



“A” IS FOR AMMONIA: Current challenges, emerging research and NREL’s vision for a decarbonized future

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April 2, 2024 – *Guest Lecture: CU Boulder Advanced Energy Topics*

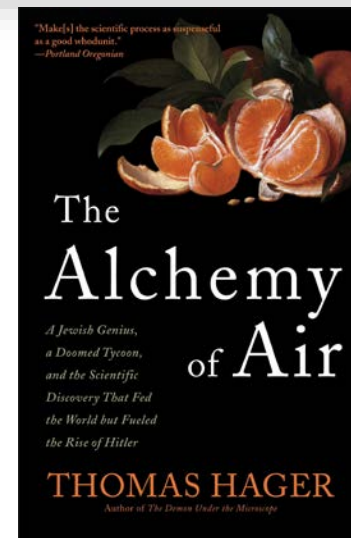
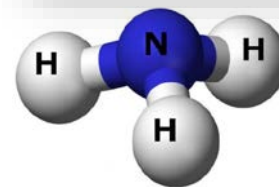
The (infamous) Haber-Bosch process

Global NH₃ production, by the numbers

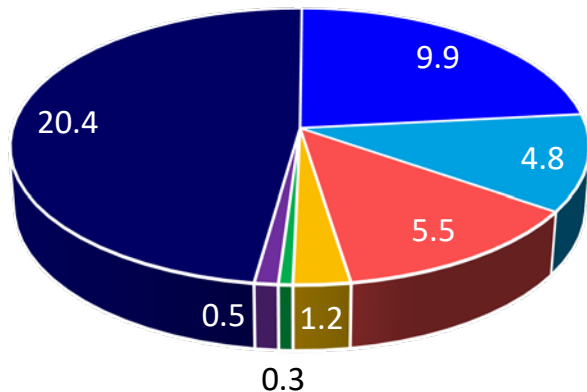
- Direct synthesis from **N₂** (from distillation of air) and **H₂** (from steam reforming of methane (SRM))
- ~180 million t produced annually, with >80% relegated solely to nitrogenous fertilizers
- **Consumes 1–2% of the global energy supply!** (and nearly ~5% of all natural gas produced)
- Accounts for ≤3% of the world's carbon emissions → ~340 million t CO₂-e yr⁻¹ (#1 of chemicals)

Haber-Bosch facts to impress (or scare) your friends at happy hour

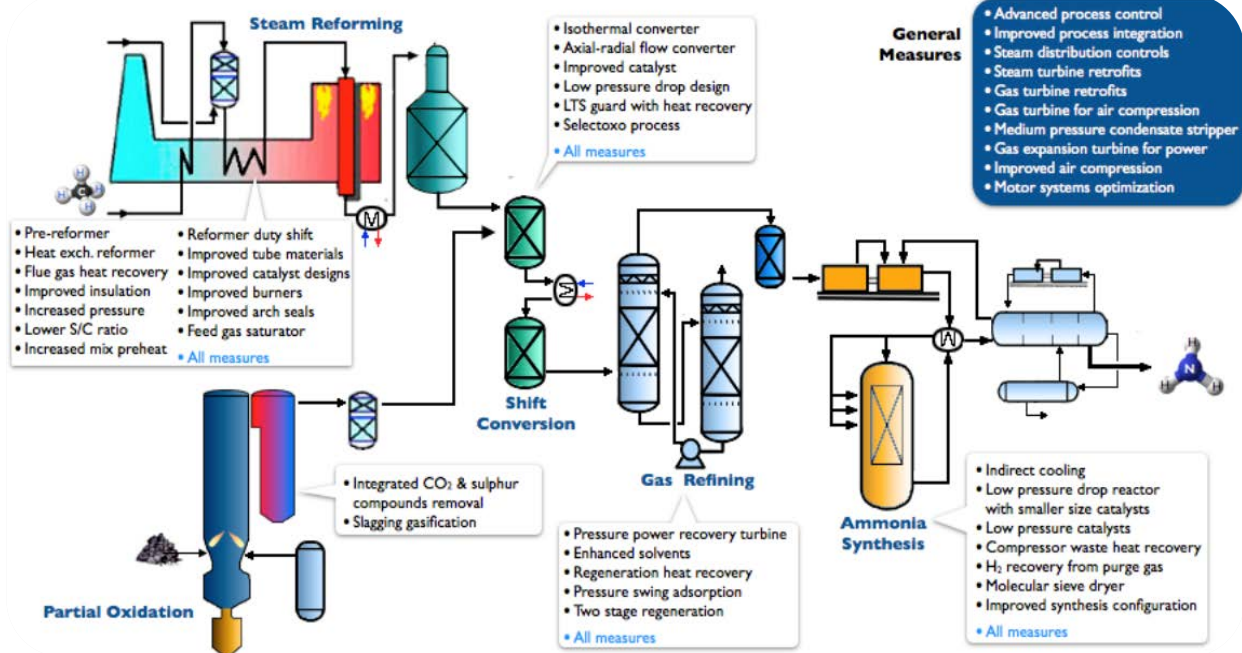
- Saved Germany's population in WWI (and, sadly, later enabled the evil of WWII and beyond...)
- Today... feeds >50% of the world's population
- Accounts for ~50% of N atoms in our bodies(!)



Estimated Energy Inputs in GJ (LHV) t⁻¹ NH₃ for U.S. Ammonia Plants (1996)



- Reformer Feed
- Reformer Fuel
- Primary Reformer
- Compressor
- Shift+CO₂ mgmt
- Flare
- Other Ops.



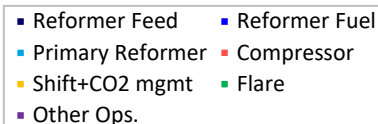
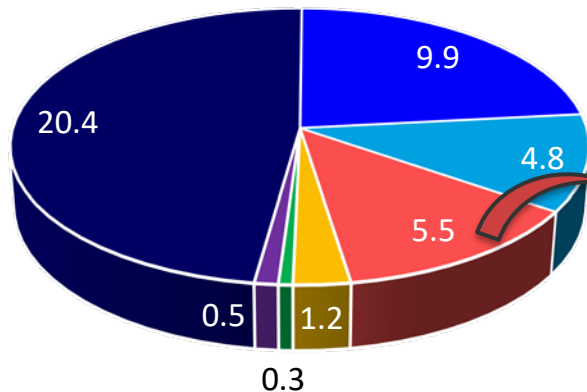
Opportunities to improve Haber-Bosch

- Numerous strategies proposed by academia and industry to lower the massive energy intensity of Haber-Bosch process
- Synthesis reactors operate at up to 300 atm and 550°C
- Catalysts are thermodynamically limited to low conversions (*and no, "new" thermochemical catalysts won't change that*)

Pie chart values rendered from: I. Rafiqul, A. Voss et al., *Energy* **30** 2487–2504 (2005). ([link](#))

Process schematic adapted from: Industrial Efficiency Technology Database, "Ammonia." Accessed 2020 September 8. ([link](#))

Estimated Energy Inputs in GJ (LHV) t⁻¹ NH₃ for U.S. Ammonia Plants (1996)



Renewable hydrogen could displace up to 70% of the pie—*but it's not the whole story.*



For example... an astonishing ~5.5–6.5 GJ t⁻¹ NH₃ energy input is required just to run compressors! (*H₂/N₂ recycle due to low per-pass conversion*)

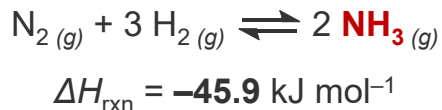
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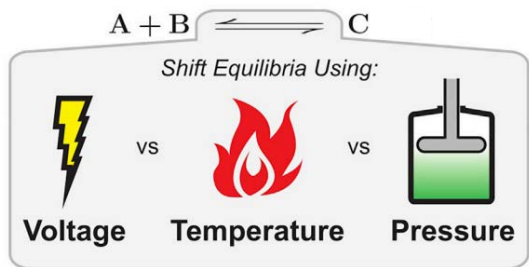
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Could renewable e⁻ be used to make NH₃?

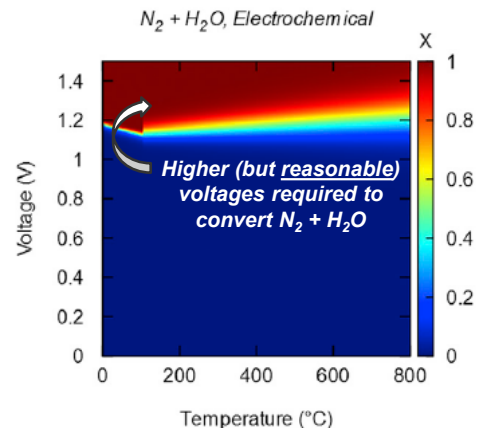
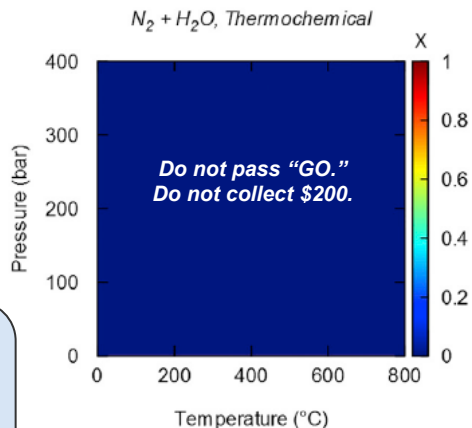
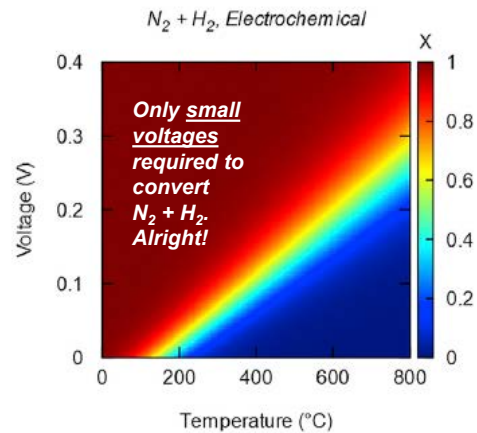
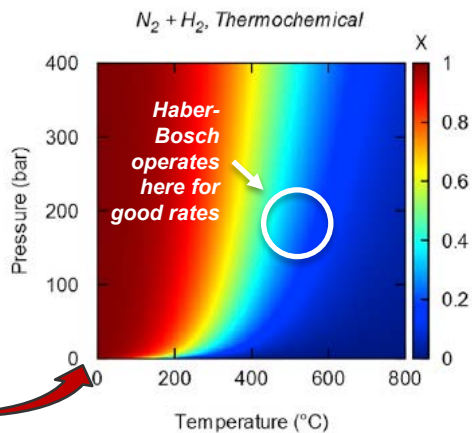


Red regions favor NH₃ formation



High T & P often mean giant capital investments!

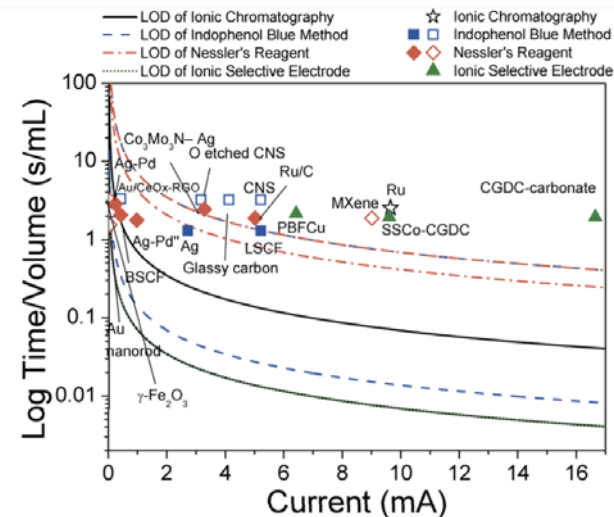
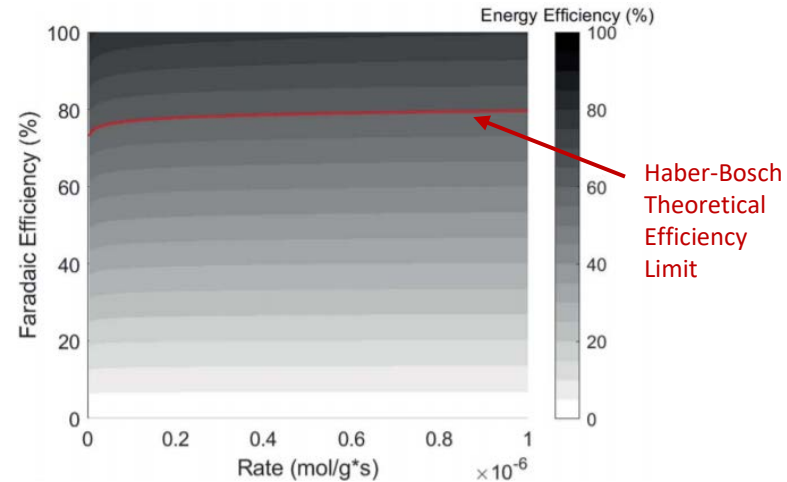
Beware of low Faradaic efficiencies. New electrocatalysts must achieve intrinsically higher selectivity to N₂ reduction. (Most would much rather make hydrogen!)



Figures adapted from: Z. J. Schiffer, K. Manthiram, *Joule* **1** 10–14 (2017). ([link](#))

Electrochemical routes and their challenges

- Electrochemical reactions scale areally, while thermochemical reactions scale volumetrically
- Energy efficiencies (~1%) are largely governed by the overall system faradaic efficiencies (FEs), which are a long way from Haber-Bosch (~60%)
 - However, optimizing solely FEs is often at the expense of the rate of ammonia production
- **Increasing FE requires design of electrocatalysts with high selectivity for nitrogen reduction**
 - Unfortunately, the redox potential for nitrogen reduction is very close to the hydrogen evolution reaction (HER), and most electrocatalyst surfaces will preferentially bind H* and not N*
- Further, critical NH₃ measurement issues from **false positives and/or easily reducible laboratory contaminants** (e.g., NO₃²⁻, NO_x), while isotopic ¹⁵N₂ gas is prohibitively expensive



Emergent R&D trends in ammonia synthesis

1. Less-energy-intensive feeds and unit operations for Haber-Bosch

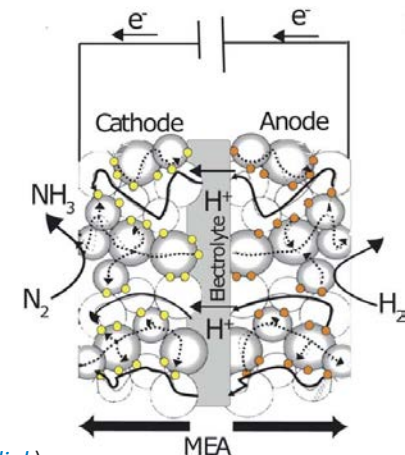
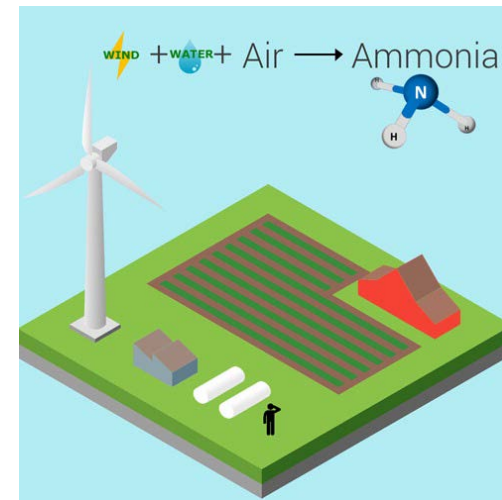
- Renewable hydrogen (e.g., from electrolysis)
- Renewable electrification in lieu of steam/nat. gas
- Energy efficient ammonia separations
 - Reactive swing sorption/desorption for ammonia recovery

2. Distributed NH₃ synthesis: miniaturized, electrified Haber-Bosch

- Modularized synthesis for synergistic agriculture, energy and chemicals sectors, with immediate opportunities for H₂@Scale integration
- Different design problem: *scaling down* & *numbering up* (vs. scaling up)
- Greatest challenges are in addressing operational intermittencies

3. Direct, low-temperature electrochemical NH₃ synthesis

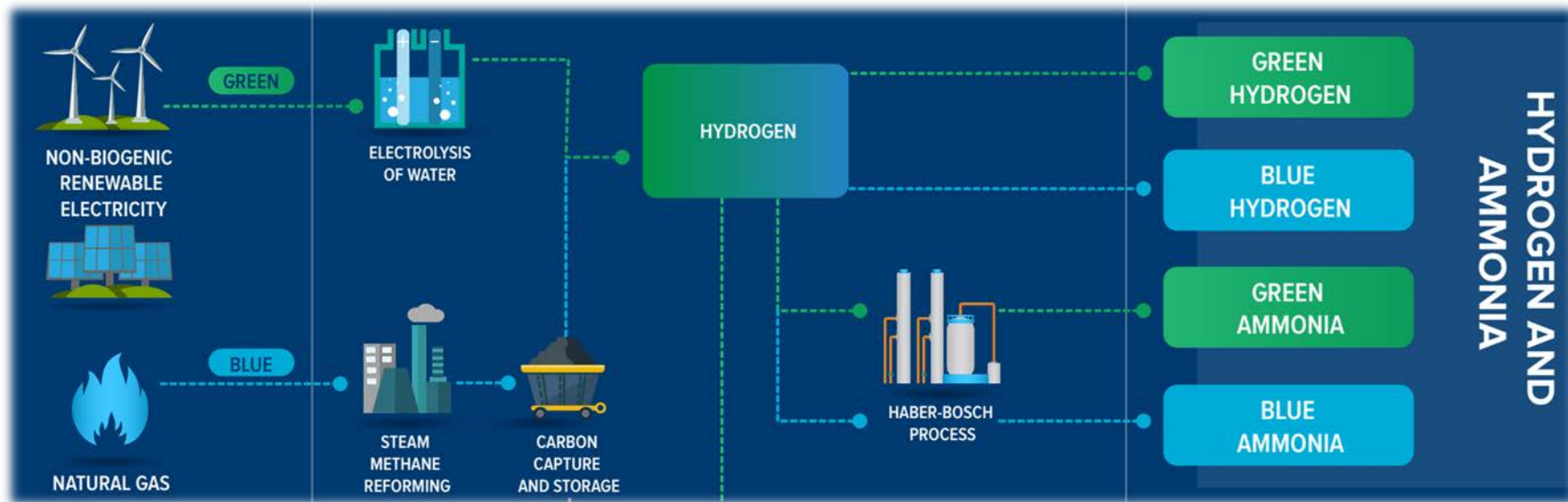
- Electrosynthesis via N₂ reduction and H₂O splitting
- Despite popularity, low rates/e⁻ efficiencies & major throughput hurdles to match Haber-Bosch scales (recall, electrochemistry scales areally)



Wind farm graphic adapted from: M. Reese, E. Cussler et al., *Ind. Eng. Chem. Res.* **55** 3742–3750 (2016). ([link](#))

Electrochemistry graphic adapted from: C. A. Fernandez, M. C. Hatzell et al., *J. Mater. Chem. A* **8** 15591–15606 (2020). ([link](#))

Taste the rainbow: colors of ammonia and hydrogen



Color matters! Carbon emissions must be negated across the broader supply chain to allow for truly renewable NH_3 .

Fly-by-nit(rogen): a case for NH_3 reforming, by the numbers

Global Ammonia Production

- 180,000,000 metric tons NH_3 yr^{-1}
- **10%:** 18,000,000 metric tons NH_3 yr^{-1}
 - Conversion potential of up to ~3,000,000 metric tons H_2 yr^{-1} , or
 - ~25,000,000 cars (**40–400×**)

vs. Global Hydrogen Fueling Infrastructure

- ~100–1000 kg H_2 day^{-1}
 - × 150 days yr^{-1}
 - × 500 stations
 - = 7,500–75,000 metric tons H_2 yr^{-1} , or
 - ~60,000–600,000 cars

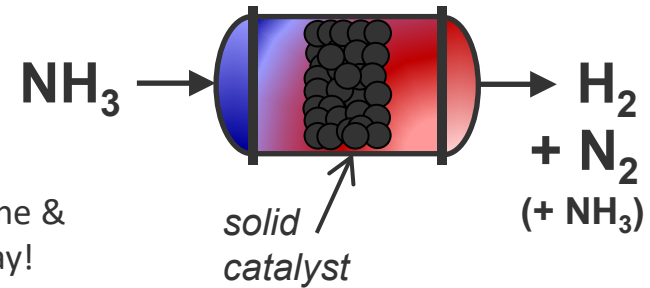
Today's global ammonia production capacity may readily meet (early) demands of the marine, aviation and off-road fuel markets, with lower technological barriers to establish liquid NH_3 fueling infrastructure. **NH_3 reforming remains one key barrier to adoption.**

Ammonia planes, trains & automobiles

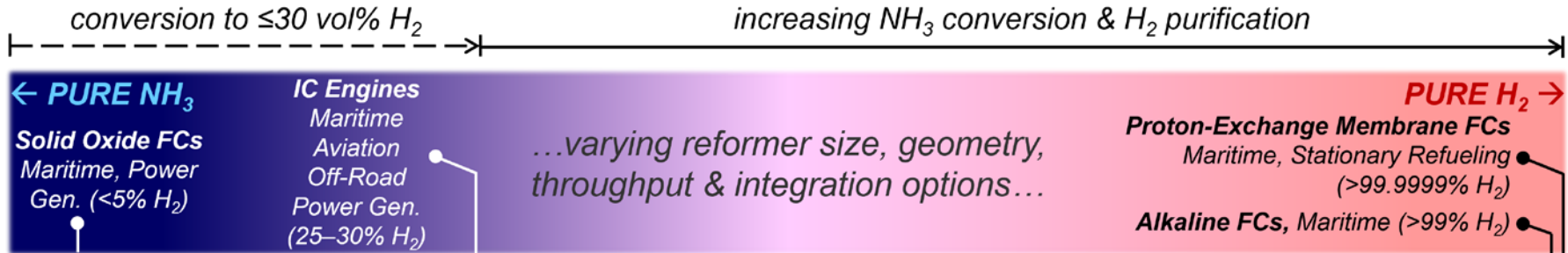
Ammonia is a promising net-zero-carbon liquid H₂ carrier and transportation fuel.



AMMONIA REFORMING (reverse of Haber-Bosch)



internal combustion (IC) engine & fuel cell (FC) designs underway!



NH₃ reforming requires (electro-/photo-)catalyst discovery, reactor design



$$\Delta H_{\text{rxn}} = +45.9 \text{ kJ mol}^{-1}$$



Shift Equilibria Using:



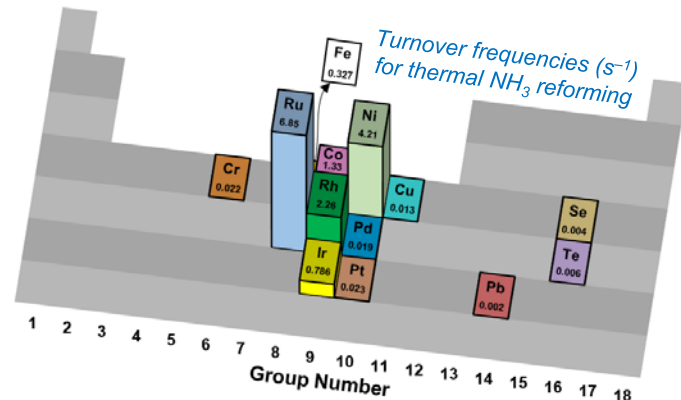
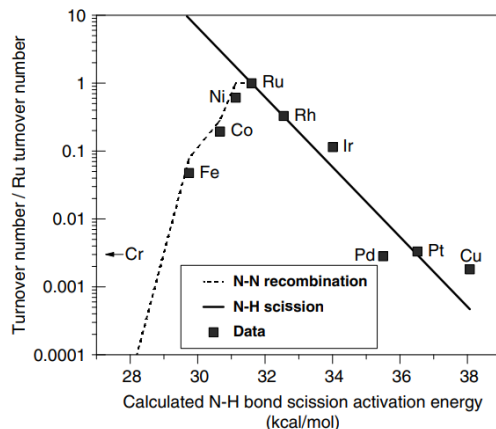
VS



VS



Voltage **Temperature** **Pressure**



CATALYST & REACTOR DESIGN CONSIDERATIONS

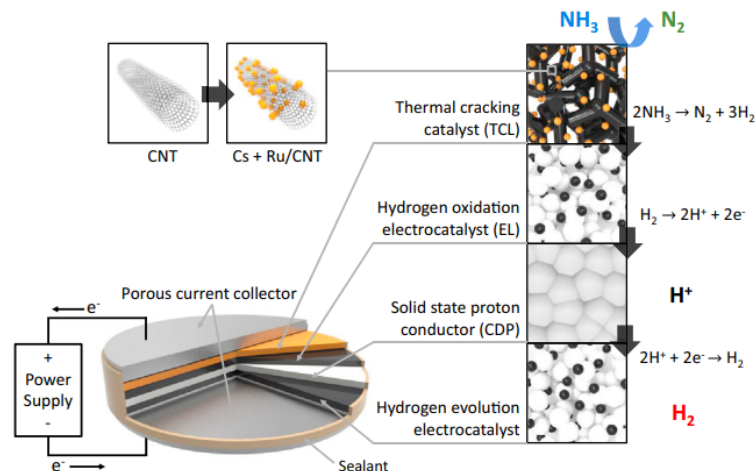
- Laboratory catalyst selectivity, activity and stability (“The Big Three”)
- Technical catalyst formulation, metal identity/loading (\$\$), tolerance to dynamic conditions*, mech. stability* (*esp. for mobile settings)
- Reforming reactor must feature light weight*, compact size*, high surface area, low pressure drop, low thermal latency, ability to reach required H₂ purity (perhaps requiring separation/purification steps)

Thermodynamic graphic adapted from: Z. J. Schiffer, A. M. Limaye, K. Manthiram, *Joule* **5** 135–148 (2021). ([link](#))

Volcano plot adapted from: J. C. Ganley, R. I. Masel et al., *Catal. Lett.* **96** 117–122 (2004). ([link](#))

Periodic TOF graphic adapted from: L. C. Caballero, N. E. Thornburg, M. M. Nigra, *Chem. Sci.* **13** 12945–12956 (2022). ([link](#))

Electrochemical cell graphic adapted from: D.-K. Lim, S. M. Haile et al., *Joule* **4** 2338–2347 (2020). ([link](#))



A ship-load of CO₂: decarbonizing maritime transportation



THINK FAST.

New maritime engines built in this decade will still be in use in 2050.

80% of global trade by volume
and over 70% by value

6.1% of world oil demand (2012)

2.5% of global GHG emissions (940 million tons CO₂/yr)

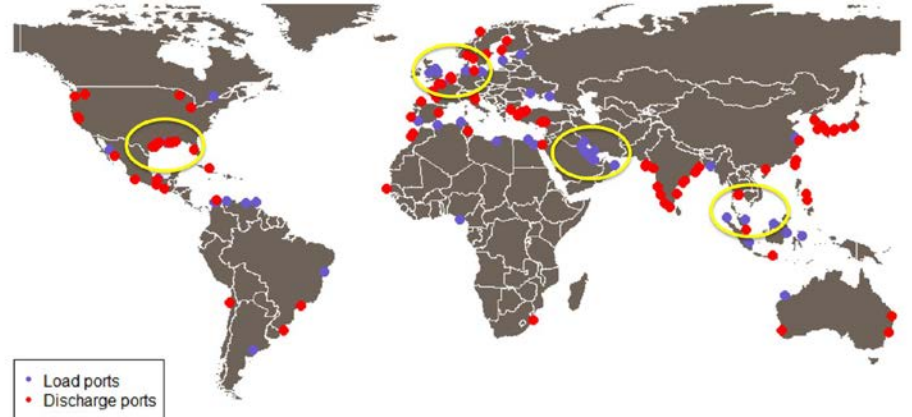
Unique opportunities in the maritime sector



Wärtsilä is already testing an ammonia engine. (Credit: Wärtsilä)

- However, combustion/ignition methods and fuel cells are not well understood for ammonia blended with other fuels

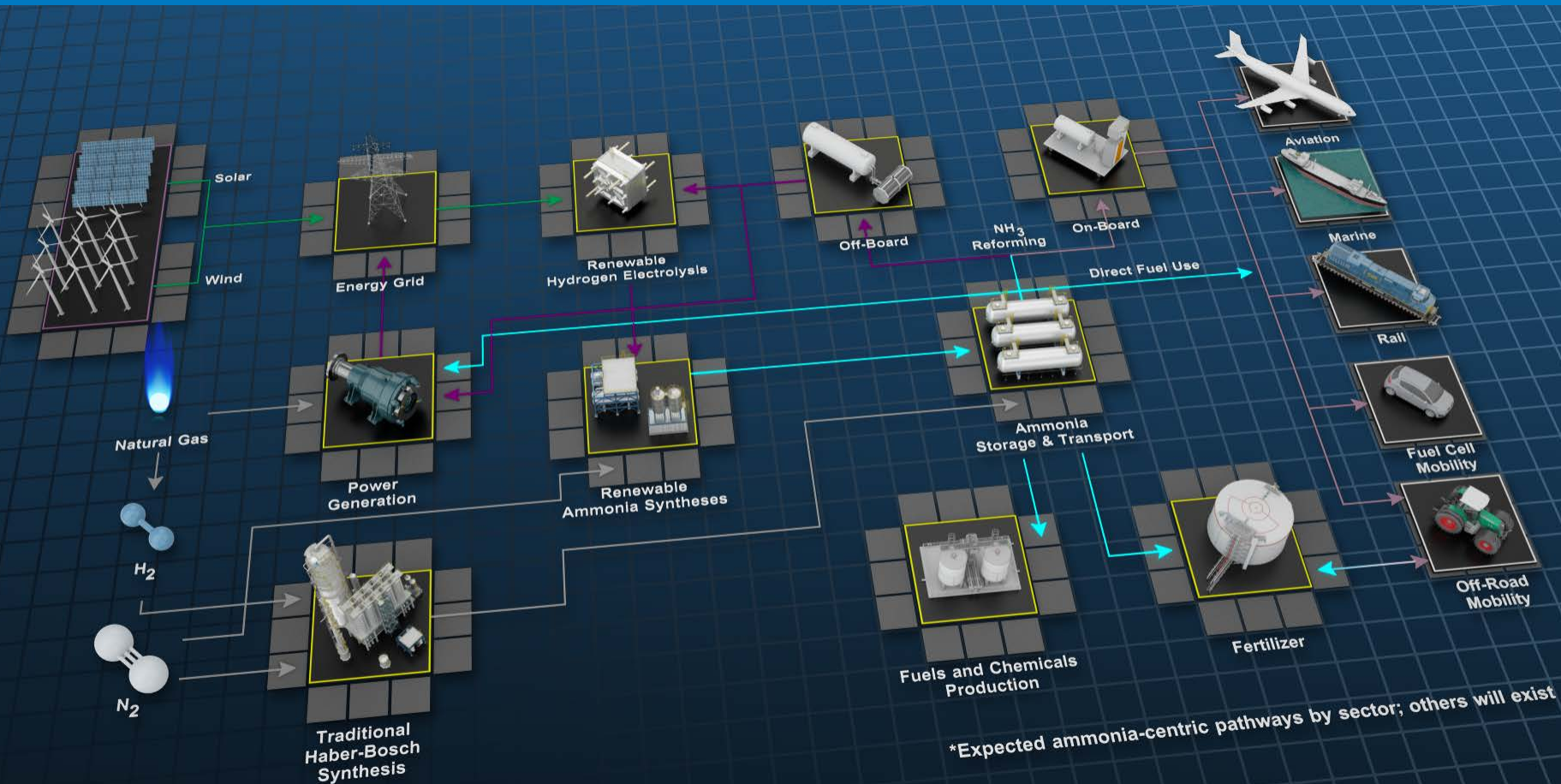
Worldwide ammonia ports



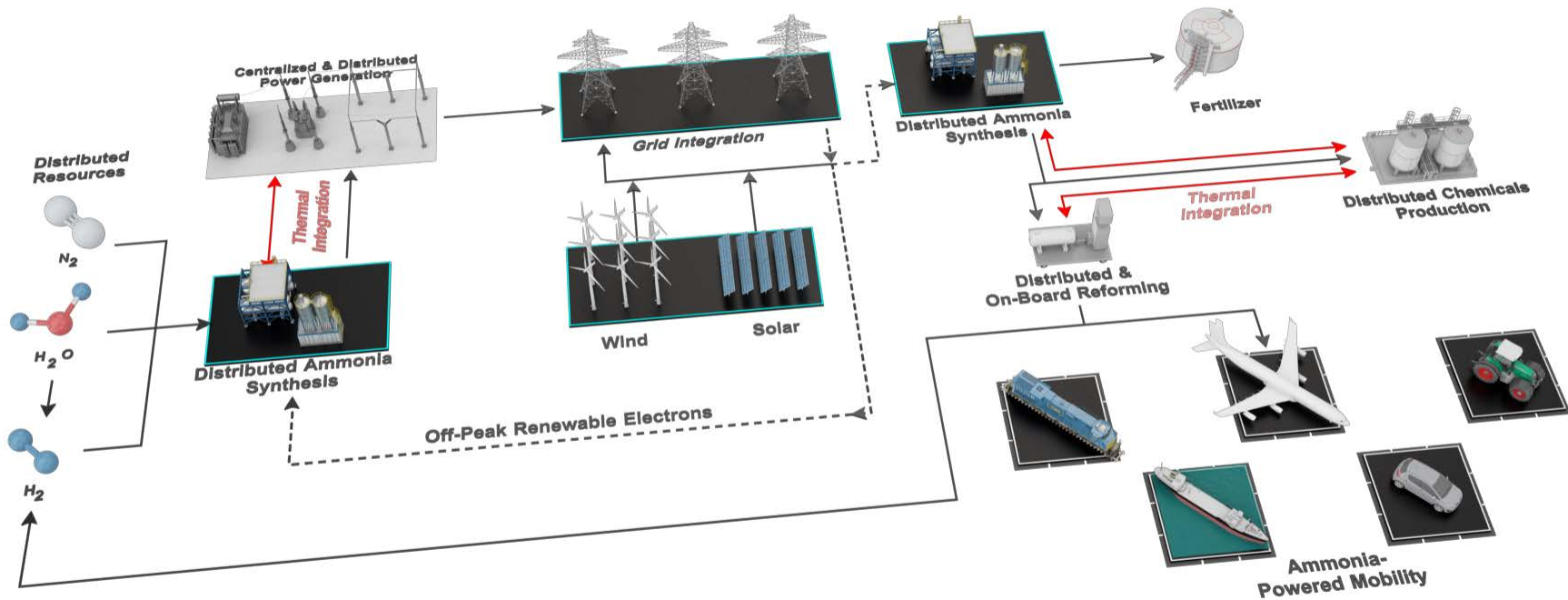
Visions for off-shore bunkering & synthesis



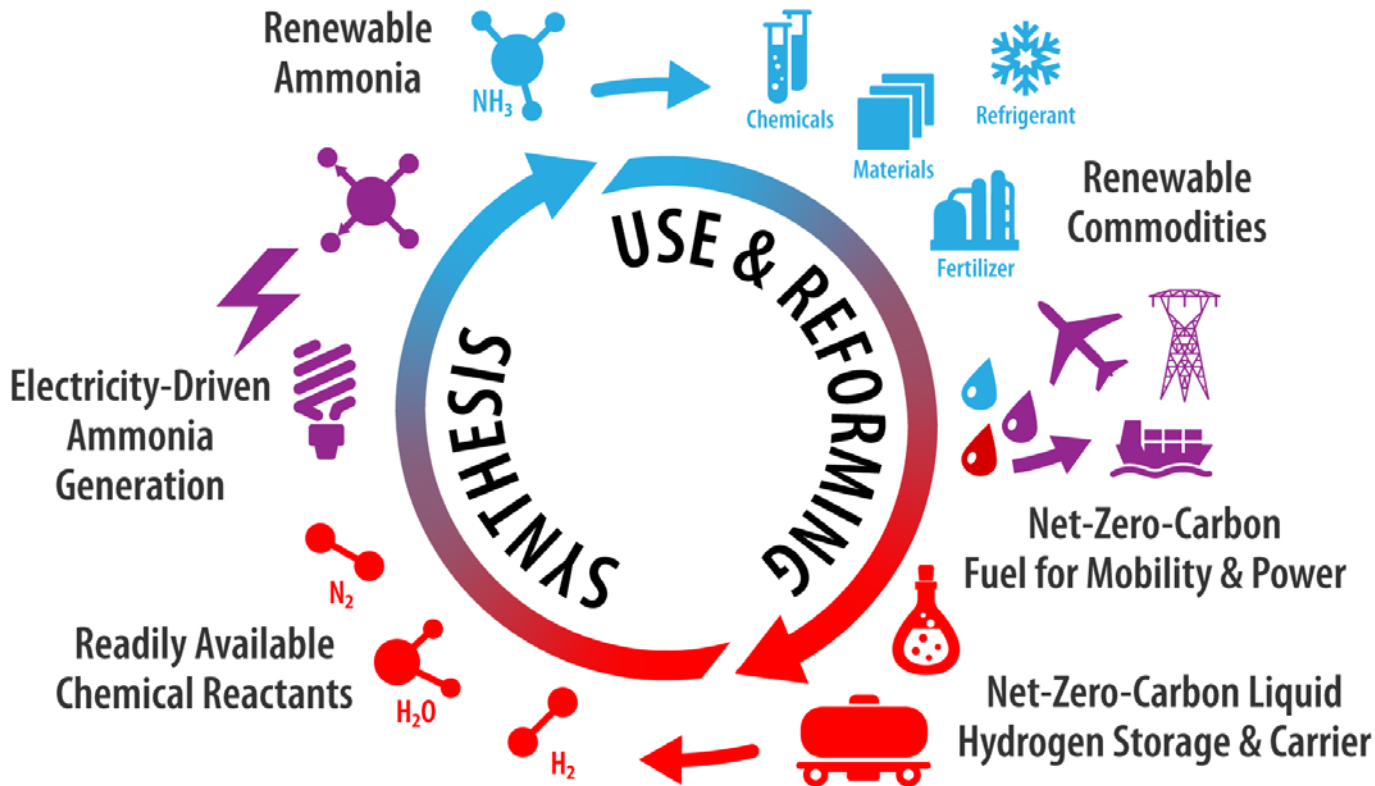
NREL's vision for the future ammonia energy economy



Accessing ammonia and its energy—before *and* after the meter



Circularity (and abundancy) of the nitrogen and hydrogen atoms



Thank You

www.nrel.gov

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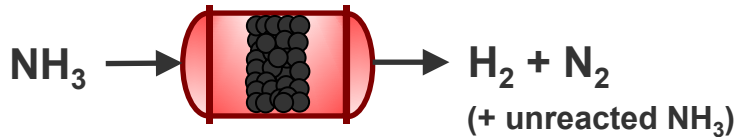
NREL is a national laboratory of the U.S. Department of Energy, Office of Energy Efficiency and Renewable Energy, operated by the Alliance for Sustainable Energy, LLC.



Ammonia vs. Hydrogen as Future Marine IC Fuels

| Physicochemical Property | CNG | Ammonia | Hydrogen |
|---|------|---------|----------|
| Density [kg/m ³ ; 1 atm, 25°C] | 0.65 | 0.70 | 0.090 |
| Boiling Point [°C; 1 atm] | -162 | -33 | -253 |
| Laminar Flame Speed [m/s; 1 atm, 25°C] | 0.40 | 0.12 | 2.0 |
| Lower Heating Value [MJ/kg] | 50 | 19 | 120 |
| Lower Heating Value [MJ/m ³ ; 1 atm, 25°C] | 33 | 13 | 9.8 |
| Stoichiometric AFR (mass basis) | 17 | 6.1 | 34 |
| Stoichiometric AFR (volume basis; 1 atm, 25°C) | 9.5 | 3.6 | 2.4 |

➤ Ammonia exhibits **several physical advantages** over hydrogen as a fuel for IC engines



- However, ammonia's poor combustion properties will require blending with a pilot fuel—or **on-board reforming** to generate partial hydrogen fuel mixtures *in situ*

Technical Barriers to Ammonia as a Marine “Fuel”

Ammonia fuel cells (SOFCs) and advanced IC engines are already under development in Europe, Japan, and Australia, but several **key technical challenges** remain:

- Current ammonia SOFCs lack sufficient power density and load response capability for ocean-going trips
- Combustion and ignition methods are not yet well understood for ammonia blends
 - Pre-mixed vs. direct injection?
 - Spark (vs. pilot) vs. compression ignition?
 - Combustion promoter(s) required?
- Turbocharger, lubricating oil, and materials of construction requirements?
- Reformers and emissions management systems are also ill-defined and could be prohibitive (e.g., high NO_x emissions)



Wärtsilä is already testing an ammonia engine. (Credit: Wärtsilä)

*...and it **aaall** hinges on NH₃ being made renewably...*

Hydrogen and Ammonia May Play Different Roles

Due to engine output size, ammonia combustion is expected to be suitable for ocean going vessels

We predict that liquefied hydrogen fuel combustion engines (H₂ gas turbines, etc.) will be put into practical use and spread after 2040.

