

Pathways for Negative-Emissions Hydrogen: Opportunities and R&D Needs

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Clear and Present Danger

Modelled mitigation pathways that limit warming to 1.5°C, and 2°C, involve deep, rapid and sustained emissions reductions.

b. Net global CO2 emissions





Past emissions (2000–2015)
Model range for 2015 emissions
Past GHG emissions and uncertainty for 2015 and 2019 (dot indicates the median)
Percentile of 2100 emission level:
95th
75th
Median
25th

5th

Requirements to stay below 2.0 deg. C warming:

- 27% reduction (-15 GTCO_{2-eq}/y) relative to 2019 levels by 2030
- 63% reduction (-35 GTCO_{2-eq}/y) relative to 2019 levels by 2050

Requirements to stay below 1.5 deg. C warming:

- 43% reduction (-24 GTCO_{2-eq}/y) relative to 2019 levels by 2030
- 84% reduction (-46 GTCO_{2-eq}/y) relative to 2019 levels by 2050

Immediate and long-term decarbonization solutions needed

Trends in Global H₂ Consumption

In 2020, global demand for H₂ reached an estimated 820 billion nm³ (~74 million metric tonnes)



Current H₂ demand dominated across ammonia synthesis, hydrocarbon refining, and methanol synthesis (~89%)

Emissions Reduction Potential of H₂

Global H₂ Production Statistics:

- ~82% produced via reforming of methane / hydrocarbons
- ~15% produced via coal gasification
- ~3% from electrolysis/other

H₂ Emission Statistics:

- 9.4 kg $CO_2 e/kg H_2$ (SMR)
- 16.8 kg CO₂e/kg H₂ (Coal)
- ~0 kg CO₂e/kg H₂ (electrolysis)

\rightarrow Avg = 10.3 kg CO₂e/kg H₂



Impact of H₂ Production on CO₂ Emissions



CO₂ Emissions (MMT/y)

Potential to avoid > 0.7 GT CO_2e/y globally + removal of ~ 1.0 GT CO_2e/y across <u>current markets</u> representing >10% of IPCC 2 deg. C 2030 target^{NREL | 4}

Carbon Emitters to Net Carbon Sinks

Product and technology pathways	Baseline carbon Intensity	Carbon Intensity using green H ₂	Carbon Intensity using carbon- negative Bio-H ₂
BF-BOF steel production (t-CO_/t-HM) (H ₂ replacing pulverized coal for hot air blast)	2.20	1.78	1.13 to 1.65
DRI-EAF gas steel production (t-CO ₂ /t-HM) (H ₂ replacing all DRI gas consumption)	1.35	0.29 (renewable electricity) 0.80 (grid electricity)	-0.61 to 0.39
NG SMR-based ammonia production (t-CO ₂ /t-NH ₃) (1:1 replacement of original H ₂ consumption)	1.97	0.27	-4.04 to -0.6
NG SMR-based methanol production (t-CO ₂ /t-MeOH) (1:1 replacement of original H ₂ consumption)	2.57 (including feedstock emission) 0.86 (energy emission)	0.77 (no feedstock emission when using external captured CO_2 as feedstock)	-3.89 to $-0.17(no feedstockemission whenusing externalcaptured CO2 asfeedstock)$
NG-based ethylene production (t-CO ₂ /t-Eth) (1:1 replacement of original H ₂ consumption)	1.42 (including feedstock emission)	0.58	-2.93 to -0.43

NE-H₂ provides opportunity to transform carbon emitters to carbon sinks across multiple sectors

*NE-H*₂ can be used synergistically with other renewables to dramatically drive down CI





Sugar-to-Jet CI: 2.03 kg CO₂e/kg (GREET)

5.0 MT/hr Jet * 2.03 kg CO₂/kg = 10.15 MT CO₂/hr 1.5 MT H₂/hr * -13 kg CO2/kg = -19.5 MT CO₂/hr

Sugar-to-Jet CI (NE-H₂) = -1.8 CO₂e/kg

Pathways for NE-H₂

Many feedstock / pathway combinations possible for NE-H₂ production



Immediate Options for NE-H₂

Immediate and longer-term decarbonization solutions needed to reach IPCC goals



Source: Columbia Center for Global Energy Policy: The Potential Role of Biohydrogen in Creating a Net-Zero World

fechnology readiness level

Challenges for Biomass-Derived NE-H₂ (Columbia)

- 1. Energy is required for harvesting, gathering, transporting, storage, and conversion of feedstocks. If biomass feedstocks are not carefully selected and the energy consumed during processing is not closely monitored and controlled, the carbon footprint can exceed that of fossil hydrogen
- Carbon-negativity of bio-H₂ is maximized through use of agricultural and municipal wastes, manures, sewages, etc. which are a constrained resource presenting obstacle to wide-scale deployment. Some feedstocks like RNG may benefit more from direct substitution of NG than H₂ formation
- 3. Cost and general feasibility tied to co-location of low-cost waste feedstocks and opportunities for geological storage of CO₂. Sites meeting both criteria not equally dispersed globally
- 4. Improving technical performance, including catalyst fouling, feedstock heterogeneity issues (inorganics), gas cleaning, low partial pressures, low volumetric yields in metabolic pathways

Will these challenges be showstoppers for widespread deployment? What would a long-term, de-risked, NE-H₂ portfolio look like?

Lower TRL Technologies for NE-H₂



Anode: $C_2H_4O_2 + 2H_2O \rightarrow 2CO_2 + 8e^- + 8H^+$

Cathode: $8H^+ + 8e^- \rightarrow 4H_2$

- Lower voltage req. than green H₂
- Pure CO₂ stream produced
- ~360 billion m³/y wastewater available globally

Non-Thermal Plasma Reforming

<u>Plasma Reforming of $CO_2 + H_2O$ Mixtures</u>

 $\begin{array}{l} H_2O \rightarrow H_2 + 0.5O_2\\ CO_2 \rightarrow CO + 0.5O_2\\ CO_2 + H_2 \rightarrow C + H_2O\\ 2CO \rightarrow CO_2 + C\\ CO + 3H_2 \rightarrow CH_4 + H_2O\\ CH_4 \rightarrow C + 2H_2 \end{array}$

Single-step process capable of sequestering carbon and producing H₂ simultaneously



Source: https://uwaterloo.ca/scholar/h95lee/research/mecs, https://www.frontiersin.org/research-topics/41852/non-thermal-atmospheric-pressure-plasma-and-its-biological-applications

Lower TRL Technologies for NE-H₂





> 7x higher CO₂ removal potential per unit energy generated for SWE than BECCS

Utilize renewable electricity to drive H_2O electrolysis and leverage OH- formation/alkalinity to spontaneously capture CO_2 in mineral bicarbonates / carbonates

Low TRL Challenges & R&D Needs

R&D Needs in MEC

- Low synthesis rates
- High internal resistances / ohmic losses
- Low durability; biofouling

- Enhancing microbe/anode electron transport
- Scalable designs

R&D Needs in NTP

- Poor selectivity
- Unoptimized reactor designs
- Degradation of heterogenous catalysts

Development of tandem catalysts

Commercially scalable designs

R&D Needs in SWE

- Land use/energy req. for minerals harvesting Minimize wastes (Cl₂)
- Optimization of electrolysis performance
- Mitigate environmental upsets

Conclusions and Way Forward

- A variety of feedstock pathway combinations exist to produce NE-H₂ spanning a wide TRL range, each with unique tradeoffs and opportunities. *There may not be a* one-size-fit-all solution
- 2. Carbon negative hydrogen can enable deep decarbonization of some of the worst polluting processes as well as synergize with other sustainable practices
- High TRL pathways relatively well characterized, need additional cross-cutting analysis to inform value proposition of emerging technologies; what are the optimal use cases? Are the near-term technologies also best suited for long-term use?
- How does the ever-evolving policy landscape impact the value proposition for NE-H₂ (e.g., 45Q vs. 45V)?

Kg of CO2 per kg of H2	Credit Value (\$)
4 - 2.5 kg CO2	\$0.60 / kg of H2
2.5 - 1.5 kg CO2	\$0.75 / kg of H2
1.5 - 0.45 kg CO2	\$1.00 / kg of H2
0.45 - 0 kg CO2	\$3.00 / kg of H2

Q&A

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