Application of Cd₂SnO₄ Transparent Conducting Oxides in CdS/CdTe Thin-Film Devices

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ABSTRACT

An unusual processing scheme has been developed for preparing single-phase cadmium stannate ($\mathrm{Cd_2SnO_4}$ or CTO) films. $\mathrm{Cd_2SnO_4}$ transparent conducting oxide (TCO) films have several significant advantages over conventional TCOs when applied to CdS/CdTe thin-film devices. They are more conductive, more transparent, have lower surface roughness, are patternable, and are exceptionally stable. $\mathrm{Cd_2SnO_4}$ -based CdS/CdTe polycrystalline thin-film solar cells with efficiencies of 14% have been fabricated. Preliminary cell results have demonstrated that device performance can be enhanced by replacing the $\mathrm{SnO_2}$ layer with a $\mathrm{Cd_2SnO_4}$ -TCO film.

INTRODUCTION

Cadmium telluride has long been recognized as a promising photovoltaic material for thin-film solar cells because of its near optimum bandgap of 1.5 eV and its high direct absorption coefficient. Small-area CdS/CdTe heterojunction solar cells with efficiencies of more than 15% and commercial-scale modules with efficiencies of 9% have been demonstrated [1,2]. In the past, little attention has been paid to the transparent conducting oxide (TCO) layer of this device structure. Conventional TCOs, primarily SnO₂ films, typically have been used as the front collector in CdS/CdTe cells and modules. However, SnO₂ films, deposited using the commercially viable SnCl₄ chemistry, have a resistivity of $\sim 5-8x10^{-4} \Omega$ cm. This yields films with an average transmission of 80% and a sheet resistivity of about 10 Ω /square. Although this may be satisfactory for small-area devices and firstgeneration modules, it does not provide adequate design latitude when trying to optimize either device performance or manufacturing costs. For example, reducing the TCO resistivity by a factor of two could allow for a TCO film that retains the same sheet resistance as the inferior TCO, yet is half as thick. This would improve the transmission and ultimately improve device performance by increasing the short-circuit current density (J_{sc}). Alternatively, a film with transmission properties similar to those of the inferior TCO, but half the sheet resistance, could be used. This would have the advantage of reducing the number of interconnects required, thereby improving throughput, reducing interconnect losses, and reducing manufacturing costs. For these reasons, it is desirable to improve the performance of the TCO films for CdS/CdTe solar cell applications.

Nozik and later Haacke et al. were the first to report $\mathrm{Cd_2SnO_4}$ TCO films deposited by r.f. sputtering [3,4]. Recent improvements in r.f. sputter-deposited $\mathrm{Cd_2SnO_4}$ and post-deposition processing have yielded films with superior properties [5,6]. In this work, we describe cadmium stannate films that we have optimized for CdS/CdTe devices. Cadmium stannate films have several significant advantages over conventional TCOs. They are more conductive, more transparent, have lower surface roughness, are patternable, and are very stable. $\mathrm{Cd_2SnO_4}$ -based CdS/CdTe polycrystalline thin-film solar cells with efficiencies of 14.0% have been fabricated. Preliminary cell results have demonstrated that device performance can be enhanced by replacing the $\mathrm{SnO_2}$ layer with a $\mathrm{Cd_2SnO_4}$ -TCO film.

EXPERIMENTAL

Single-phase Cd₂SnO₄ films have been prepared using an unusual processing scheme developed in our laboratory [5,6]. CTO films were prepared by r.f. magnetron sputtering from a commercial hot-pressedoxide target with a composition of 33 mol% SnO2 and 67 mol% CdO. Corning 7059 glass or soda-lime glass substrates were placed on a water-cooled sample holder parallel to the target surface. Deposition was performed at an oxygen partial pressure of 10-20x10⁻³ torr at nominally room temperature, and a deposition rate of ~100 Å/min. After deposition, all samples were annealed in a tube furnace in an Ar or Ar/CdS atmosphere at elevated temperatures of 580°-680°C for 10-30 minutes. electrical, optical, and compositional properties of CTO films were characterized using Hall effect, optical and infrared spectroscopy, X-ray diffraction (XRD), scanning electron microscopy (SEM), and atomic force microscopy (AFM) measurements. We have also processed a limited number of CTO-based CdS/CdTe devices. devices, the CTO film thickness varied from 2500Å to 6000Å, with resistivities of 7 to 3 Ω /sq., respectively. The CdS thickness was 800-1000Å, and the CdTe film thickness was around 10 μm . The details of the CdTe cell fabrication procedure is described elsewhere [7].

RESULTS AND DISCUSSION

Cadmium stannate material properties

The conductivity of TCOs can be improved by either increasing the carrier concentration or the electron mobility. By optimizing sputtering and post-annealing conditions of CTO films, electron mobilities as high as 65 cm²/Vs have been achieved for a carrier concentration of 2x10²⁰ cm⁻³. Even at a carrier concentration of ~9x10²⁰ cm⁻³, mobilities as high as 55 cm²/Vs have been observed. These mobilities are 2-3 times higher than those of commercial SnO₂ films doped to similar levels. carriers are thought to result from oxygen deficiencies in the films, accommodated either as oxygen vacancies or cadmium interstitials, or a combination of both. results in resistivities as low as $1.3x10^{-4} \Omega$ -cm in the best CTO samples [7]. This is almost seven times lower than conventional SnCl₄-based SnO₂ films and 2.5 times lower than the best SnO₂ films deposited using tetramethyltinbased precursors ($\rho_{TMT} = 3.3x10^{-4} \Omega$ -cm). Table 1 shows a comparison of electrical properties of more typical Cd₂SnO₄ films prepared at NREL and commercially available SnCl₄-based SnO₂ films. The reduced resistivity of CTO films provides a clear advantage. For example, a sheet resistance of $\sim 9 \Omega/\text{sq}$. can be achieved with a CTO film of less than 2000Å, as compared to nearly 10,000Å for the SnO₂ film. Alternatively, thicker films can be used to reduce the device series resistance, thus improving its fillfactor and efficiency. Reducing the series resistance in modules is of particular interest, because manufacturers can optimize cell width between laser scribe lines, thereby reducing I2R and optical losses.

Table 1. Electrical Properties of CTO and SnO₂ Films.

	•			
Sample	t (Å)	ρ (Ω-cm)	Rs $(\Omega/sq.)$	
SnO ₂ (SnCl ₄)	10000	8.6x 10 ⁻⁴	8.6	
CTO245	1900	1.8x10 ⁻⁴	9.5	
CTO229	2550	1.8x10 ⁻⁴	6.9	
CTO250	3950	2.0x10 ⁻⁴	5.0	
CTO252	6500	2.2x10 ⁻⁴	3.1	

 ${\rm Cd_2SnO_4}$ films have significantly better optical properties than conventional ${\rm SnO_2}$ films. This is, in part, due to the lower resistivities, which allow thinner films to be used. For example, Fig. 1 shows the transmittance and absorbance of ${\rm Cd_2SnO_4}$ and ${\rm SnO_2}$ films with similar sheet resistivities (~10 Ω /sq.). The CTO film has superior transmission when compared to the ${\rm SnO_2}$ film, even though the CTO film has a slightly lower sheet resistivity. The absorbance of the CTO film, in the visible range, is much smaller than that of the ${\rm SnO_2}$ film. For example, at 600 nm, the absorbance of the CTO film is only 0.8%, compared to

12.2% for the $\rm SnO_2$ film. This appears to be due to the unusually high electron mobility of CTO films. The improved transmission and absorbance of CTO films is ideal for the superstrate structure of CdS/CdTe devices and will ultimately yield improved short-circuit currents.

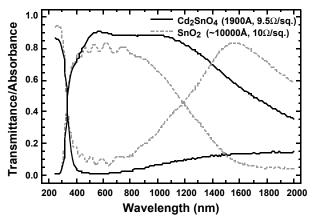


Fig. 1. Transmission and absorbance of a CTO and SnO_2 film with a sheet resistivity of ~10 Ω /square.

Fig. 2 shows the absorbance data of the four CTO films listed in Table 1. As expected, the absorbance increases with an increase in CTO film thickness. However, more interesting is the shift in optical bandgap to longer wavelengths with increasing CTO film thickness. The thinnest CTO film has a bandgap that is near that of a SnO₂ film (see Fig. 1). We do not believe that the bandgap shift is a result of the Moss-Burstein shift, because the thinnest CTO film has the lowest carrier concentration (the plasma edge is also located at longer wavelengths). Although we do not completely understand this effect, we believe that it may be a result of changes in film composition throughout its thickness. X-ray photoelectron spectroscopy (XPS) and XRD analysis is under way to test this hypothesis.

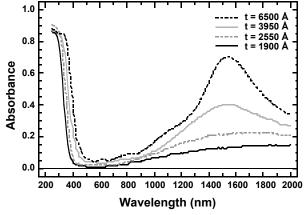


Fig. 2. The absorbance of four CTO samples with different thickness.

TCO surface roughness is also an important parameter when considering device processing issues. It is well known that in heterojunction solar cells, higher short-circuit currents can be obtained by reducing the window layer absorption. In CdS/CdTe cells, this is achieved by reducing the CdS thickness to yield an enhanced blue spectral reponse. Unfortunately, as the CdS is thinned, pinholes form and cause localized CdTe/SnO₂ junctions with inferior open-circuit voltages and fill factors. The probability of forming pinholes likely increases as the TCO surface roughness increases. AFM measurements indicate an average surface roughness of ~210Å for SnCl₄-based SnO₂ films. In contrast, the Cd₂SnO₄ films have a very smooth surface with an average roughness of ~20Å, an order of magnitude lower than that of the SnO₂ film [7]. This reduction in surface roughness could lead to improved device performance.

Cadmium stannate films also have excellent adhesion and thermal stability [7]. Our experimental results show that the resistivity of a $SnCl_4$ -based SnO_2 film degrades when annealed in an Ar ambient for 20 minutes at a temperature of $500^{\circ}C$. In contrast, CTO films are stable up to temperatures as high as $650^{\circ}C$ [7]. This is particularly important when considering that close-spaced sublimation (CSS) CdTe deposition temperatures can exceed $600^{\circ}C$.

The properties of CTO films described above are well correlated to the microstructure of the film. For example, Fig. 3 shows typical XRD data for an as-deposited and an Ar/CdS-annealed CTO film. The as-deposited films are amorphous, exhibit extremely high resistivities (~5000 Ω -cm), have a lower optical bandgap, and have poor transmission characteristics. After the Ar/CdS anneal, the film crystallizes into a single-phase spinel structure with optical and electrical properties similar to those described in Table 1. After crystallization, there is no evidence that secondary phases, such as CdO, SnO $_2$,

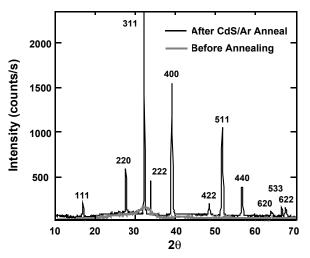


Fig. 3. Typical XRD pattern of an as-deposited and an annealed Cd_2SnO_4 film.

or CdSnO $_3$, form in our CTO films. We find that Ar/CdS anneals drastically reduce the temperature required for film recrystallization. We have demonstrated the same improvement in CTO film properties at 580° C anneals on low-cost, soda-lime glass, achieving a resistivities as low as $1.84 \times 10^{-4}~\Omega$ -cm. This value is similar to values obtained for the higher-temperature process.

Cadmium-stannate-based CdS/CdTe devices

The improved material properties of CTO films provide a strong impetus to incorporate them in a superstrate device structure. We find that processing parameters that were optimum for SnO2-coated glass are not suitable for CTO-coated glass. This is particularly evident when comparing the post-CdTe-deposition CdCl₂ anneal. CdTe/CdS/SnO₂ devices that have not been CdCl₂ treated, exhibit open-circuit voltages (Voc) of about 550 mV. To obtain open-circuit voltages of >800 mV, it is necessary to soak the sample in a saturated CdCl₂:MeOH solution with a concentration of greater than 50% (typically between 50%-100%) and subsequently anneal the film at 400°C. At higher concentrations (near 100%), the films exhibit severe adhesion problems, thereby reducing yields. In contrast, CdTe/CdS/Cd2SnO4 devices require a lower CdCl₂ conentration. CdTe/CdS/Cd₂SnO₄ devices, which have not been CdCl2-treated, exhibit open-circuit voltages as high as 650 mV (nearly 100 mV higher than the SnO₂based devices). Furthermore, CdCl₂ concentrations as low as 25% are adequate to provide a V_{∞} of >800 mV.

To compare the differences between Cd_2SnO_4 - and SnO_2 -based CdTe cells, a set of CdTe cells was prepared using identical cell fabrication procedures. The cells were soaked in a saturated 25% CdCl₂:MeOH solution for zero to 20 minutes and then annealed at 400°C for 30 minutes. As shown in Fig. 4, the efficiency of Cd_2SnO_4 -based devices is always higher than that of SnO_2 -based devices

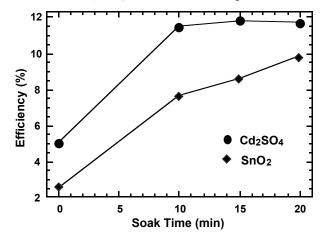


Fig. 4. CdTe cell efficiency as a function of soak time for a 25% CdCl₂:MeOH solution (note: under different processing conditions, higher efficiency devices have been achieved both with and without a CdCl₂ treatment).

for a given soak time. Earlier work has shown that this is also true when varying the $\mathrm{CdCl_2}$ concentration and keeping the soak time constant [7]. This result could, in part, be caused by the reduced surface roughness of the CTO film. However, it is likely that the chemistry of the CTO/CdS interface is also an important factor. Fig. 4 also indicates that the efficiency of CTO-based CdS/CdTe devices is less dependent on the $\mathrm{CdCl_2}$ heat-treatment. This result suggests the potential for improved process reproducibility and yield.

In addition to improved device performance at lower $CdCl_2$ concentration treatments, CTO-based devices have better adhesion than their SnO_2 counterparts. Although the SnO_2 -based devices improve with increased $CdCl_2$ concentrations, the adhesion degrades. At 100% $CdCl_2$ concentrations, most CdTe films will blister and eventually separate from the SnO_2 superstrate. Although CTO-based devices require lower $CdCl_2$ concentrations for optimum performance, most devices remain intact after a 100% $CdCl_2$ treatment. For example, in a set of 16 CTO-based devices treated at 100% $CdCl_2$, only 2 out of 16 devices resulted in adhesion failure. This could have significant process yield ramifications.

As described earlier, the improved transmission and absorbance of CTO films is ideal for superstrate CdS/CdTe devices and should yield improved device $J_{\rm sc}$. Fig. 5 shows the relationship between the TCO film transmission and the resulting $J_{\rm sc}$, for devices deposited on both CTO and SnO_2 superstrates. It is clear that by replacing the SnO_2 film with a CTO superstrate, the reduced optical losses yield an increase in $J_{\rm sc}$. For example, replacing the $10,000\text{\normalfont}$ SnO $_2$ film with a 2500\text{\normalfont} CTO film yielded an increase in $J_{\rm sc}$ of more than 1.5 mA/cm². It is important to note that devices fabricated on the thicker CTO films always gave the highest fill factors. Fill factors of 75% have been achieved using 6000\text{\normalfont}-thick CTO films in ~1 cm² devices. These films have resulted in devices with

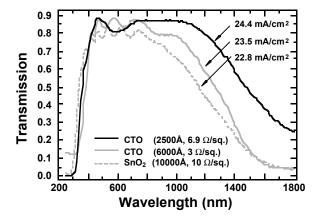


Fig. 5. Relationship between $\rm J_{\rm sc}$ and transmission of $\rm SnO_2$ and CTO films.

efficiencies of up to 14% (V_{oc} =0.805 V, J_{sc} =23.5 mA/cm², FF=74%), as measured under standard conditions. Therefore, thicker CTO films, with lower sheet resistivities, may be better suited for module applications.

CONCLUSIONS AND FUTURE WORK

Cd₂SnO₄ transparent conducting oxides have several significant advantages over conventional tin oxides. They are more conductive, more transparent, have a lower surface roughness, are patternable, and are exceptionally stable. Cd₂SnO₄-based CdS/CdTe polycrystalline thin-film solar cells with efficiencies of 14.0% have been prepared for the first time. The preliminary cell results have demonstrated that, by replacing the conventional SnO2 with a Cd₂SnO₄ film, CdS/CdTe device performance can be enhanced. Cd₂SnO₄-based CdTe cells yield higher shortcircuit currents and efficiencies, and are less dependent on the CdCl₂ treatment than their SnO₂ counterparts. We believe that the performance of Cd₂SnO₄-based CdS/CdTe cells is far from optimized. Future work will concentrate on improving the efficiency of Cd₂SnO₄-based CdS/CdTe cells through the optimization of Cd₂SnO₄ film preparation and cell fabrication and on developing more manufacturable processes.

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REFERENCES

- [1] J. Britt and C. Ferekides, "Thin-Film CdS/CdTe Solar Cell With a 15.8% Efficiency" *Applied Physics Letters*, **62**, pp. 2851-2852 (1993).
- [2] K. Zweibel, H. Ullal, and B. von Roedern, "Progress and Issues in Polycrystalline Thin-Film PV," *Proc. 25th IEEE Photovoltaic Spec. Conf.*, (1996) pp. 745-750.
- [3] A.J. Nozik, "Optical and Electrical Properties of Cd₂SnO₄: A Defect Semiconductor," *Phys. Rev. B.*, **6**, No. 2, pp. 453-459 (1972).
- [4] G. Haacke, W.E. Mealmaker, and L.A. Siegel, "Sputter Deposition and Characterization of Cd₂SnO₄ Films," *Thin Solid Films*, **55**, pp. 67-81 (1978).
- [5] T.J. Coutts, X. Wu, W.P. Mulligan, and J.M. Webb, "High-Performance, Transparent Conducting Oxides Based on Cadmium Stannate," *J of Electronic Materials*, **25**, No. 6, pp. 935-943 (1996).
- [6] X. Wu, W.P. Mulligan, and T.J. Coutts, "Recent Developments in RF-sputtered Cadmium Stannate Films," *Thin Solid Films*, **286**, pp.274-276 (1996).
- [7] X. Wu, P. Sheldon, T.J. Coutts, D.H. Rose, W.P. Mulligan, and H.R. Moutinho, "CdS/CdTe Thin-Film Devices Using a Cd₂SnO₄ Transparent Conducting Oxide," *Proc.* 14th NREL/SNL PV Program Review Meeting, pp.693-702 (1996).