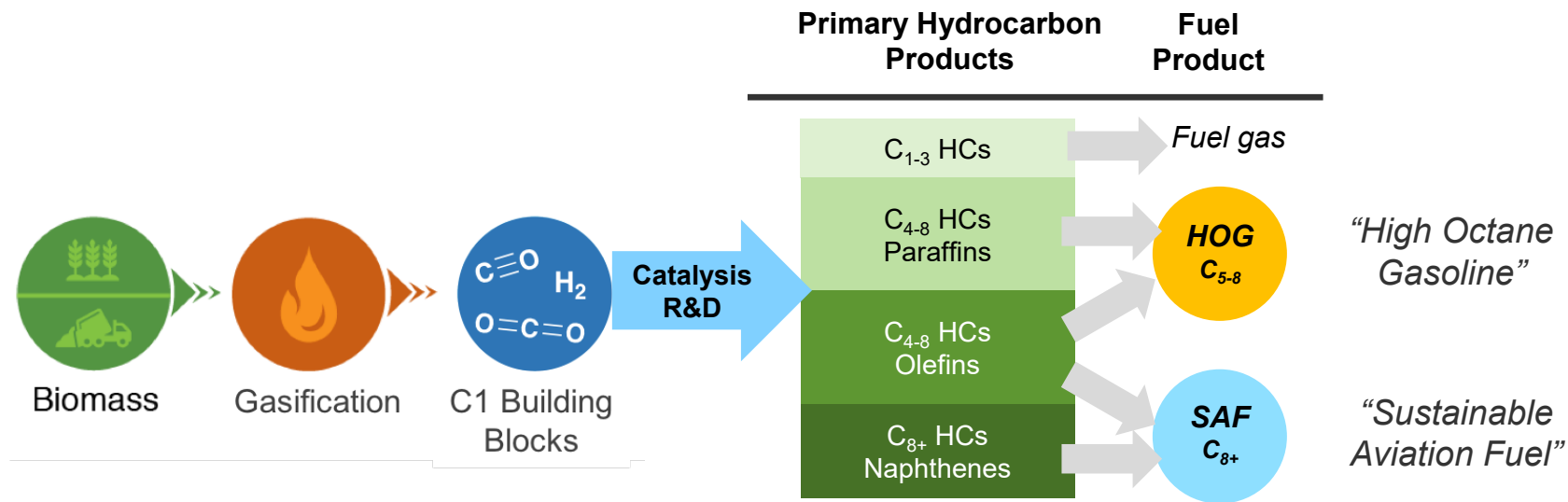
“High Octane Gasoline”

SAF
C₈₊

“Sustainable Aviation Fuel”

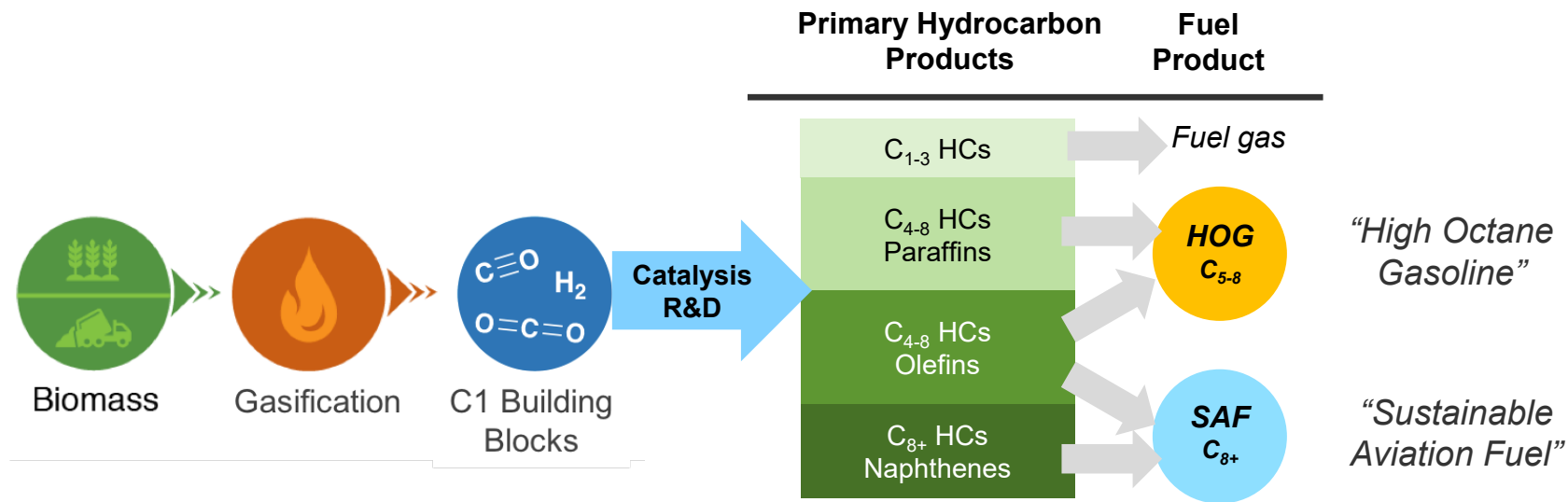
D. A. Ruddy et al. *Nat. Catal.*, **2019**, 2, 632

Conversion of biomass to sustainable hydrocarbon products



D. A. Ruddy et al. *Nat. Catal.*, 2019, 2, 632

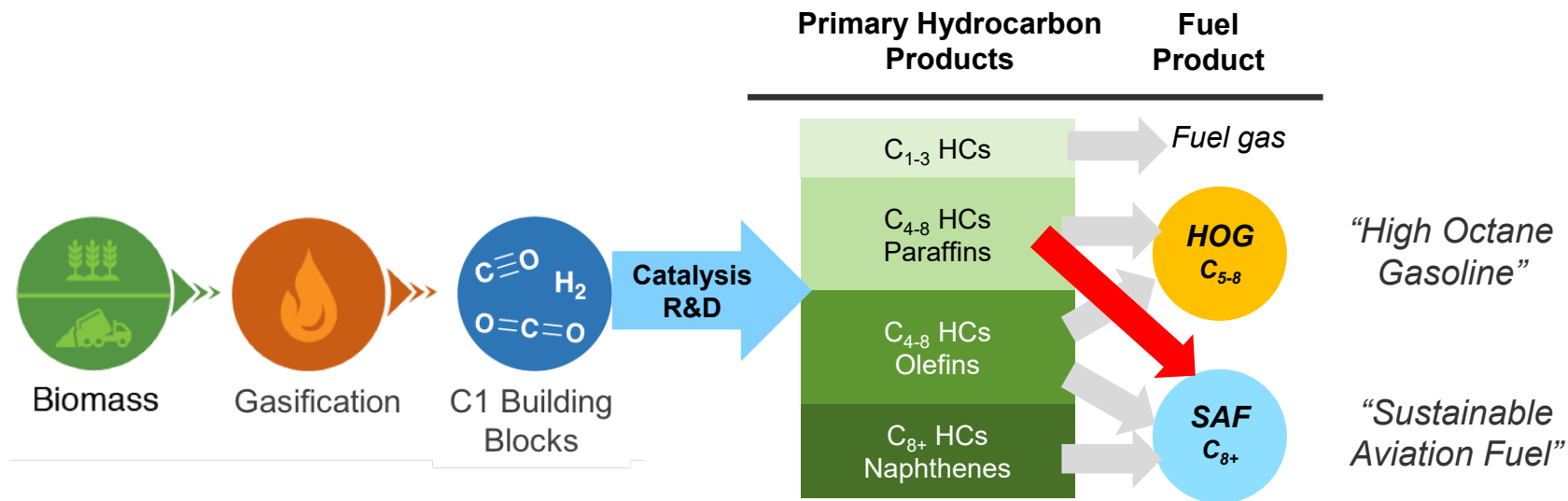
Conversion of biomass to sustainable hydrocarbon products



D. A. Ruddy et al. *Nat. Catal.*, **2019**, 2, 632

Development of a biorefinery process to convert renewable C1 intermediates into a suite of fuel products to meet market demand

Conversion of biomass to sustainable hydrocarbon products

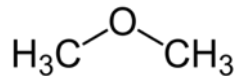


D. A. Ruddy et al. *Nat. Catal.*, **2019**, 2, 632

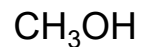
Development of a biorefinery process to convert renewable C1 intermediates into a suite of fuel products to meet market demand

DME homologation on acid zeolites

Dimethyl
Ether (DME)



or Methanol



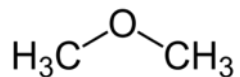
Ahn et al., *Angew. Chem.*, **2009**, 48, 3814

Simonetti et al., *J. Catal.*, **2011**, 277, 173

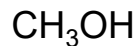
Simonetti et al., *Chem. Cat. Chem*, **2011**, 3, 704

DME homologation on acid zeolites

Dimethyl
Ether (DME)

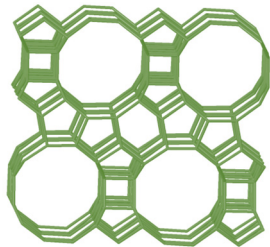


or Methanol

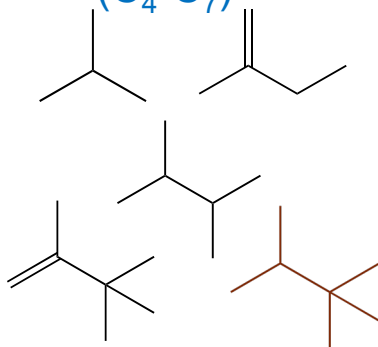


175-225 °C,
100-1000 kPa

Large-pore
acidic zeolites

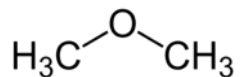


Branched
hydrocarbons
(C₄-C₇)



DME homologation on acid zeolites ...

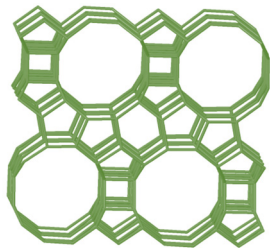
Dimethyl
Ether (DME)



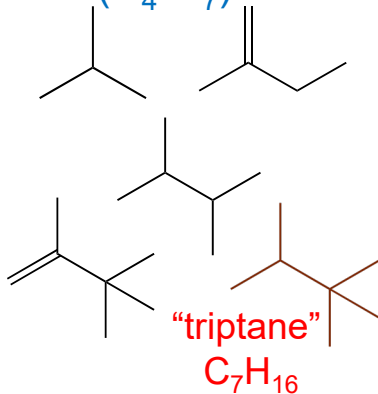
or Methanol
 CH_3OH

175-225 °C,
100-1000 kPa

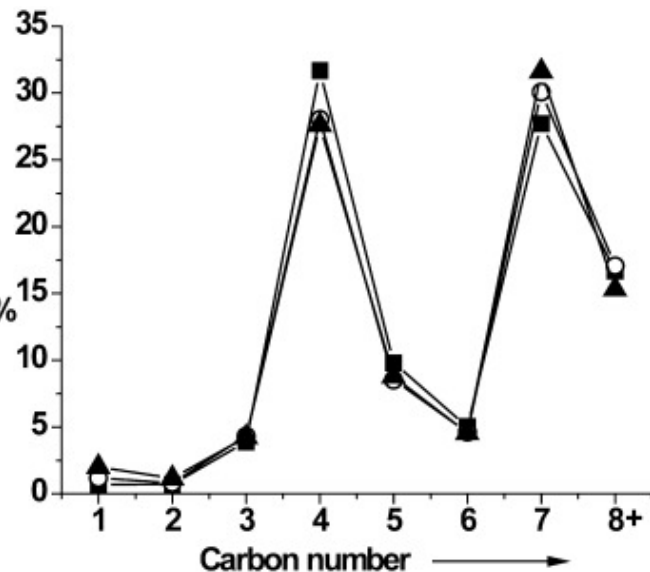
Large-pore
acidic zeolites



Branched
hydrocarbons
(C_4 - C_7)



↑
Carbon
selectivity %



Beta zeolite higher selectivity to C_7 due to stabilization of the bulky transition state in larger cage/intersections

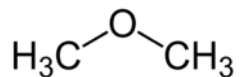
Ahn et al., *Angew. Chem.*, **2009**, 48, 3814

Simonetti et al., *J. Catal.*, **2011**, 277, 173

Simonetti et al., *Chem. Cat. Chem.*, **2011**, 3, 704

DME homologation on acid zeolites and the limitations

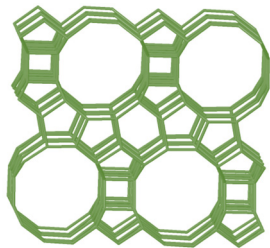
Dimethyl
Ether (DME)



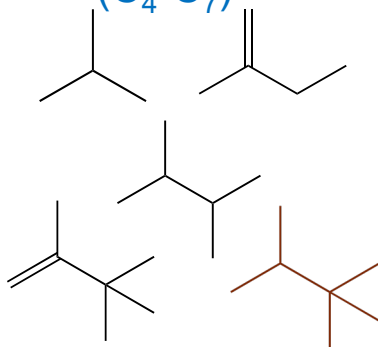
or Methanol
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175-225 °C,
100-1000 kPa

Large-pore
acidic zeolites



Branched
hydrocarbons
(C_4 - C_7)



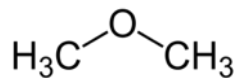
Hydrogen Deficiency



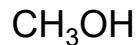
Need an additional 2H per
alkane produced

DME homologation on acid zeolites and the limitations

Dimethyl Ether (DME)

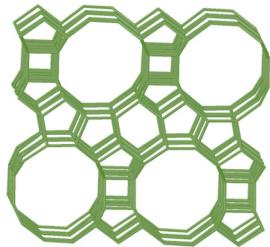


or Methanol

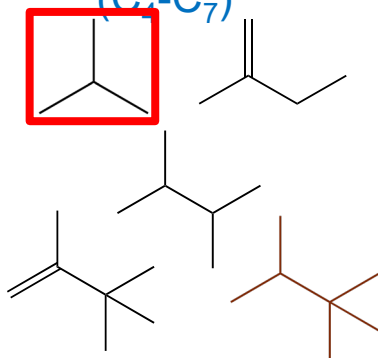


175-225 °C,
100-1000 kPa

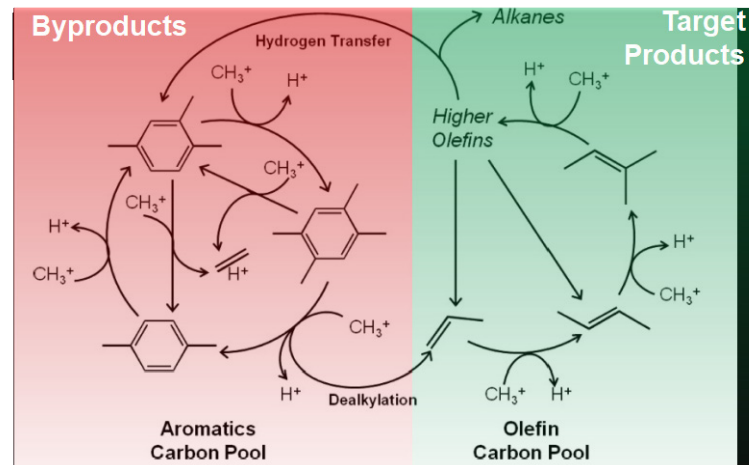
Large-pore
acidic zeolites



Branched
hydrocarbons
(C₄-C₇)



Mechanism



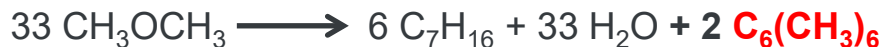
S. Ilias, A. Bhan, *ACS Catal.*, **2013** 3,18

Hydrogen Deficiency

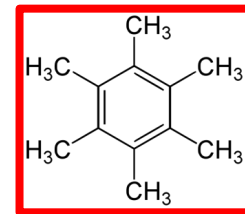


Need an additional 2H per
alkane produced

Yield Loss – aromatics and C₄



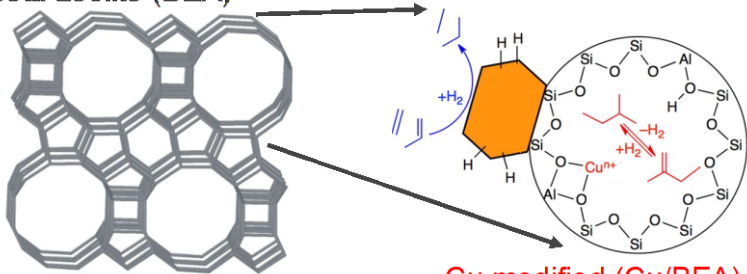
H/BEA leads to formation of heavy
unsaturated hydrocarbons or terminal
products like isobutane



“HMB”

Cu-modified beta zeolites for DME conversion

Beta zeolite (BEA)



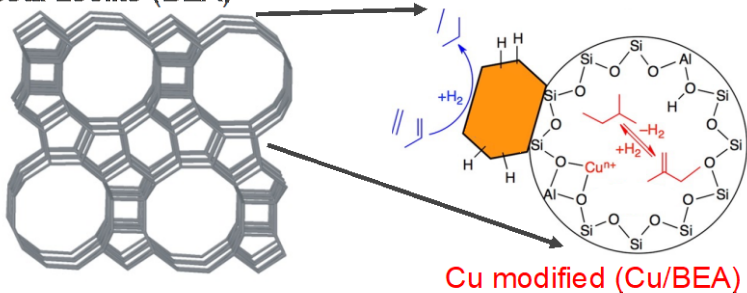
Cu modified (Cu/BEA)

J. A. Schaidle et al. *ACS Catal.*, **2015**, 5, 1794

US Pat # 9,803,142 B1
US Pat App 62/482,315
US Pat App 62/515,087

Cu-modified beta zeolites for DME conversion

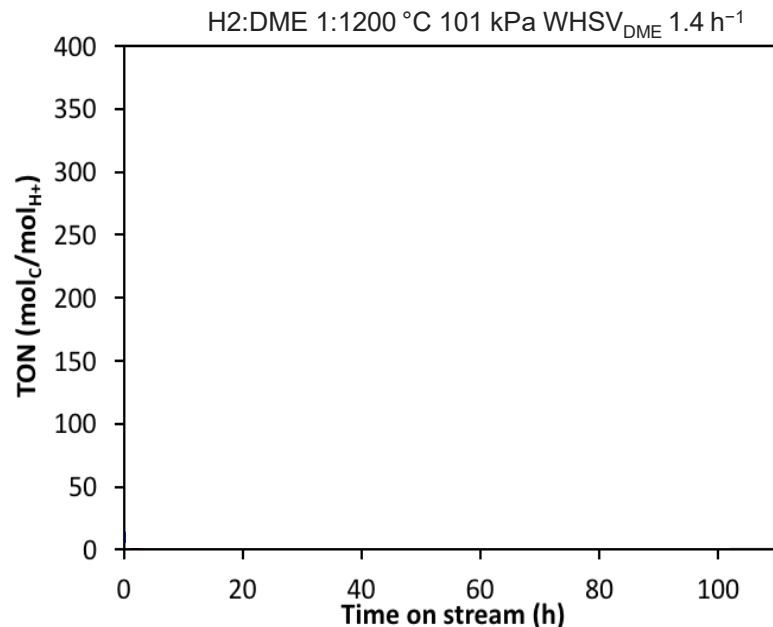
Beta zeolite (BEA)



Cu modified (Cu/BEA)

J. A. Schaidle et al. *ACS Catal.*, **2015**, 5, 1794

- Maintains high selectivity to branched C₄ and C₇ hydrocarbons
- High yield for HOG range products

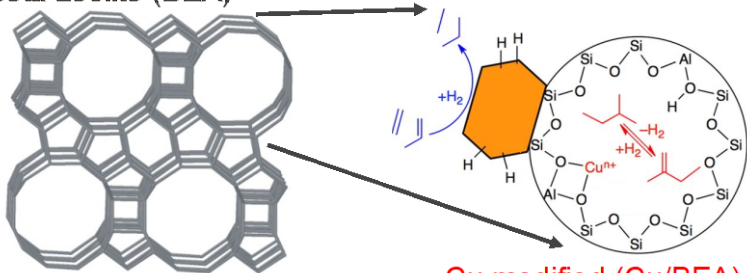


D. A. Ruddy et al. *Nat. Catal.*, **2019**, 2, 632

US Pat # 9,803,142 B1
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Cu-modified beta zeolites for DME conversion

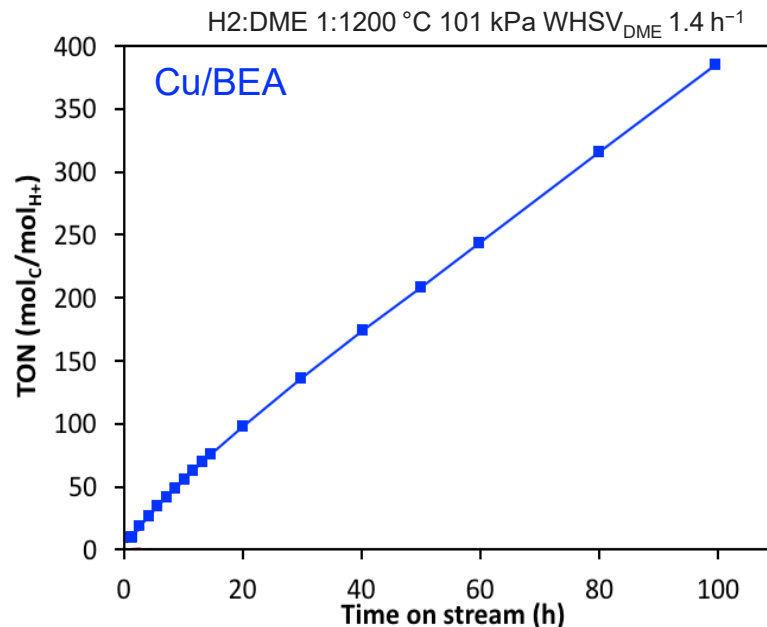
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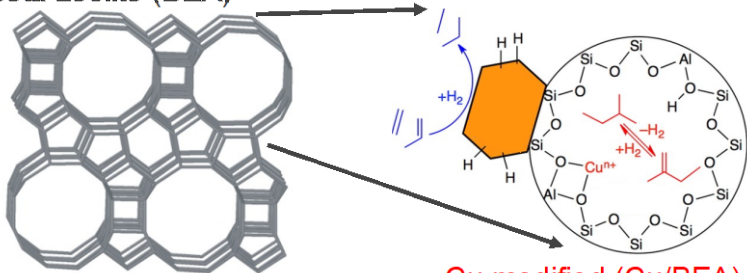
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D. A. Ruddy et al. *Nat. Catal.*, **2019**, 2, 632

Cu-modified beta zeolites for DME conversion

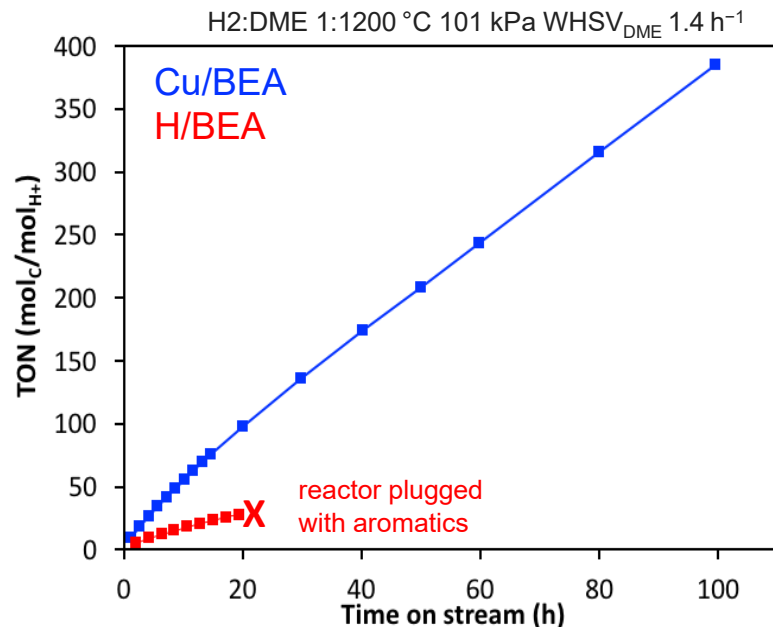
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J. A. Schaidle et al. *ACS Catal.*, **2015**, 5, 1794

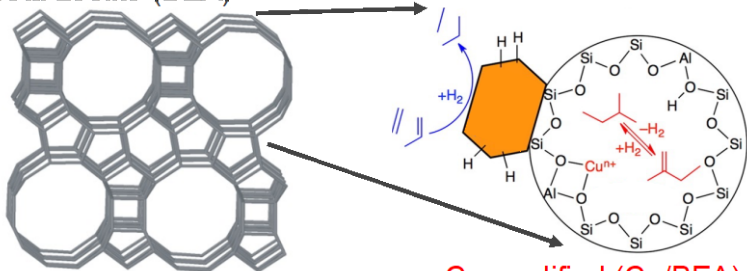
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D. A. Ruddy et al. *Nat. Catal.*, **2019**, 2, 632

Cu-modified beta zeolites for DME conversion

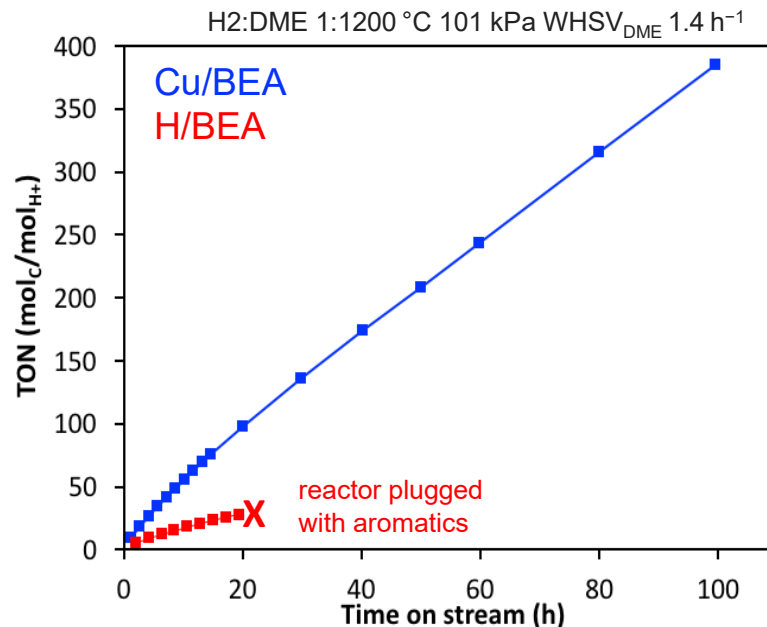
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J. A. Schaidle et al. *ACS Catal.*, **2015**, 5, 1794

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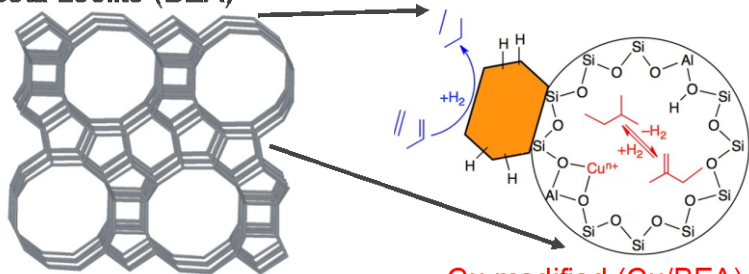
D. A. Ruddy et al. *Nat. Catal.*, **2019**, 2, 632

Cu/BEA had improved hydrocarbon yields over H/BEA while minimizing aromatic formation and increased catalyst lifetime

US Pat # 9,803,142 B1
US Pat App 62/482,315
US Pat App 62/515,087

Utilizing zeolites in a single reactor for syngas conversion

Beta zeolite (BEA)

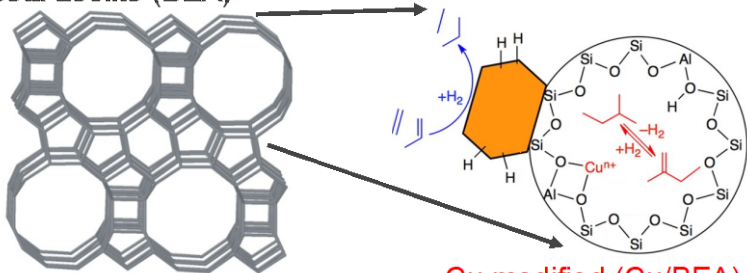


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J. A. Schaidle et al. *ACS Catal.*, 2015, 5, 1794

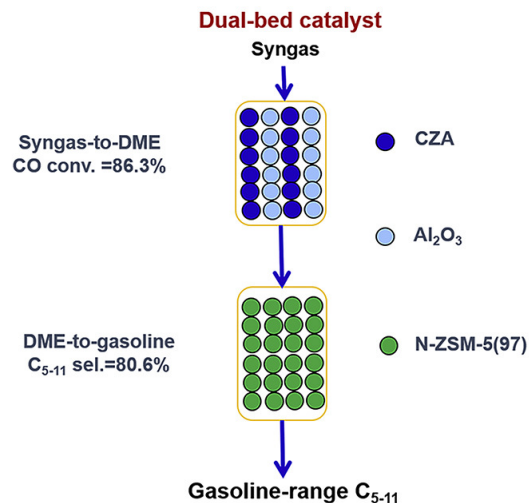
Utilizing zeolites in a single reactor for syngas conversion

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J. A. Schaidle et al. *ACS Catal.*, 2015, 5, 1794



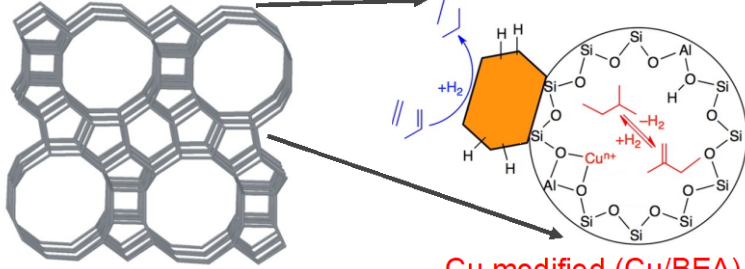
- Nano-sized ZSM-5 downstream of commercial methanol and DME synthesis catalysts
- High selectivity to gasoline range hydrocarbon products

Y. Ni et al. *Chem Catal.*, 2021, 1, 383

A dual-bed catalyst achieved an 80.6% gasoline selectivity at 86.3% CO conversion

Conversion of syngas in a single reactor system

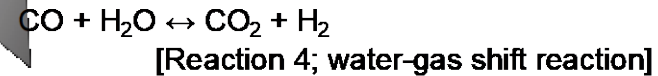
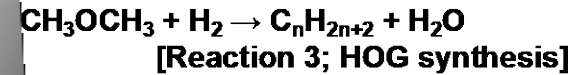
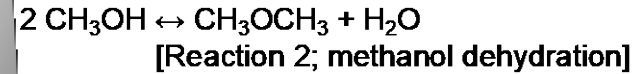
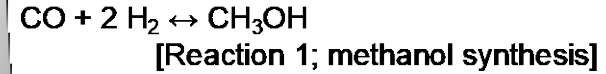
Beta zeolite (BEA)



Cu modified (Cu/BEA)

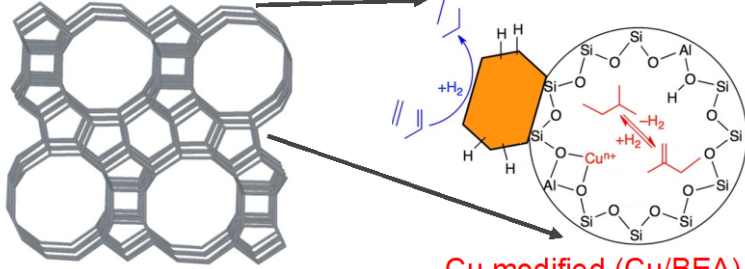
J. A. Schaidle et al. *ACS Catal.*, 2015, 5, 1794

CO₂-rich Syngas



Conversion of syngas in a single reactor system

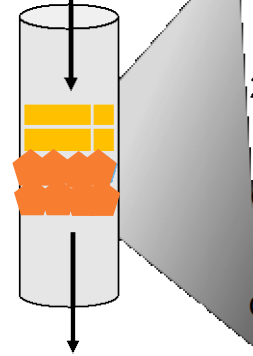
Beta zeolite (BEA)



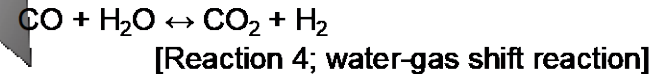
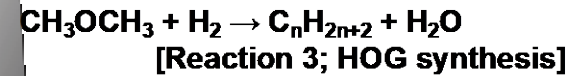
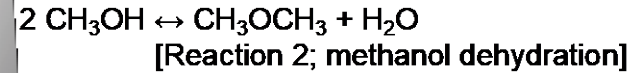
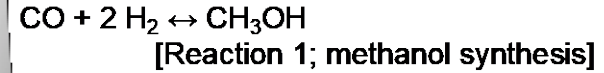
Cu modified (Cu/BEA)

Systematic configuration and composition changes to improve performance

CO₂-rich Syngas

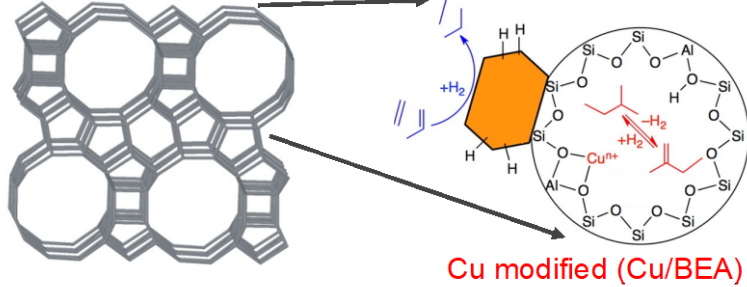


Hydrocarbons



Conversion of syngas in a single reactor system

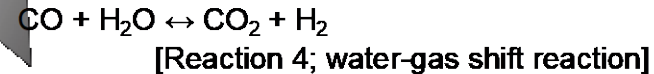
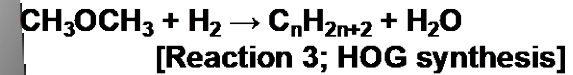
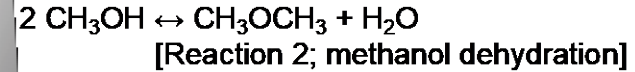
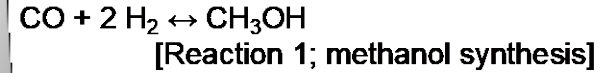
Beta zeolite (BEA)



CO₂-rich Syngas



Hydrocarbons

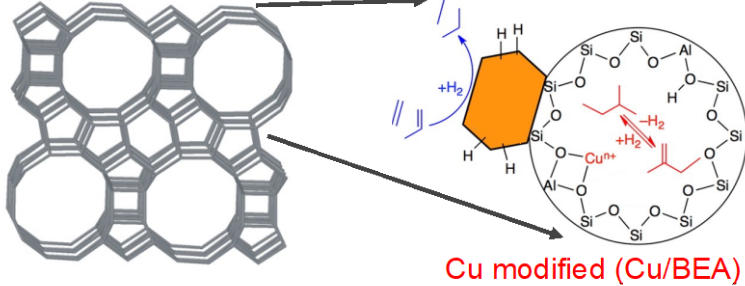


Systematic configuration and composition changes to improve performance

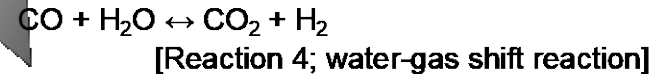
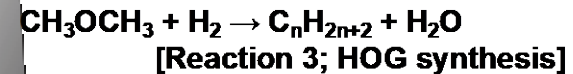
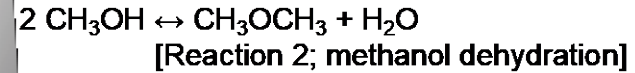
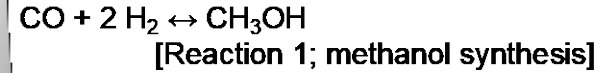
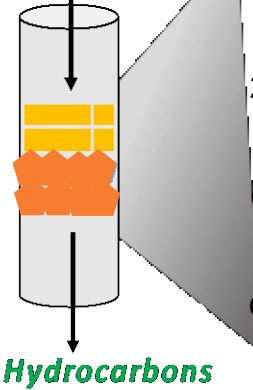
Compare direct syngas to hydrocarbon performance with DME homologation previously measured on Cu/BEA

Conversion of syngas in a single reactor system

Beta zeolite (BEA)



CO₂-rich Syngas

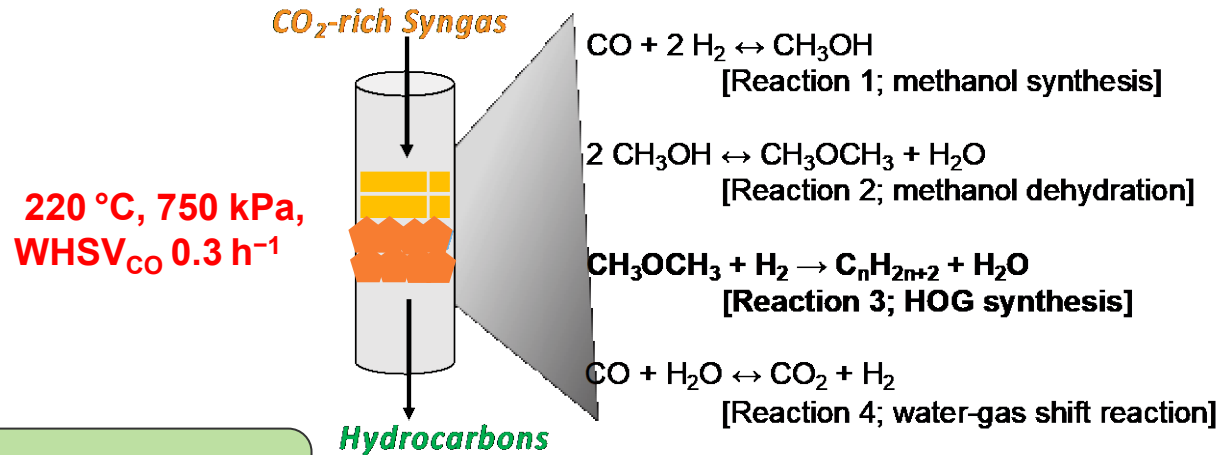


Systematic configuration and composition changes to improve performance

Compare direct syngas to hydrocarbon performance with DME homologation previously measured on Cu/BEA

Study CO₂ rich syngas simulated feed to understand its impact on hydrocarbon yields and overall performance

Conversion of syngas in a single reactor system

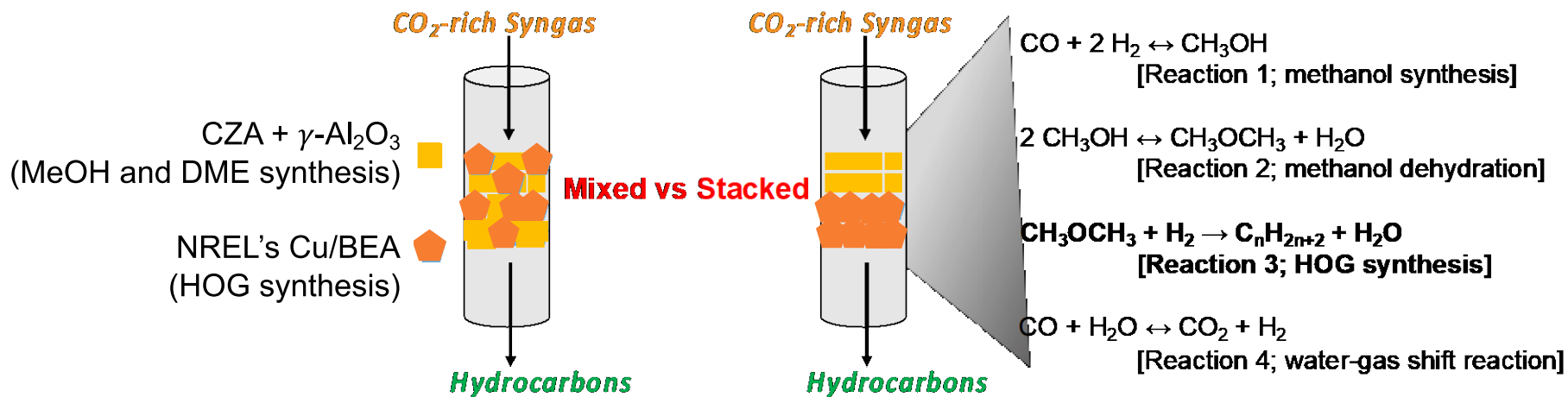


Systematic configuration and composition changes to improve performance

Utilize commercially available catalysts for the syngas to DME reactions

- “CZA” Cu/ZnO/Al₂O₃, Megamax 800, Clariant
- “A” γ-Al₂O₃, NorPro® SA6173, St. Gobain-Norpro

Conversion of syngas in a single reactor system



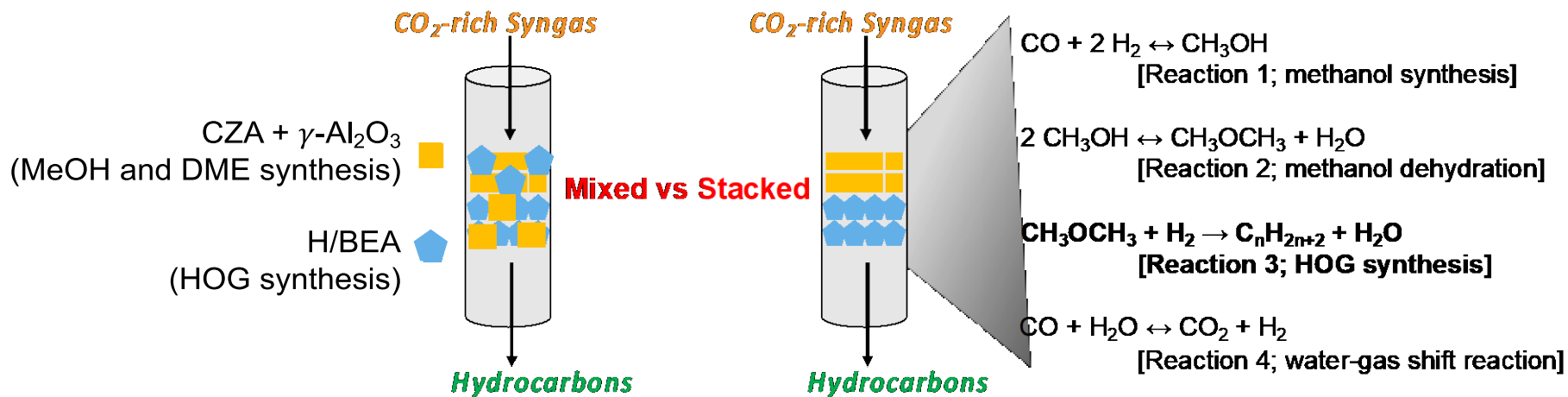
“mixed-bed” catalyst#1+catalysts #2

“stacked-bed” top-catalyst|bottom-catalyst

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Conversion of syngas in a single reactor system



“mixed-bed” catalyst#1+catalysts #2

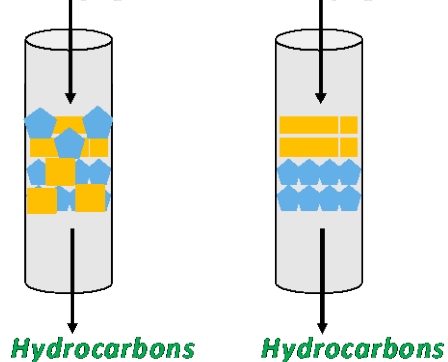
“stacked-bed” top-catalyst|bottom-catalyst

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Reactor configuration influence on syngas conversion

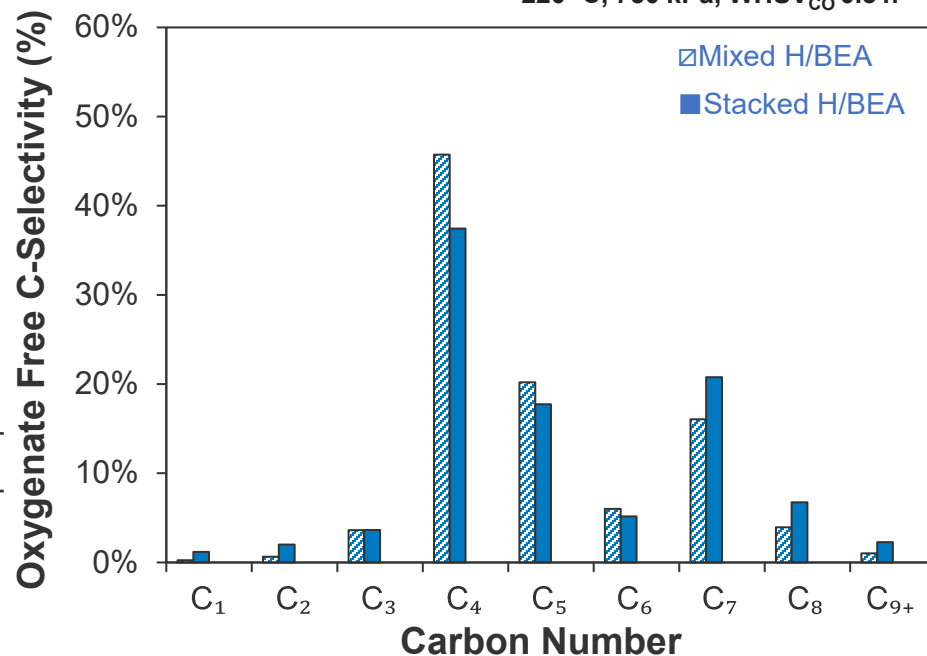
CO₂-rich Syngas *CO₂-rich Syngas*



H/Beta

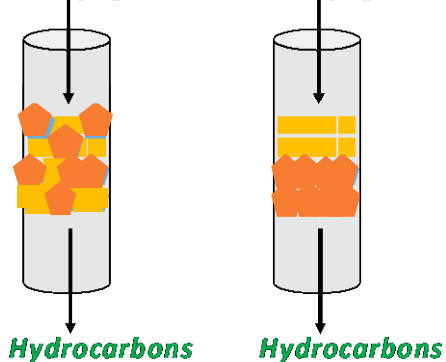
	Stacked	Mixed
CO Conversion %	64.1	40.9
C₄₊ Product Yield %	6.4	28.9
CO₂ Selectivity %	36.8	49.8

220 °C, 750 kPa, WHSV_{CO} 0.3 h⁻¹

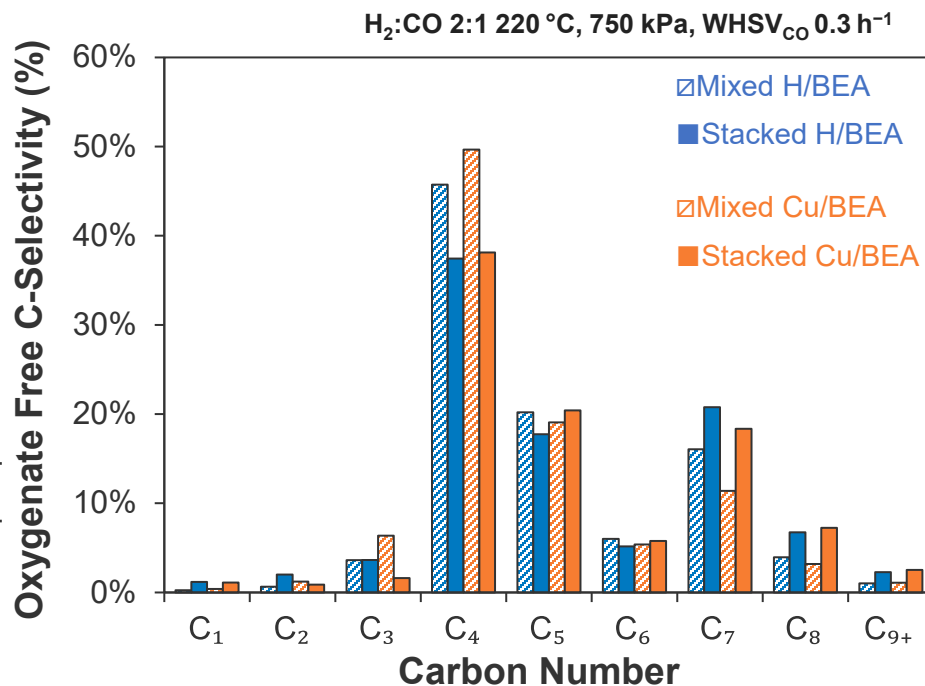


Reactor configuration influence on syngas conversion

CO₂-rich Syngas *CO₂-rich Syngas*

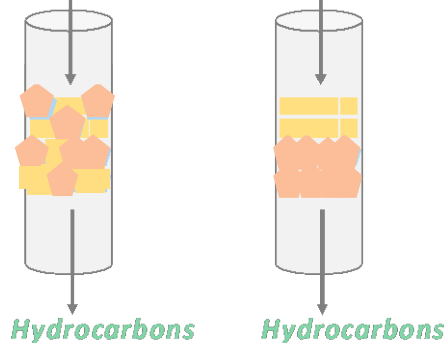


	H/Beta		Cu/Beta	
	Stacked	Mixed	Stacked	Mixed
CO Conversion %	64.1	40.9	46.5	21.2
C₄₊ Product Yield %	6.4	28.9	23.6	24.5
CO₂ Selectivity %	36.8	49.8	38.1	52.6

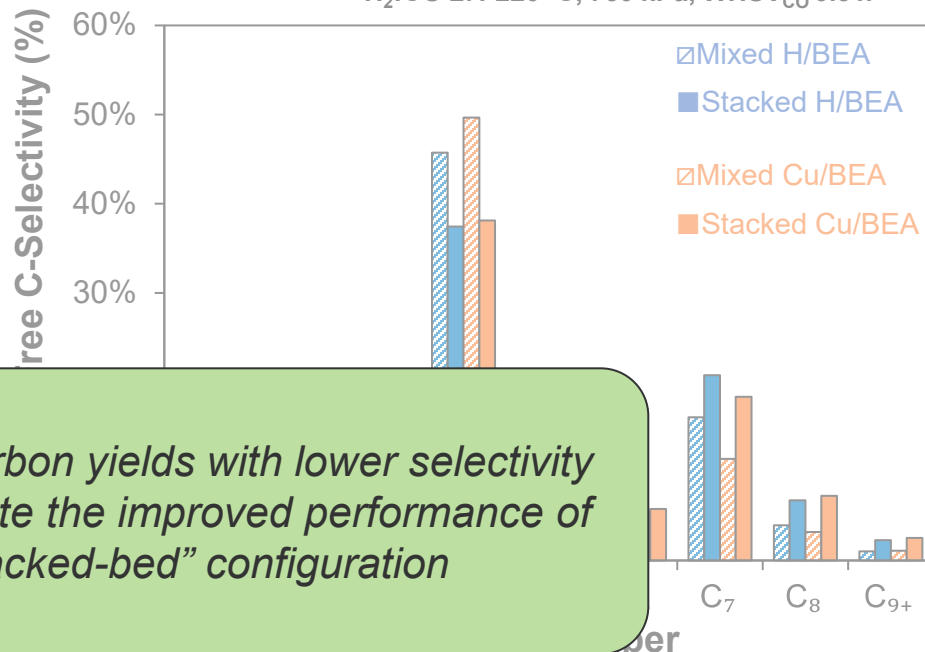


Reactor configuration influence on syngas conversion

CO₂-rich Syngas CO₂-rich Syngas



H₂:CO 2:1 220 °C, 750 kPa, WHSV_{CO} 0.3 h⁻¹



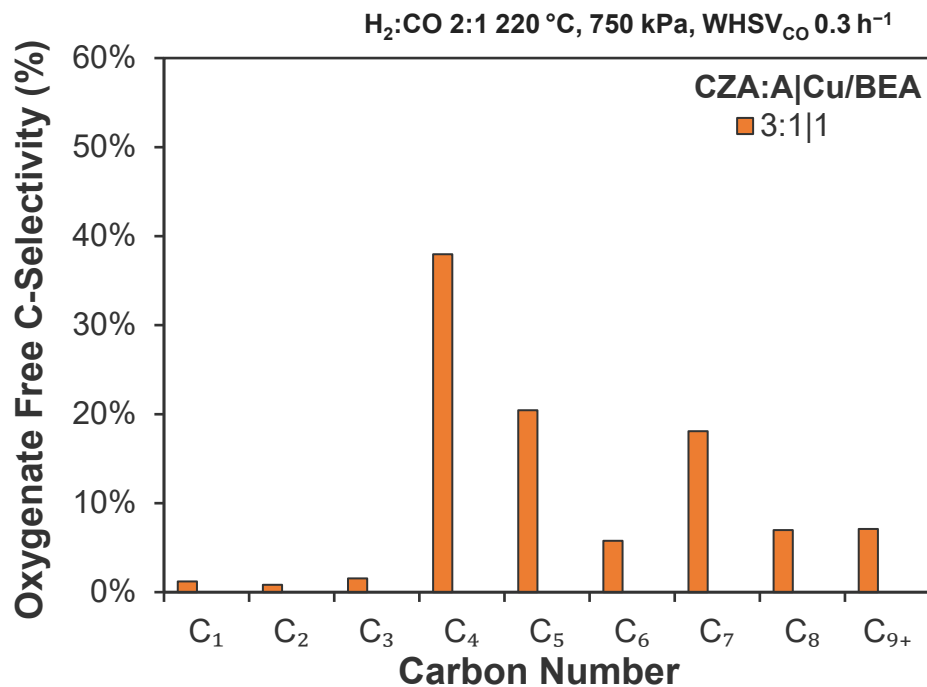
Balancing high hydrocarbon yields with lower selectivity towards CO₂ demonstrate the improved performance of Cu/BEA in a “stacked-bed” configuration

	Stacked H/BEA	Stacked Cu/BEA
CO Conversion %	64	64
C ₄₊ Product Yield %	6.4	6.4
CO ₂ Selectivity %	36.8	52.6

Consideration of Cu/BEA content on hydrocarbon formation

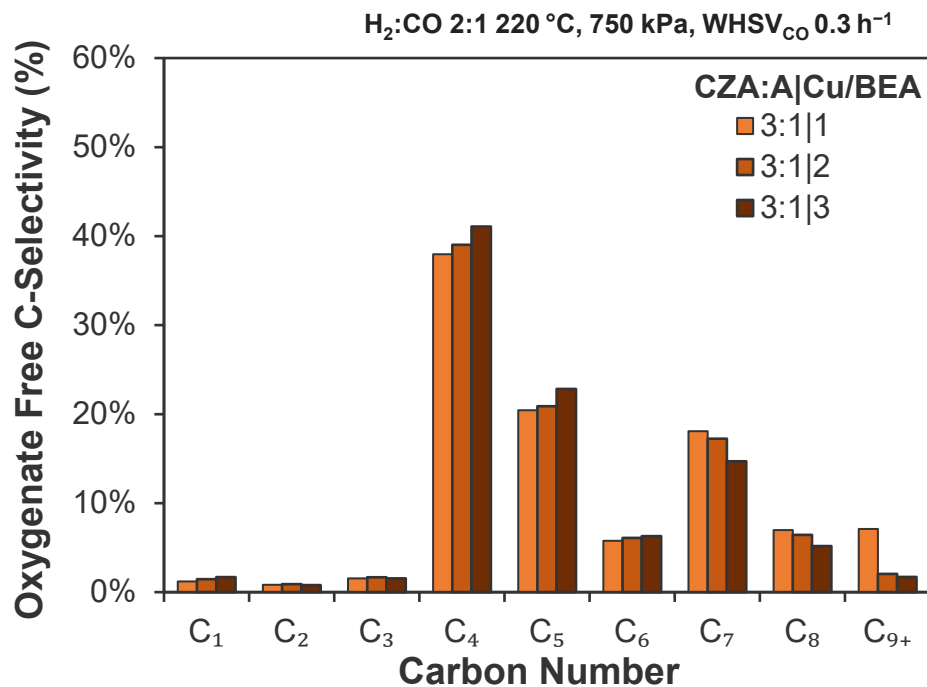
	CZA: γ -Al ₂ O ₃ Cu/BEA
	3:1 1
CO Conversion %	46.5
C ₄₊ Product Yield %	23.6
CO ₂ Selectivity %	38.1

Measured at same time on stream (~12h)



Consideration of Cu/BEA content on hydrocarbon formation

	CZA: γ -Al ₂ O ₃ Cu/BEA		
	3:1 1	3:1 2	3:1 3
CO Conversion %	46.5	56.4	55.8
C ₄₊ Product Yield %	23.6	19.3	24.7
CO ₂ Selectivity %	38.1	39.9	40.0

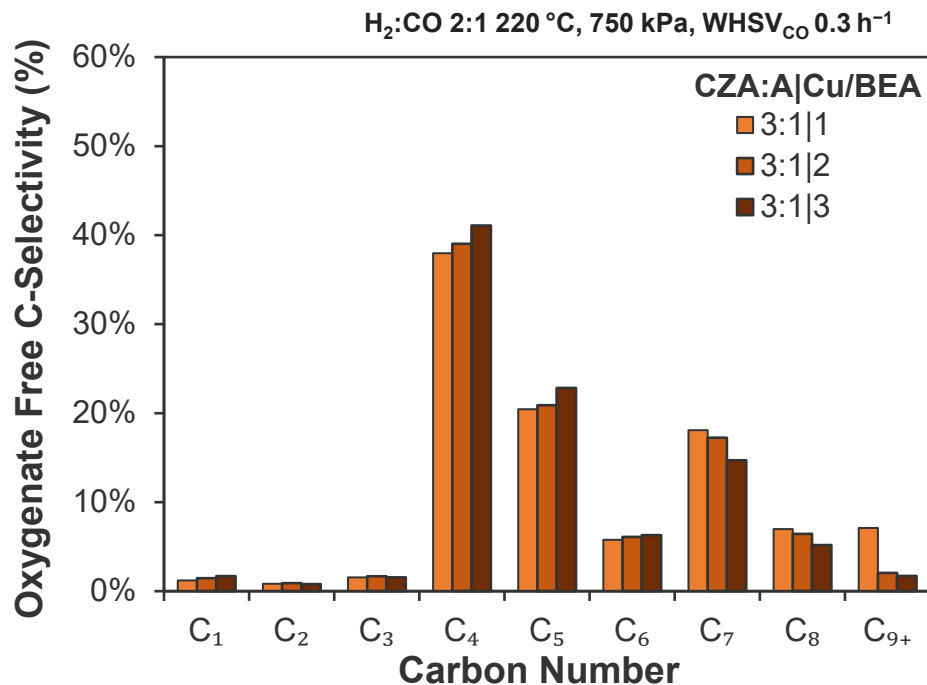


Measured at same time on stream (~12h)

Consideration of Cu/BEA content on hydrocarbon formation

	CZA:γ-Al ₂ O ₃ Cu/BEA		
	3:1 1	3:1 2	3:1 3
CO Conversion %	46.5	56.4	55.8
C ₄₊ Product Yield %	23.6	19.3	24.7
CO ₂ Selectivity %	38.1	39.9	40.0
DME Product Yield %	20.7	10.9	3.5
Total HC Productivity (g g _{Cu/BEA} ⁻¹ h ⁻¹)	0.105	0.063	0.053
Turnover Number (mol _{HC} mol _{H⁺} ⁻¹)	86	64	44

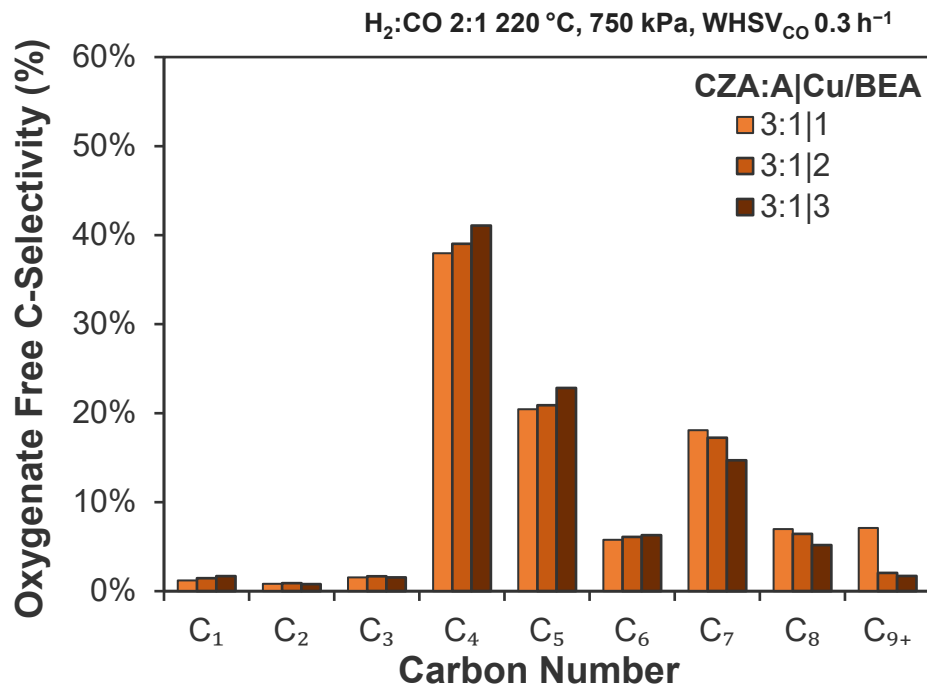
Measured at same time on stream (~12h)



Consideration of Cu/BEA content on hydrocarbon formation

	CZA: γ -Al ₂ O ₃ Cu/BEA		
	3:1 1	3:1 2	3:1 3
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Measured at same time on stream (~12h)

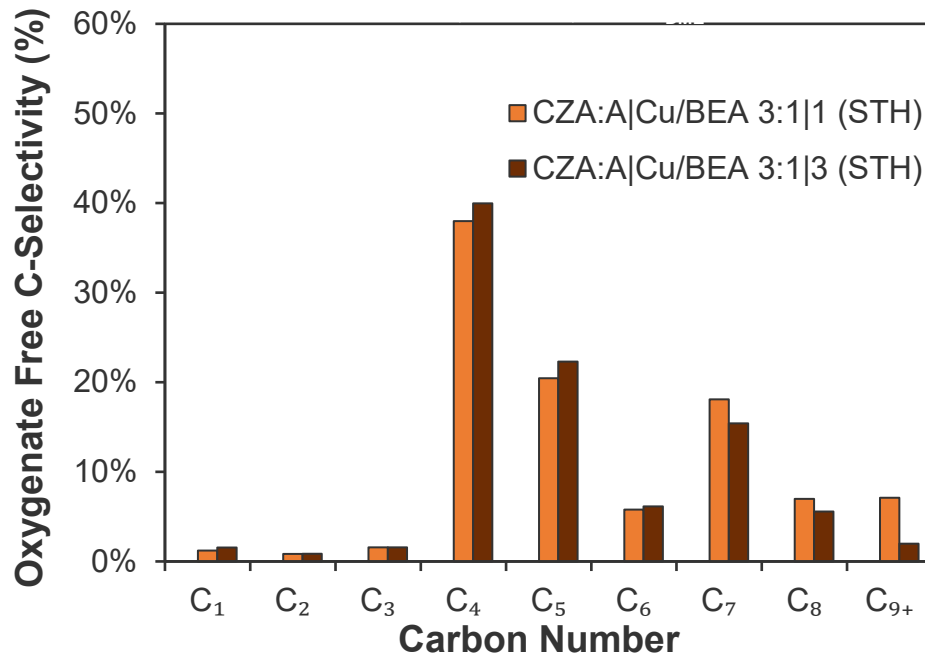


Improved hydrocarbon production achieved when considering the efficient utilization of in-situ formed of DME

Syngas conversion compared to DME conversion

Syngas: H₂:CO 2:1 220 °C, 750 kPa, WHSV_{CO} 0.3 h⁻¹

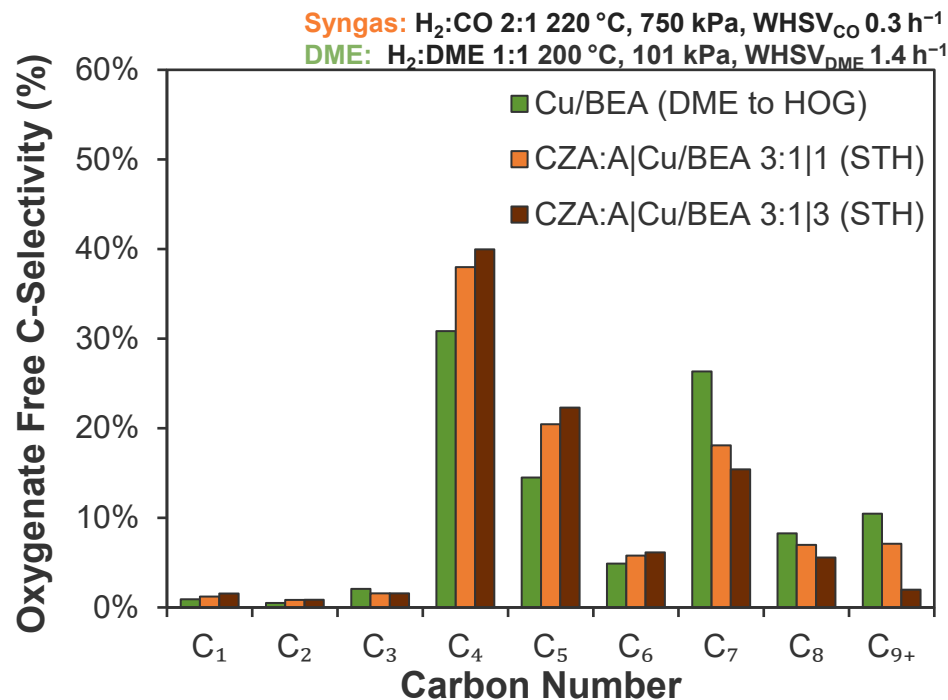
	CZA+A Cu/Beta (STH)	
	3:1 1	3:1 3
CO		
Conversion %	46.5	55.0
C₄₊ C-Yield %	23.6	23.8
HC Productivity (g g _{Cu/Beta} ⁻¹ h ⁻¹)	0.105	0.051
Turnover Number (mol _{HC} mol _{H+} ⁻¹)	86	77



Syngas conversion compared to DME conversion

	Cu/Beta (DME)	CZA+A Cu/Beta (STH)	
		3:1 1	3:1 3
DME or CO Conversion %	9.3	46.5	55.0
C₄₊ C-Yield %	6.7	23.6	23.8
HC Productivity (g g _{Cu/Beta} ⁻¹ h ⁻¹)	0.050	0.105	0.051
Turnover Number (mol _{HC} mol _{H+} ⁻¹)	83	86	77

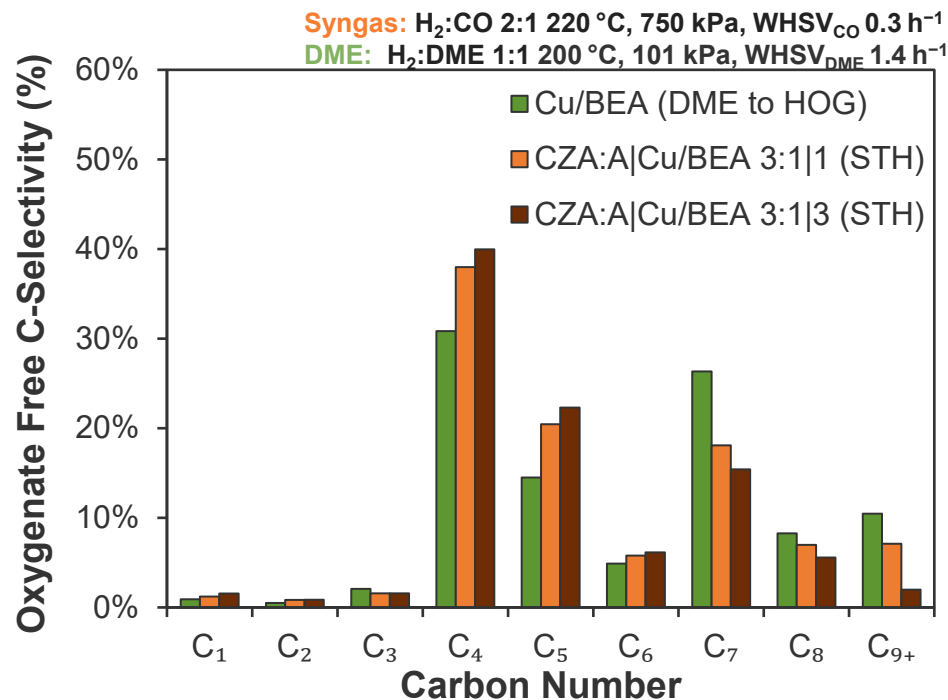
Measured at comparable TON



Syngas conversion compared to DME conversion

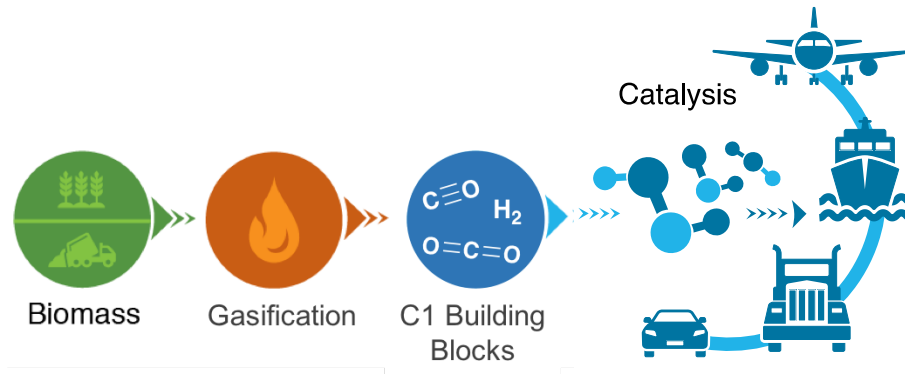
	Cu/Beta (DME)	CZA+A Cu/Beta (STH)	
		3:1 1	3:1 3
DME or CO Conversion %	9.3	46.5	55.0
C₄₊ C-Yield %	6.7	23.6	23.8
HC Productivity (g g _{Cu/Beta} ⁻¹ h ⁻¹)	0.050	0.105	0.051
Turnover Number (mol _{HC} mol _{H+} ⁻¹)	83	86	77

Measured at comparable TON

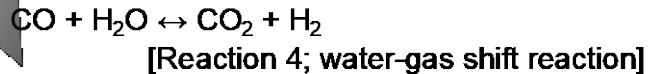
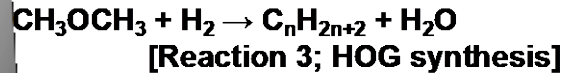
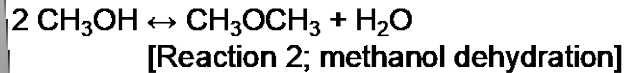
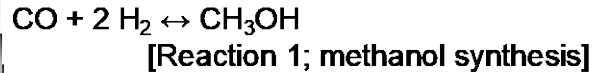


Direct syngas to hydrocarbon performance improved over DME homologation while maintaining high selectivities to HOG range products

Introduction of CO₂ into syngas

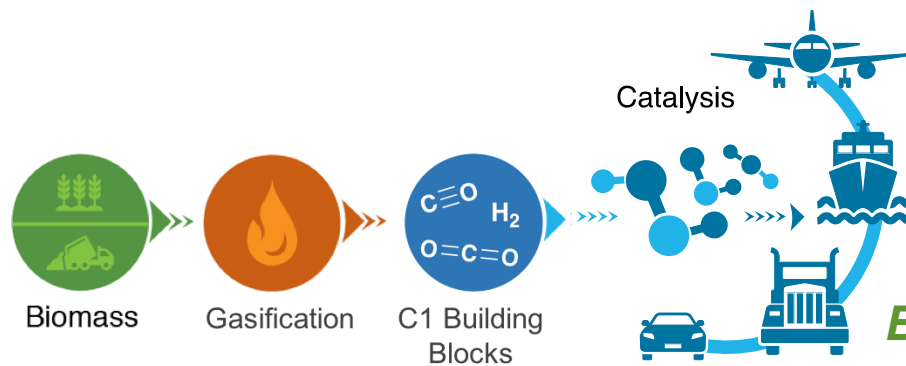


CO₂-rich Syngas

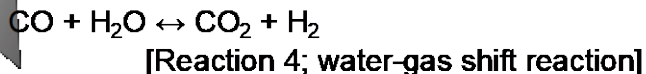
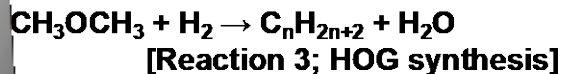
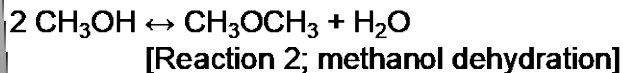
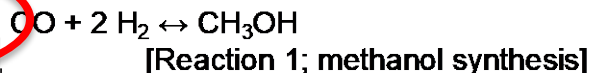
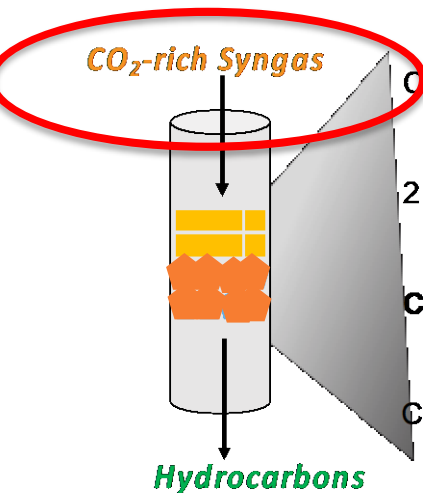


Hydrocarbons

Introduction of CO₂ into syngas



Biomass derived sources of syngas



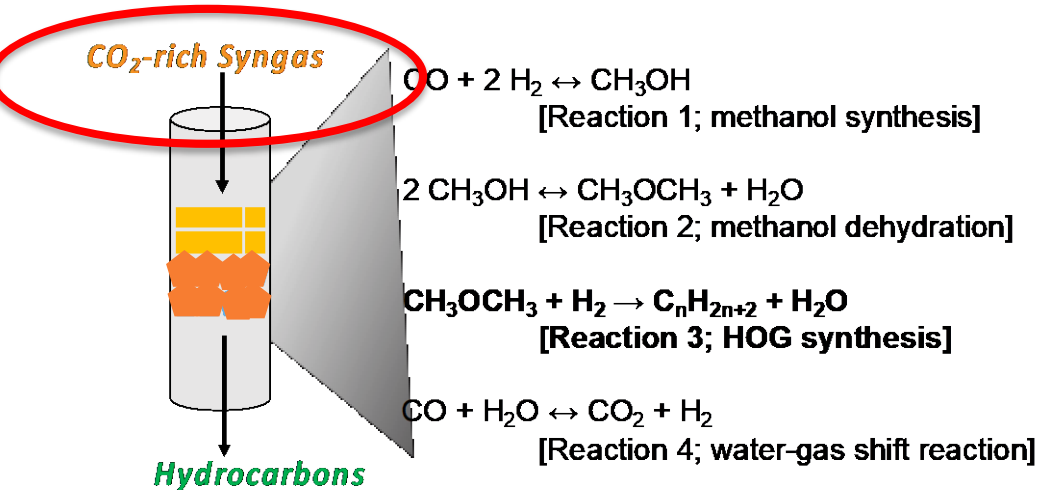
Feedstock	H₂ (vol %)	CO (vol %)	CO₂ (vol %)
Miscanthus	57.8	21.6	20.4
Switchgrass	57.9	22.2	19.7
Forest Residue	56.7	20.0	22.9

D.P. Dupuis et al. *Appl. Energy*, 2019, 241, 25

Introduction of CO₂ into syngas

Assess syngas conversion in the single reactor with co-fed CO₂

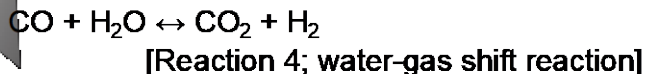
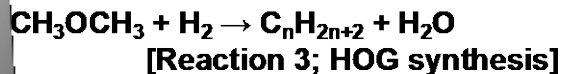
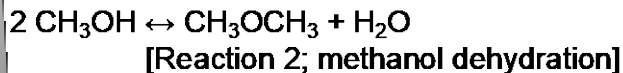
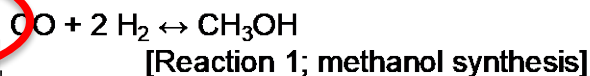
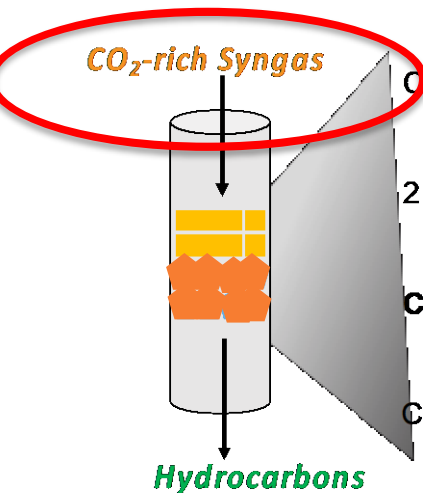
- Molar ratios of H₂:CO:CO₂
2:1:0.8
- CZA+A|Cu/BEA bed
composition 3:1|3



Introduction of CO₂ into syngas

Assess syngas conversion in the single reactor with co-fed CO₂

- Molar ratios of H₂:CO:CO₂
2:1:0.8
- CZA+A|Cu/BEA bed
composition 3:1|3



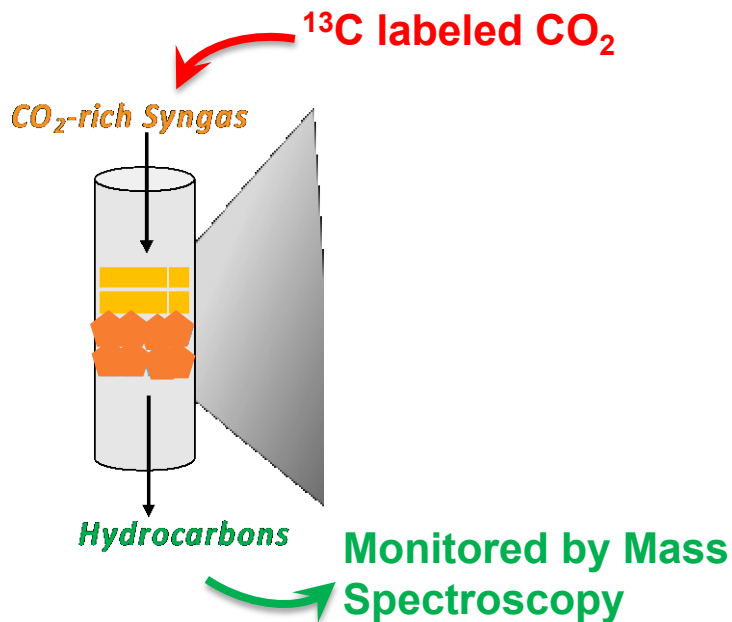
CZA+A|Cu/BEA 3:1|3

	Without CO ₂	With CO ₂
CO+CO ₂ Conversion %	77.3	27.0
C ₄₊ Product Yield %	40.8	23.8
CO ₂ Selectivity %	38.3	28.4
Hydrocarbon Productivity (g g _{Cu/BEA} ⁻¹ h ⁻¹)	0.098	0.054

CO₂ incorporation into hydrocarbon products

Assess syngas conversion in the single reactor with co-fed CO₂

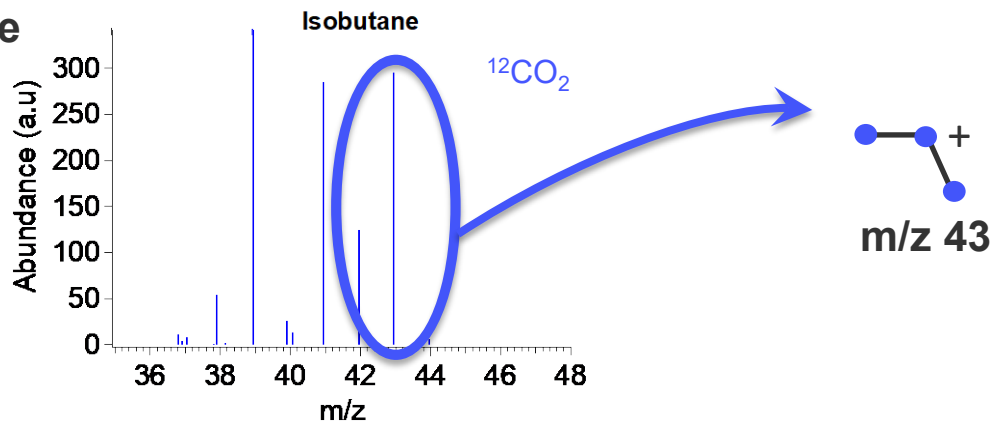
- Molar ratios of H₂:CO:CO₂
2:1:0.8
- CZA+AlCu/BEA bed
composition 3:1|3



CO₂ incorporation into hydrocarbon products

Assess syngas conversion in the single reactor with co-fed CO₂

- Molar ratios of H₂:CO:CO₂ 2:1:0.8
- CZA+A|Cu/BEA bed composition 3:1|3

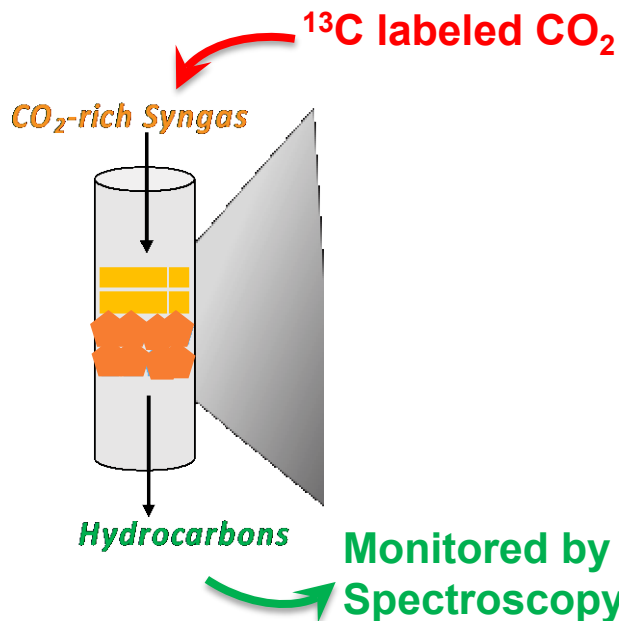
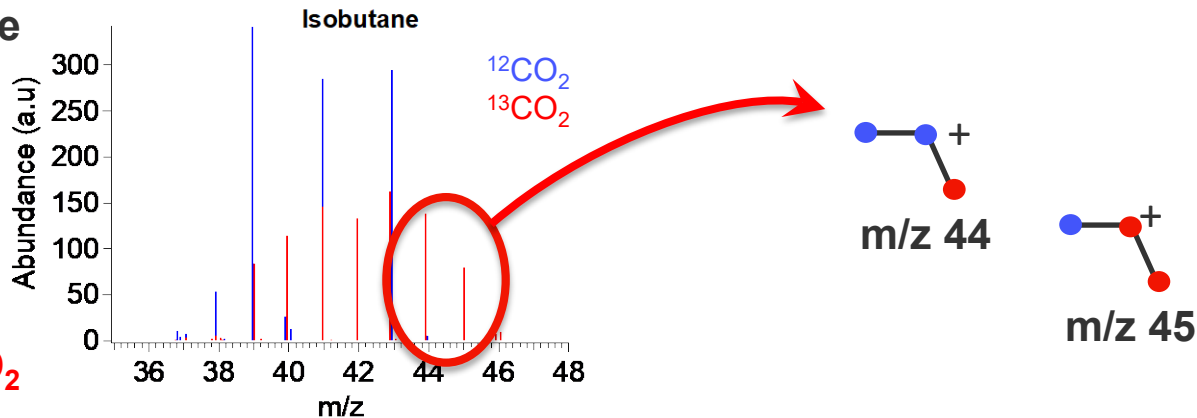


Hydrocarbons
Monitored by Mass Spectroscopy

CO₂ incorporation into hydrocarbon products

Assess syngas conversion in the single reactor with co-fed CO₂

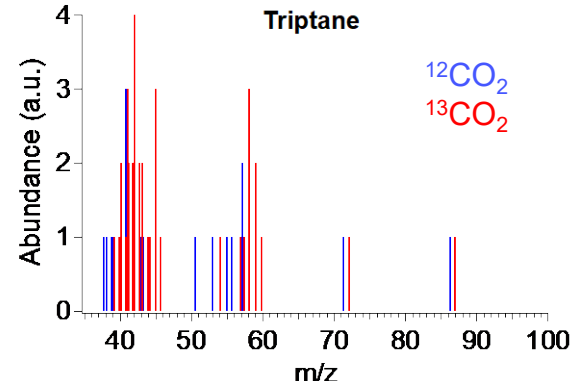
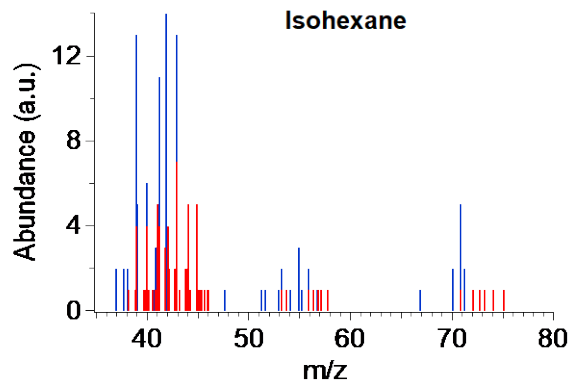
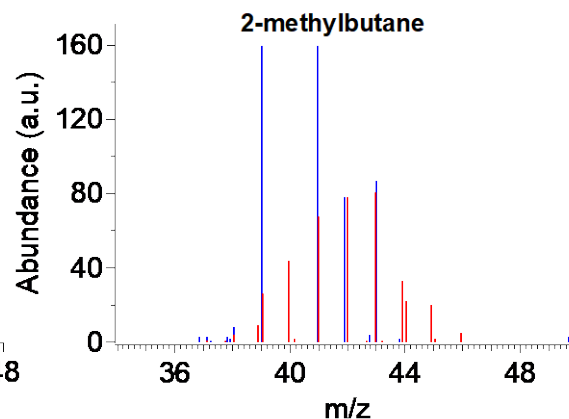
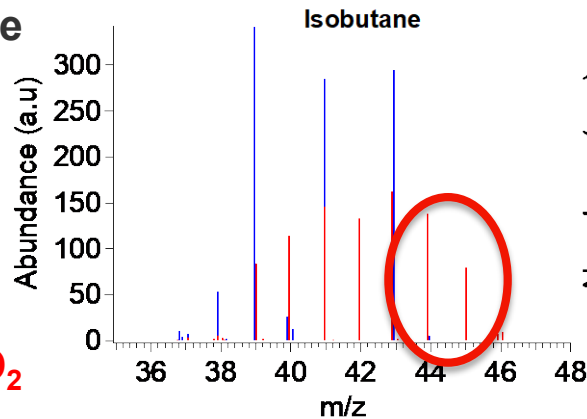
- Molar ratios of H₂:CO:CO₂ 2:1:0.8
- CZA+A|Cu/BEA bed composition 3:1|3



CO₂ incorporation into hydrocarbon products

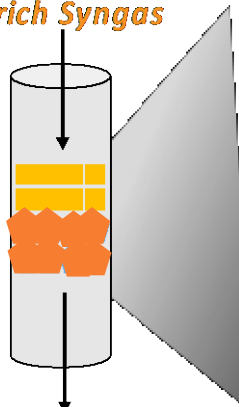
Assess syngas conversion in the single reactor with co-fed CO₂

- Molar ratios of H₂:CO:CO₂ 2:1:0.8
- CZA+A|Cu/BEA bed composition 3:1|3



¹³C labeled CO₂

CO₂-rich Syngas



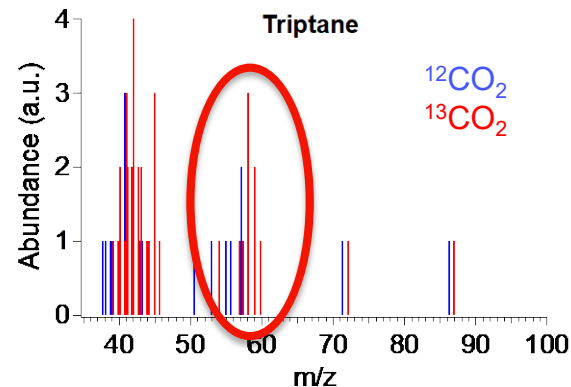
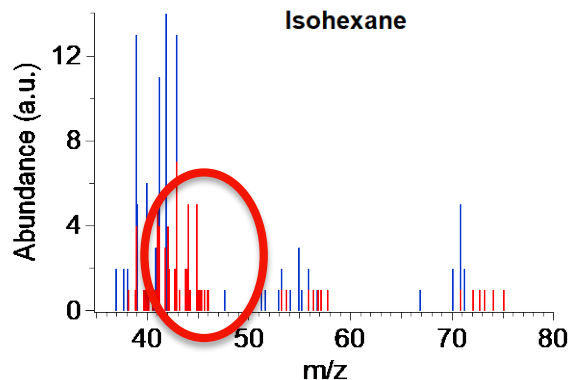
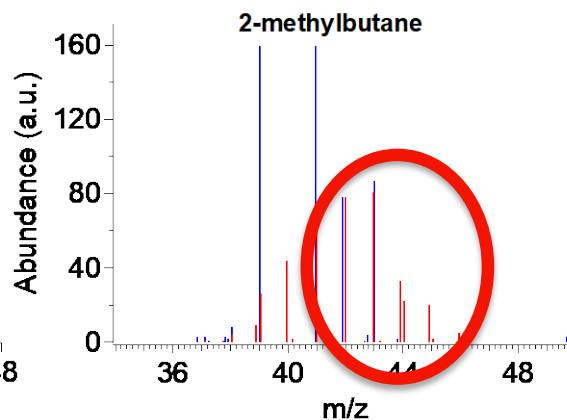
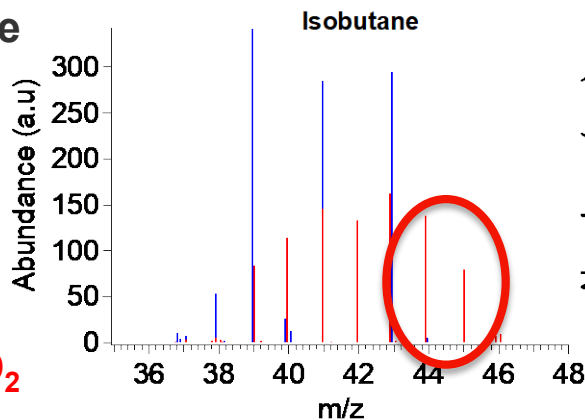
Hydrocarbons

Monitored by Mass Spectroscopy

CO₂ incorporation into hydrocarbon products

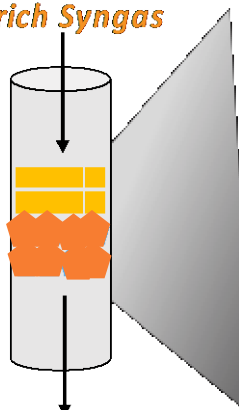
Assess syngas conversion in the single reactor with co-fed CO₂

- Molar ratios of H₂:CO:CO₂ 2:1:0.8
- CZA+A|Cu/BEA bed composition 3:1|3



¹³C labeled CO₂

CO₂-rich Syngas



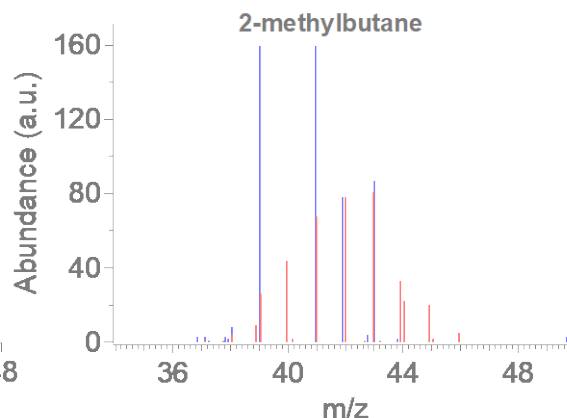
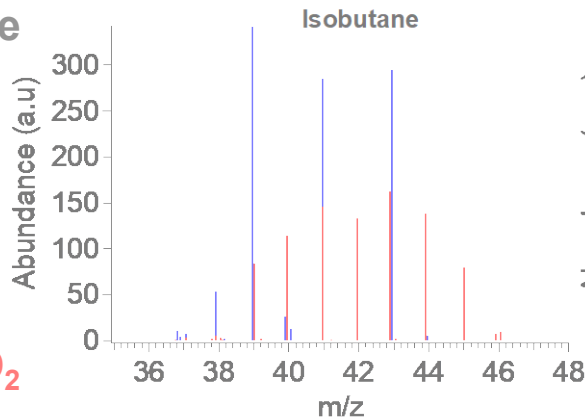
Hydrocarbons

Monitored by Mass Spectroscopy

CO₂ incorporation into hydrocarbon products

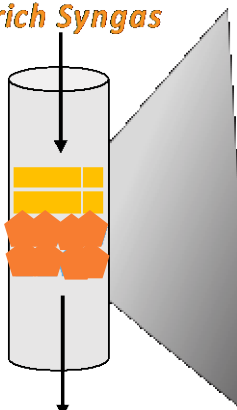
Assess syngas conversion in the single reactor with co-fed CO₂

- Molar ratios of H₂:CO:CO₂ 2:1:0.8
- CZA+A|Cu/BEA bed composition 3:1|3



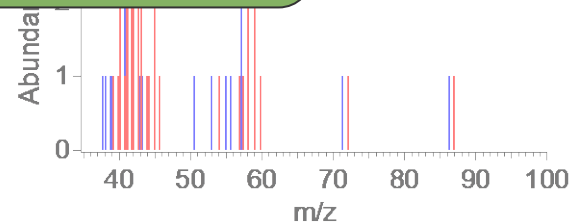
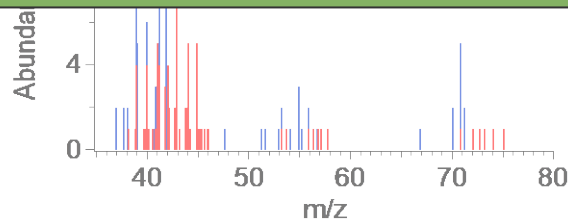
¹³C labeled CO₂

CO₂-rich Syngas



Hydrocarbons

Incorporation of CO₂ into hydrocarbon products, demonstrating viability of process to convert CO₂ rich syngas with improved carbon efficiency

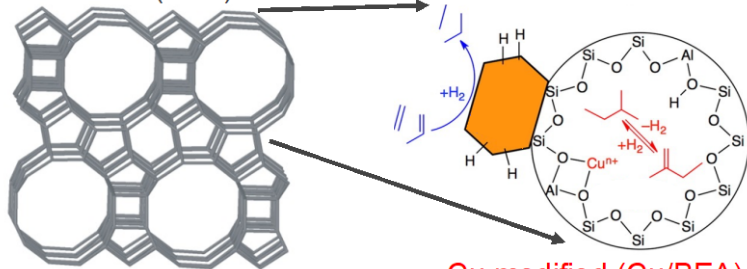


¹²CO₂
¹³CO₂

Monitored by Mass Spectroscopy

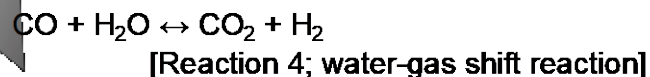
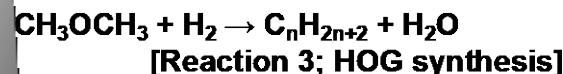
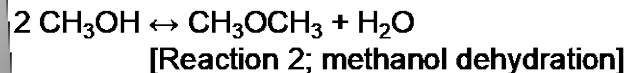
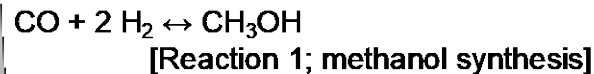
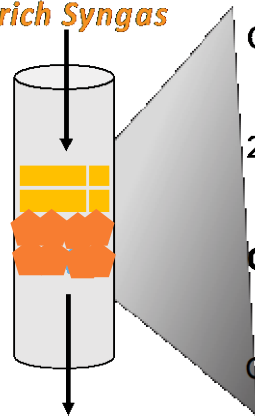
Conversion of CO₂-rich syngas in a single reactor system

Beta zeolite (BEA)



Cu modified (Cu/BEA)

CO₂-rich Syngas



Systematic configuration & composition changes demonstrate ability to improve performance

Direct syngas to hydrocarbon performance improved over DME homologation while maintaining high selectivities to HOG range products

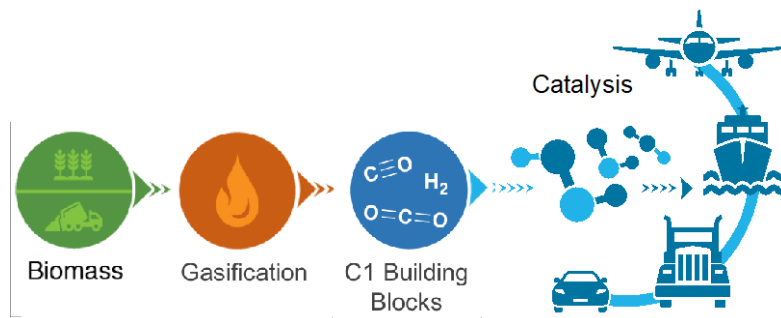
Incorporation of CO₂ into hydrocarbon products, demonstrating viability of process to convert CO₂ rich syngas with improved carbon efficiency

Bioenergy Technologies Office



Thank you, questions?

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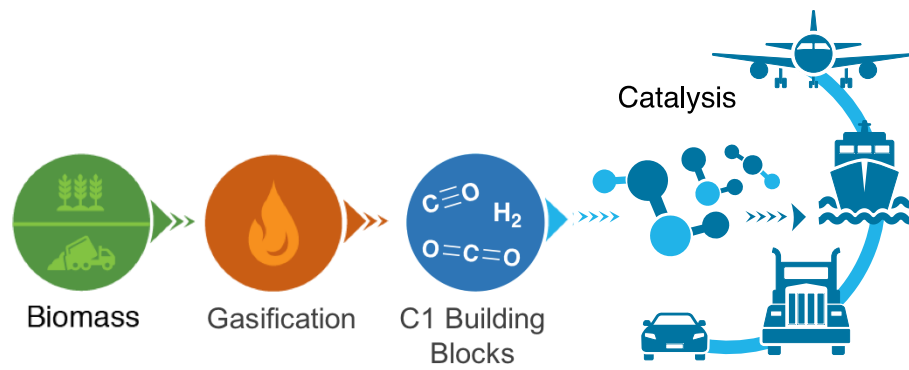
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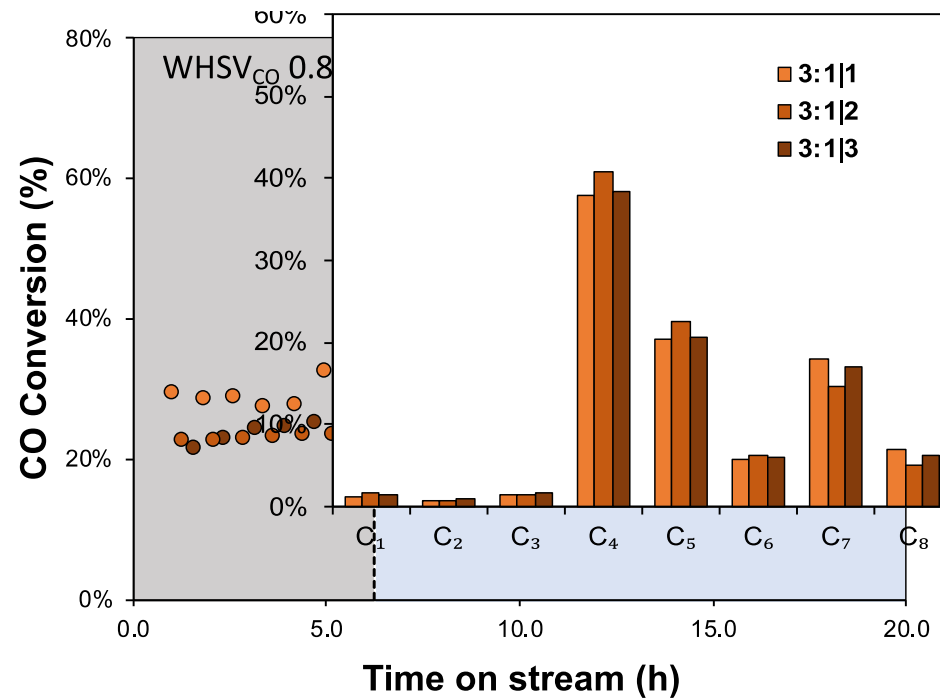
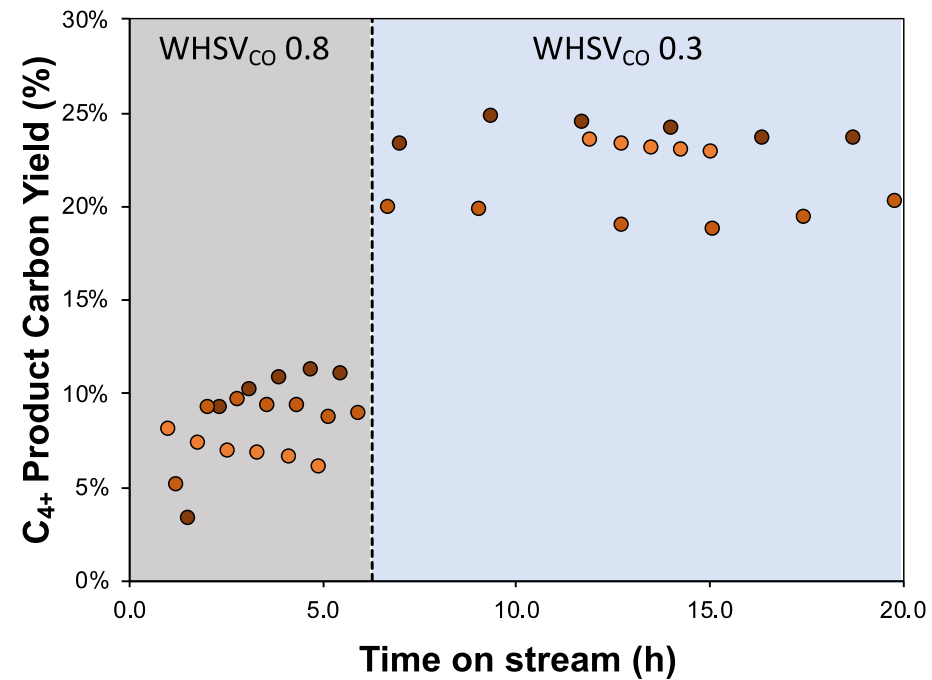
This work was authored by the National Renewable Energy Laboratory, operated by Alliance for Sustainable Energy, LLC, for the U.S. Department of Energy (DOE) under Contract No. DE-AC36-08GO28308. Funding provided by U.S. Department of Energy Office of Energy Efficiency and Renewable Energy Bioenergy Technologies Office. The views expressed in the article do not necessarily represent the views of the DOE or the U.S. Government. The U.S. Government retains and the publisher, by accepting the article for publication, acknowledges that the U.S. Government retains a nonexclusive, paid-up, irrevocable, worldwide license to publish or reproduce the published form of this work, or allow others to do so, for U.S. Government purposes.

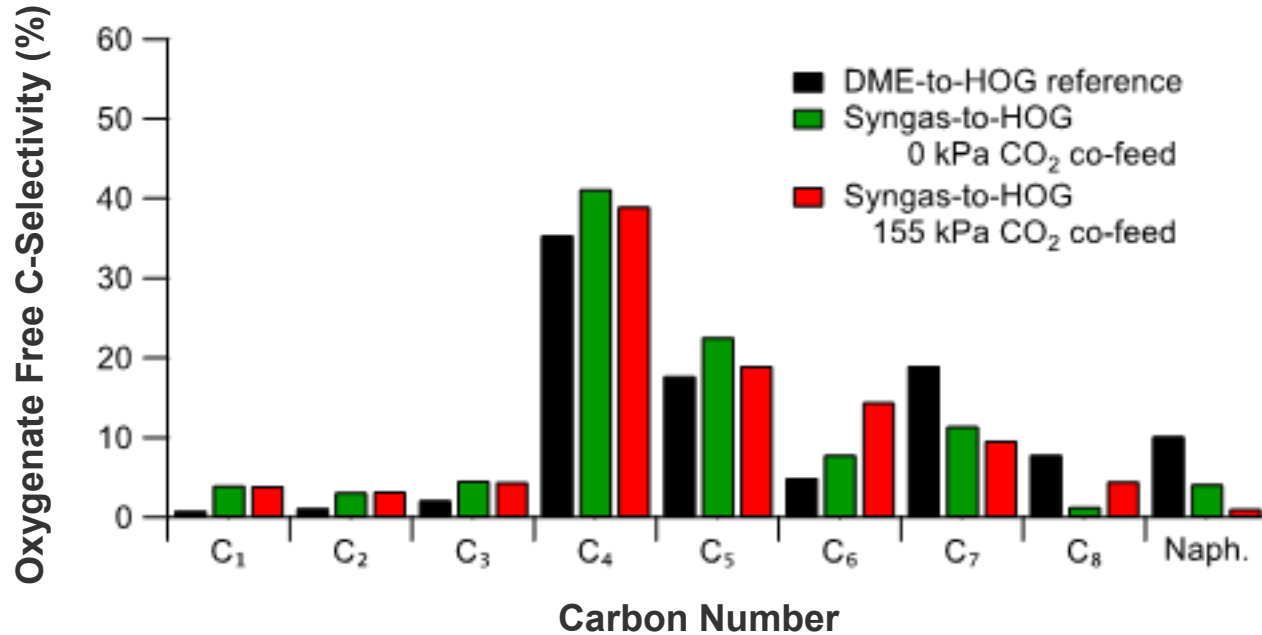
Process Intensification for Direct Conversion of Biomass-Based Syngas to High Octane Gasoline

Claire Nimlos, Connor Nash, Anh To, Dan Dupuis, Daniel Ruddy*

Catalytic Carbon Transformation & Scale-up Center, National Renewable Energy Laboratory







*Additional
hydrocarbon
species monitored*

